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Liquid Metals Enable the Next Generation of Ultrathin Transparent Conductive Materials

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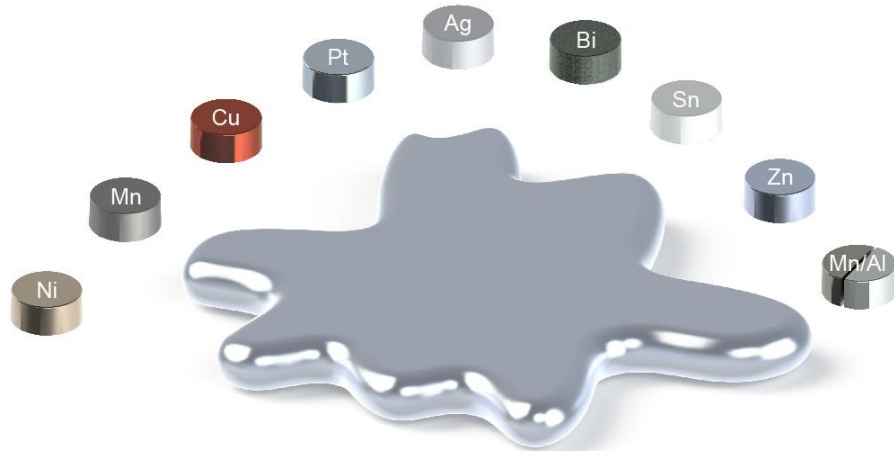


What are Liquid Metals (LM) ?

Liquid Metals (LM):

- Post transition elements
Ga, In, Tl, Sn, Pb, Al, Bi
- Zinc group (group 12) elements
Zn, Cd and Hg

	5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen
	13 Al Aluminum	14 Si Silicon	15 P Phosphorus	16 S Sulfur
29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic
47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony
79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth
				84 Po Polonium



Metallic core

- Liquid electrodes
- Synthesis of nanoparticles
- Energy storage
- Microfluidics
- Electronic and telecommunication elements
- Plasmonics
- Stress and motion sensing

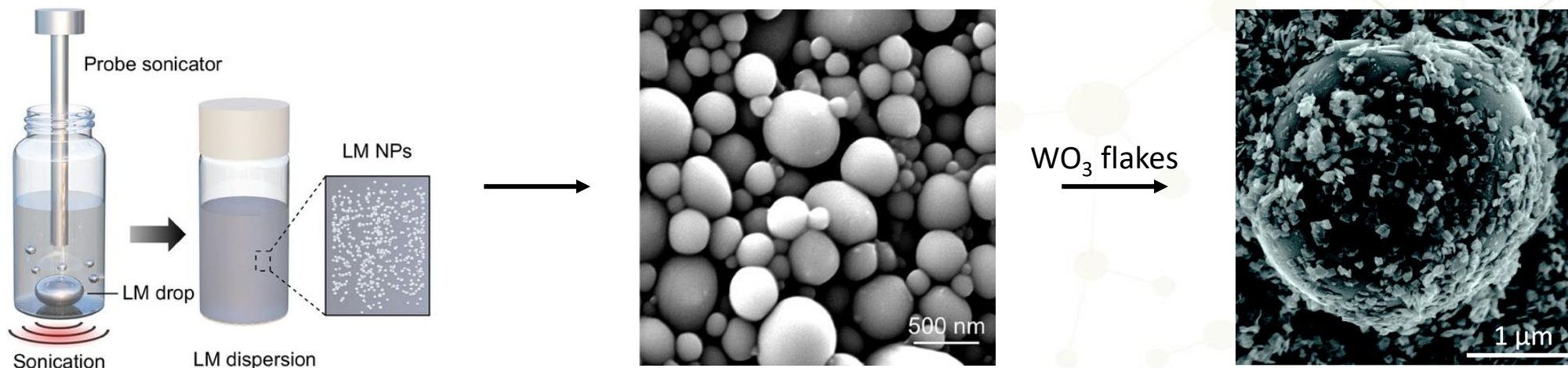


Interface

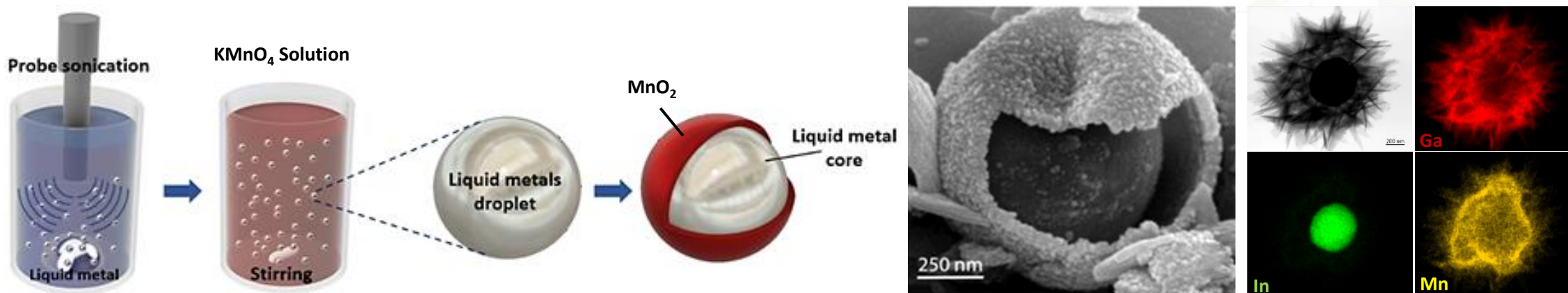
- Catalysis
- Synthesis of 2D materials
- Chemical sensing
- Electrocatalysis
- Tension modulation
- Surface wetting
- Mechanical tunability

Synthesis and functionalisation of LM nanoparticles

Physical functionalisation



Chemical functionalisation



Han J. *et al.*, Liquid metals for tuning gas sensitive layers, *Journal of Materials Chemistry C* 2019,7, 6375-6382.

Ghasemian M.B. *et al.*, Coating of gallium-based liquid metal particles with molybdenum oxide and oxysulfide for electronic band structure modulation, *Nanoscale*, 2023, 15, 5891-5898.

Ghasemian M.B. *et al.*, Self-Limiting Galvanic Growth of MnO₂ Monolayers on a Liquid Metal—Applied to Photocatalysis, *Advanced Functional Materials* 2019,29, 1901649.

Two-dimensional (2D) materials

- Two-dimensional (2D) materials are planar structures with nanometer thicknesses
- Electronic band structure of 2D materials usually changes with composition and thickness variation

Preparation methods of 2D materials:

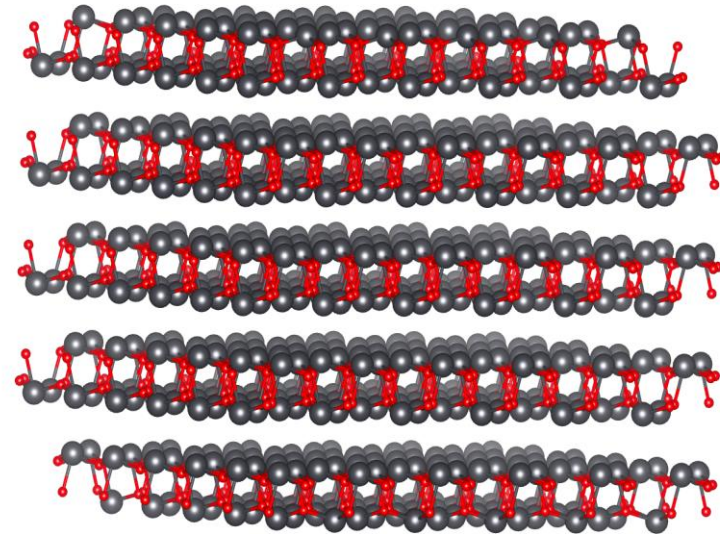
1. Exfoliation:

- Mechanical cleavage
- Chemical exfoliation
- Ion intercalation
- Surfactant assisted ultrasonication

2. Deposition:

- Chemical vapour deposition (CVD)
- Molecular beam epitaxy (MBE)
- Pulsed laser deposition (PLD)

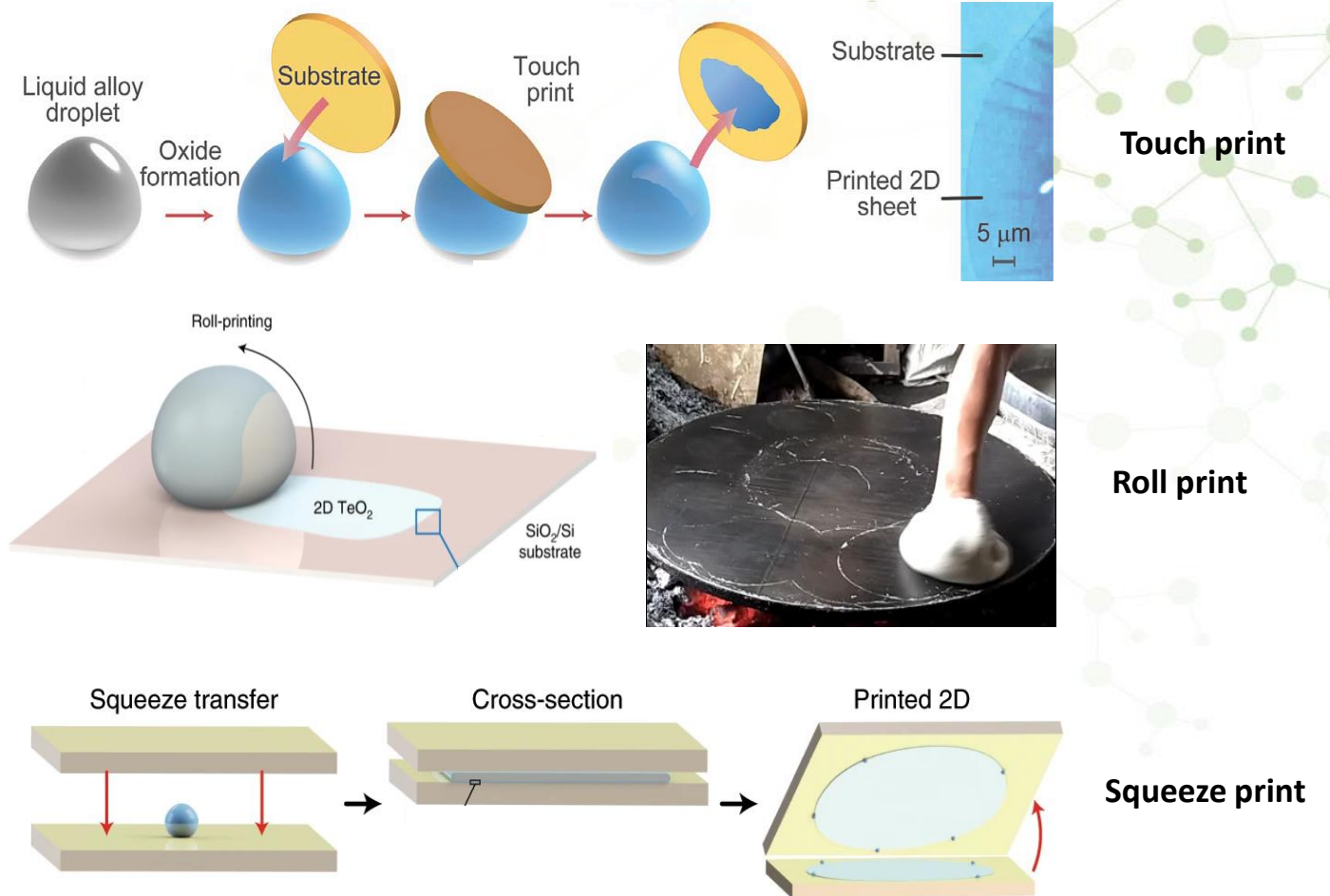
3. Liquid metal-based printing



Synthesis of 2D and ultrathin materials by LM printing

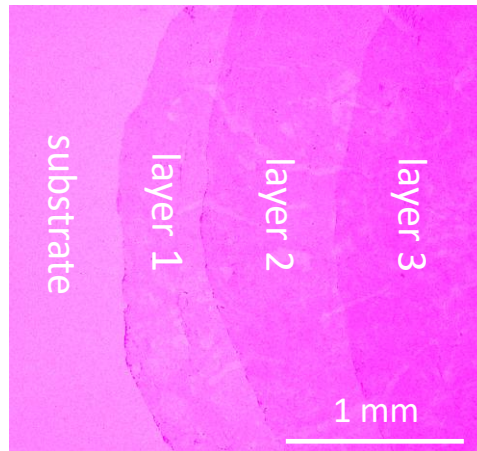
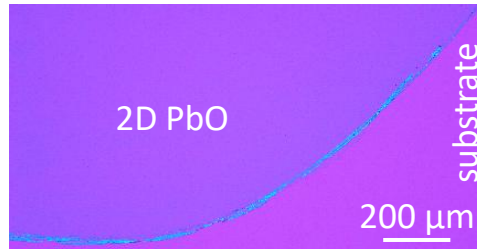
Liquid metal printing principal:

- An atomically thin layer of oxide grows on the surface of liquid metals
- There is no covalent bond between liquid metal and its surface oxide
- There are strong van der Waals interactions between surface oxide and the transfer substrate

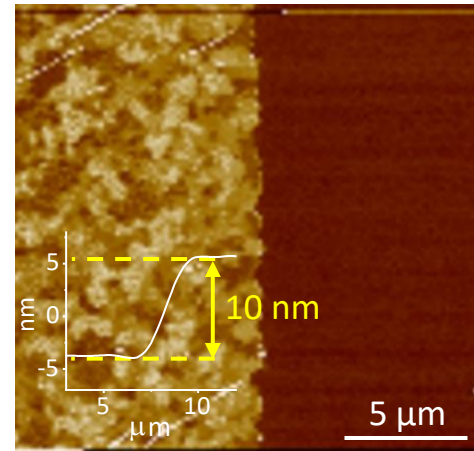
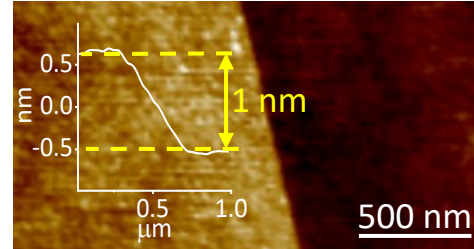


Environmentally friendly PbO monolayers by LM

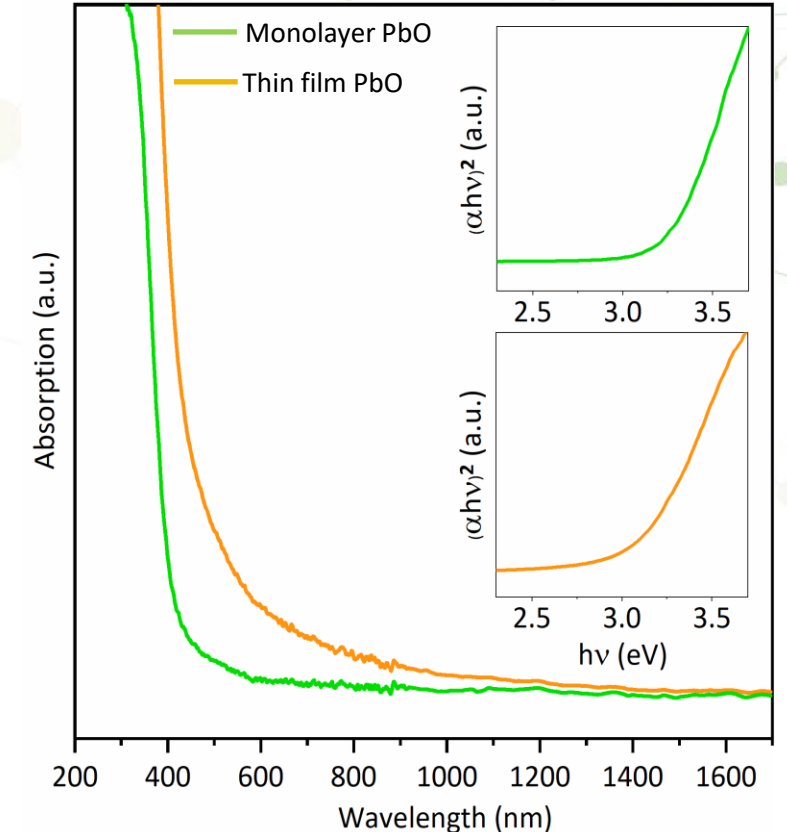
Optical images



AFM images



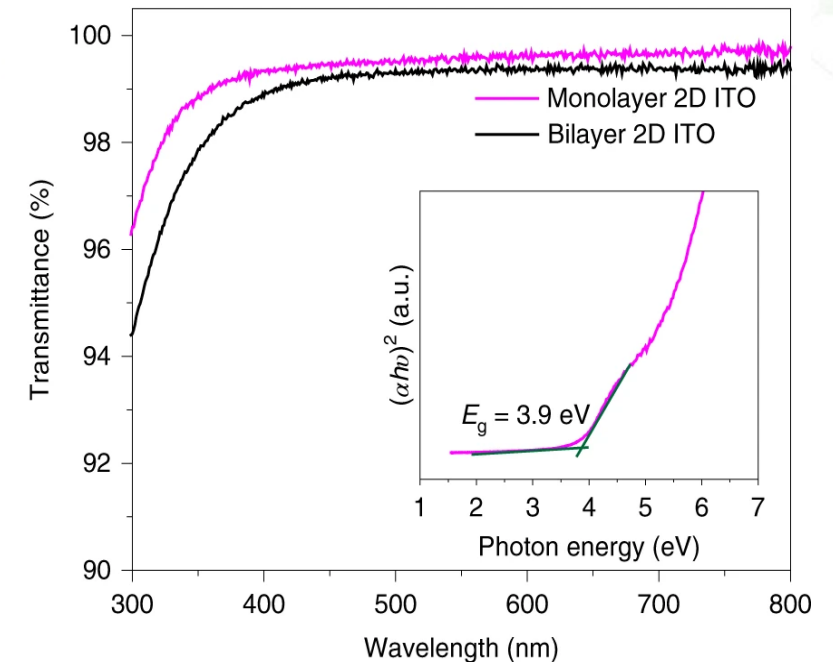
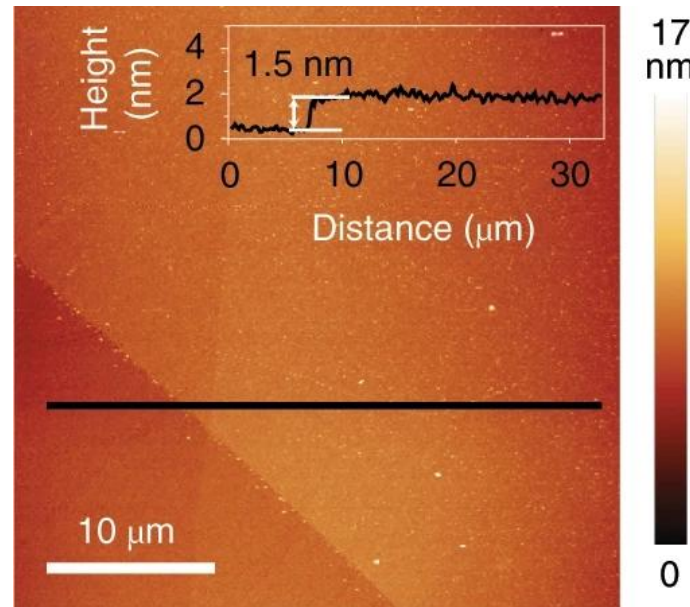
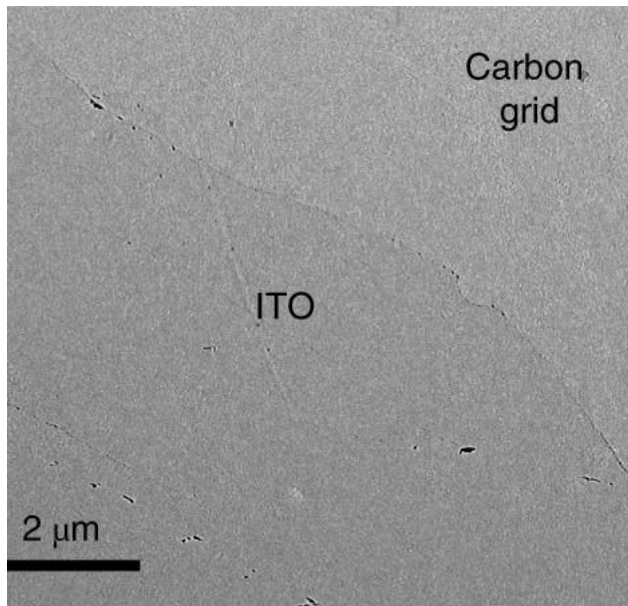
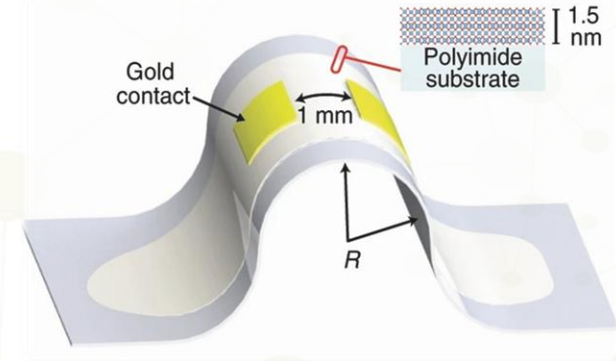
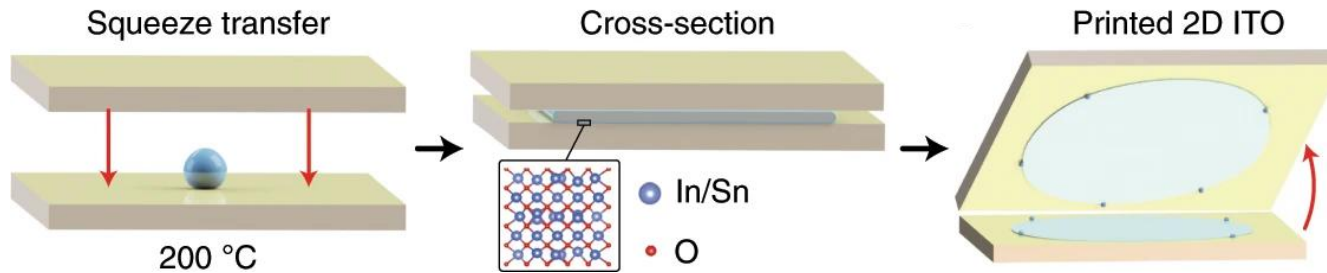
UV- Vis and Tauc plots for band gap measurements



Harvesting PbO sheets with different thicknesses from the surface of liquid Pb through the touch-printing process

- PbO monolayers showed a piezoelectric coefficient of ~ 30 pm/V
- 10^4 times less Pb than commercial PZT thin films

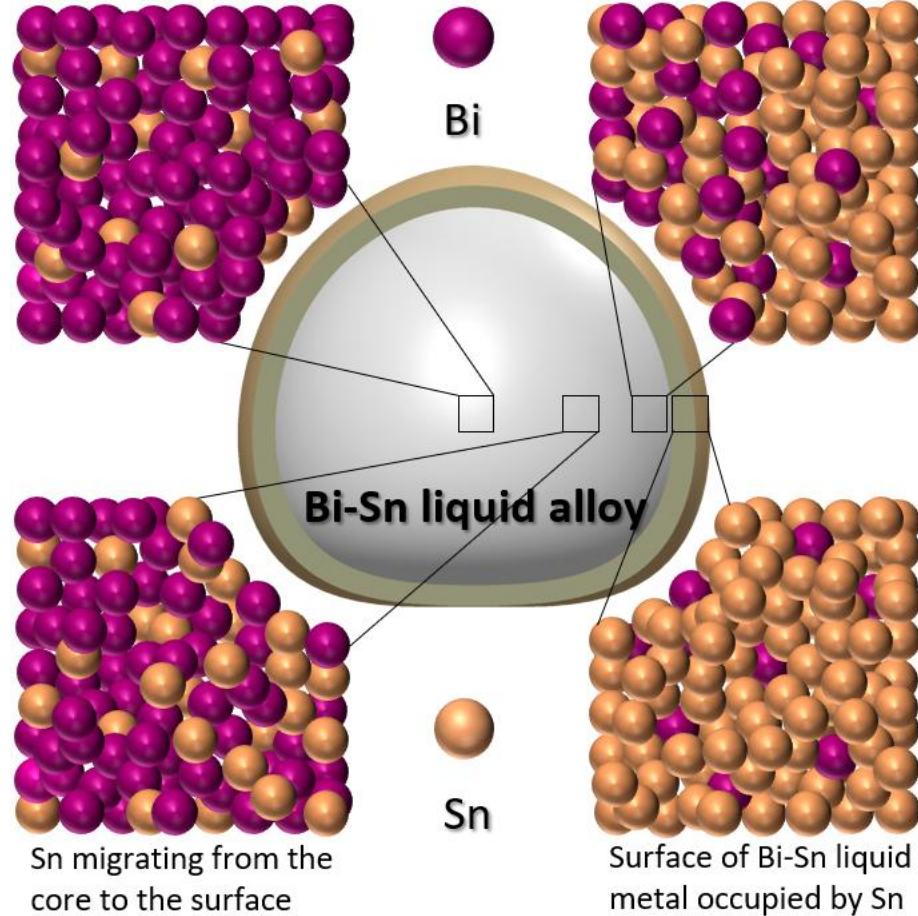
Liquid metal-based ultrathin transparent ITO



Liquid metals as a doping strategy

The core of Bi-Sn liquid metal with both Bi and Sn

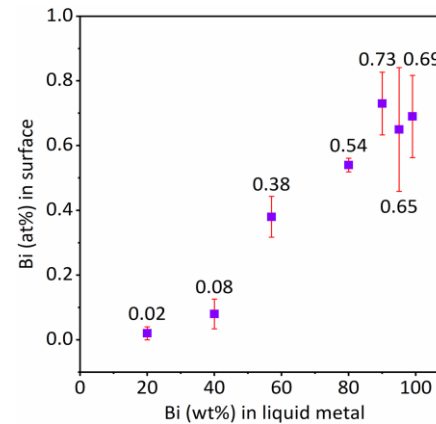
Sn migrating from the core to the surface



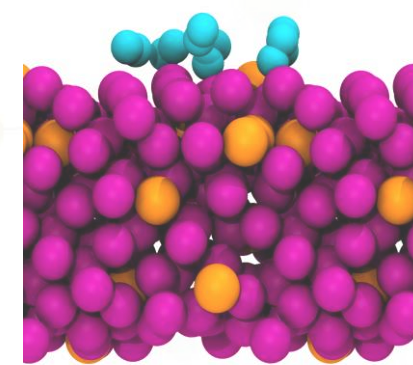
$$\text{Bi}_2\text{O}_3 \Delta G_f = -574 \text{ kJ/mol}$$

$$\text{SnO}_2 \Delta G_f = -578 \text{ kJ/mol}$$

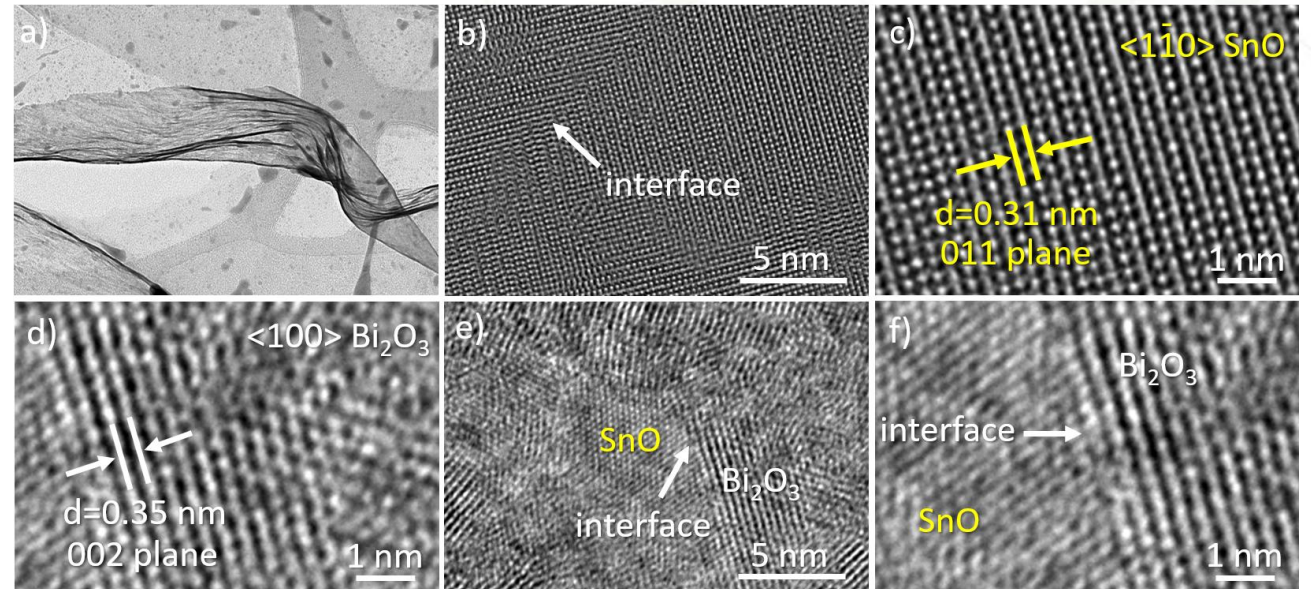
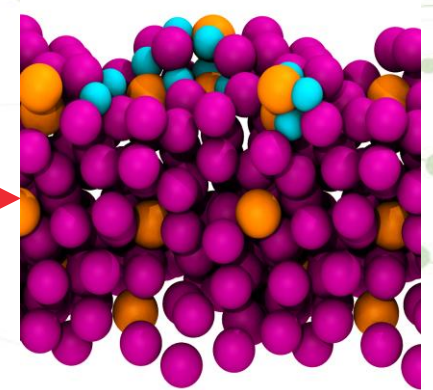
Quantitative XPS



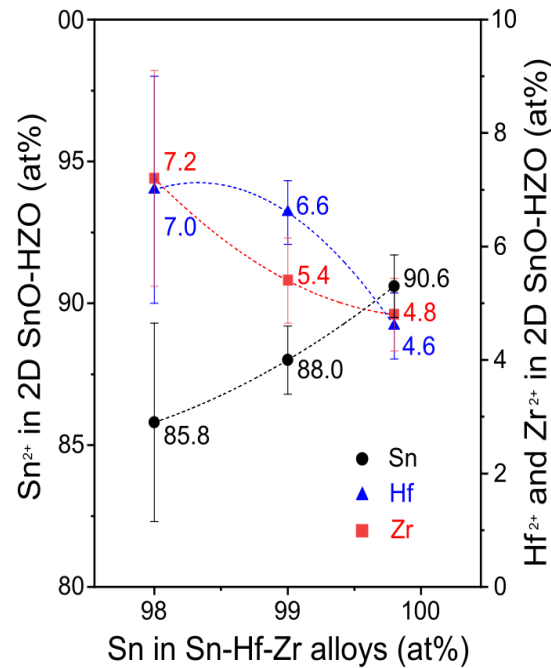
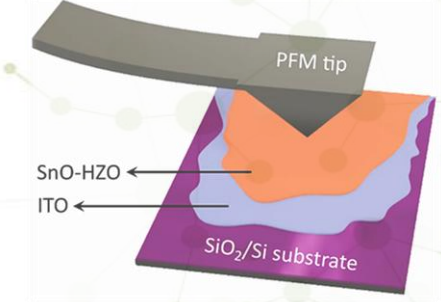
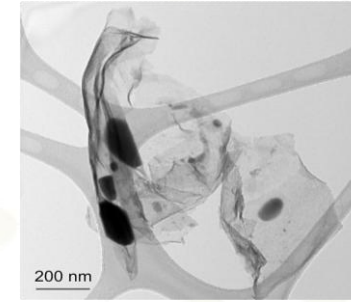
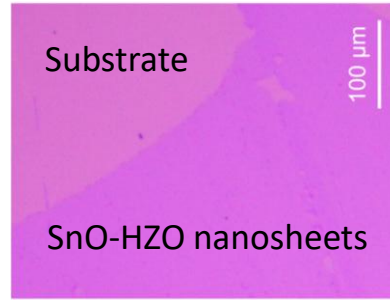
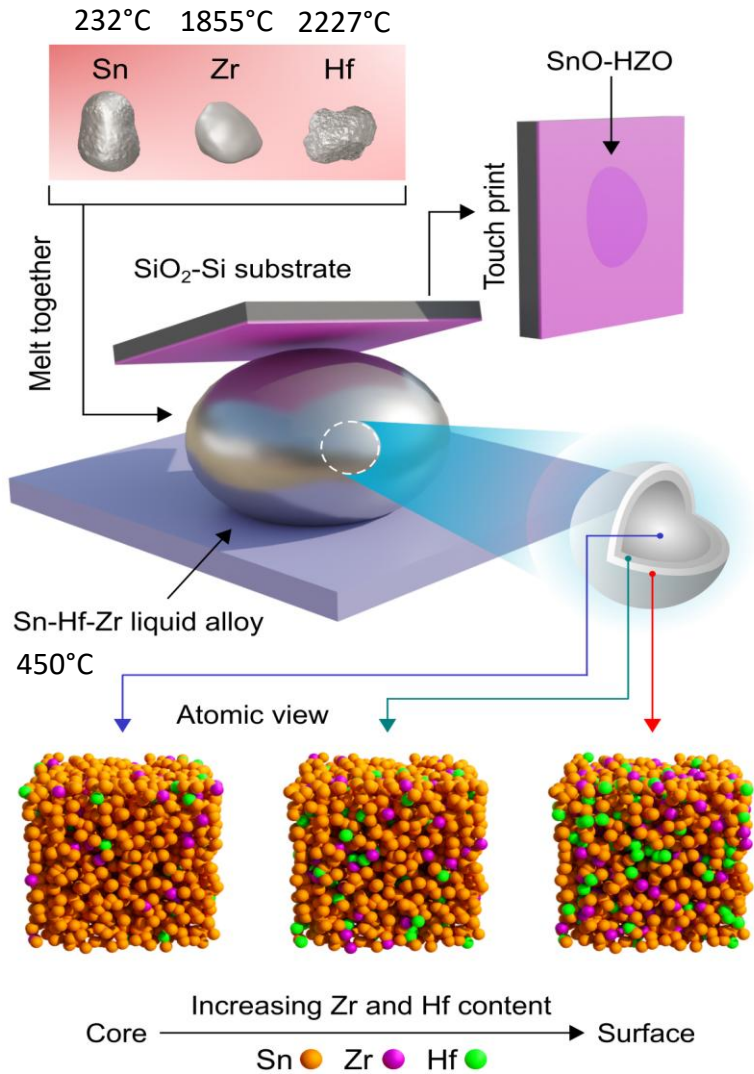
Initial configuration



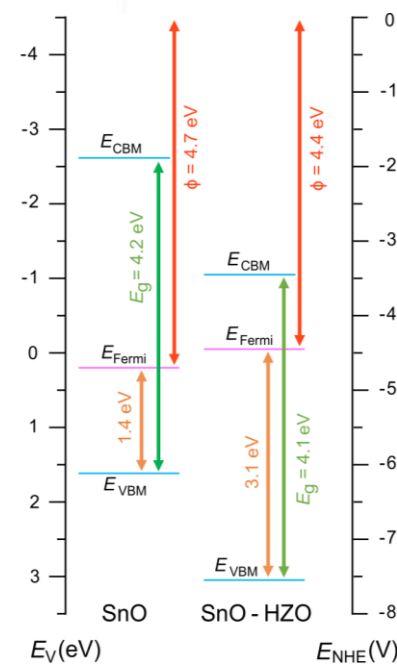
Final configuration



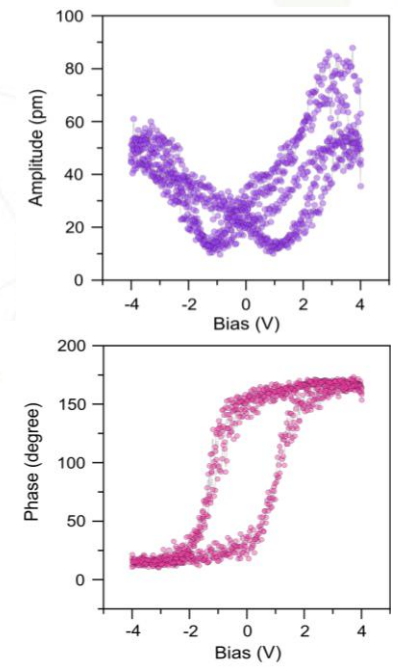
2D materials from high-melting-point metals



$$\text{SnO}_2 \Delta G_f = -578 \text{ kJ/mol}$$



$$\text{HfO}_2 \Delta G_f = -1145 \text{ kJ/mol}$$

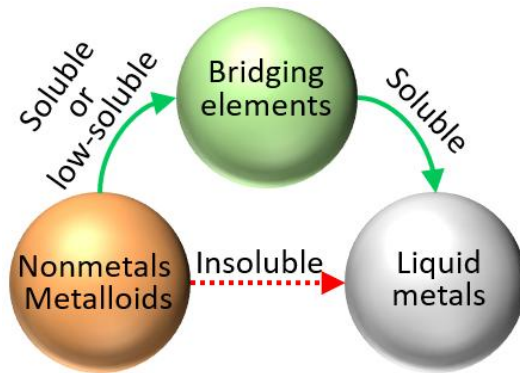


$$\text{ZrO}_2 \Delta G_f = -1100 \text{ kJ/mol}$$

Bridge solubility in LMs

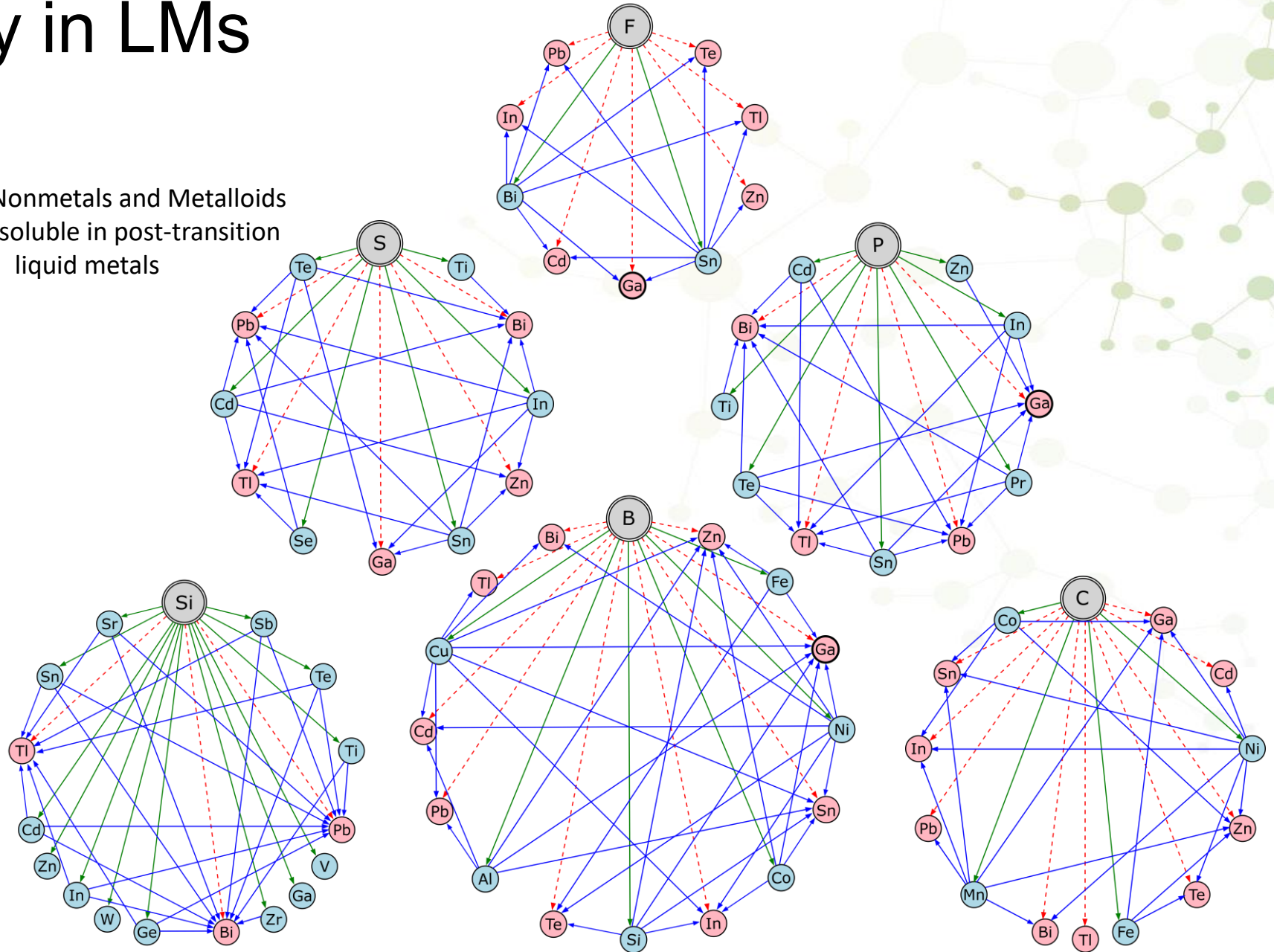
5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen	9 F Fluorine	10 Ne Neon
13 Al Aluminium	14 Si Silicon	15 P Phosph...	16 S Sulfur	17 Cl Chlorine	18 Ar Argon
31 Ga Gallium	32 Ge Germani...	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton
49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon
81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon
113 Nh Nihonium	114 Fl Flerovium	115 Mc Moscovi...	116 Lv Livermor...	117 Ts Tenness...	118 Og Oganess...

Reactive Nonmetals and Metalloids
Mostly insoluble in post-transition
liquid metals

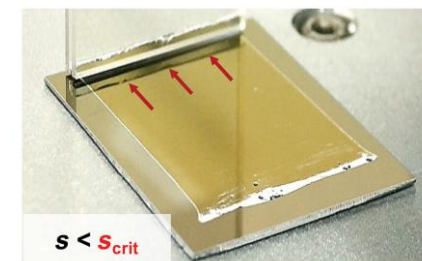
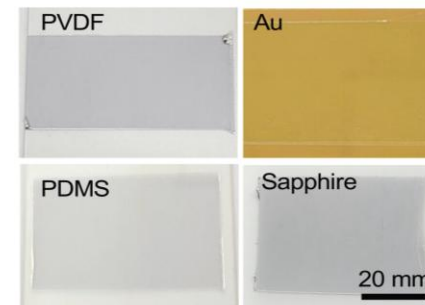
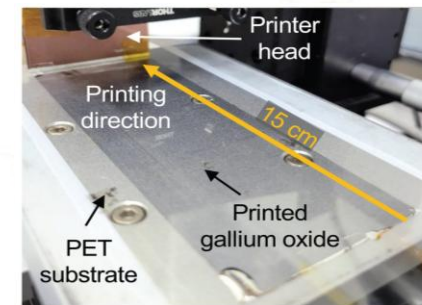
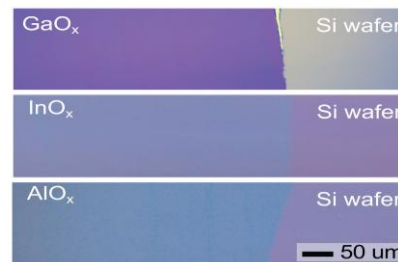
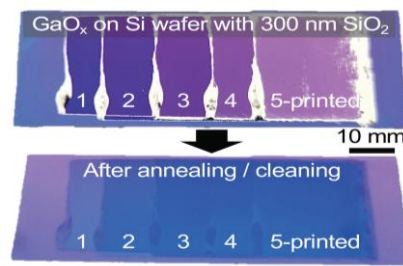
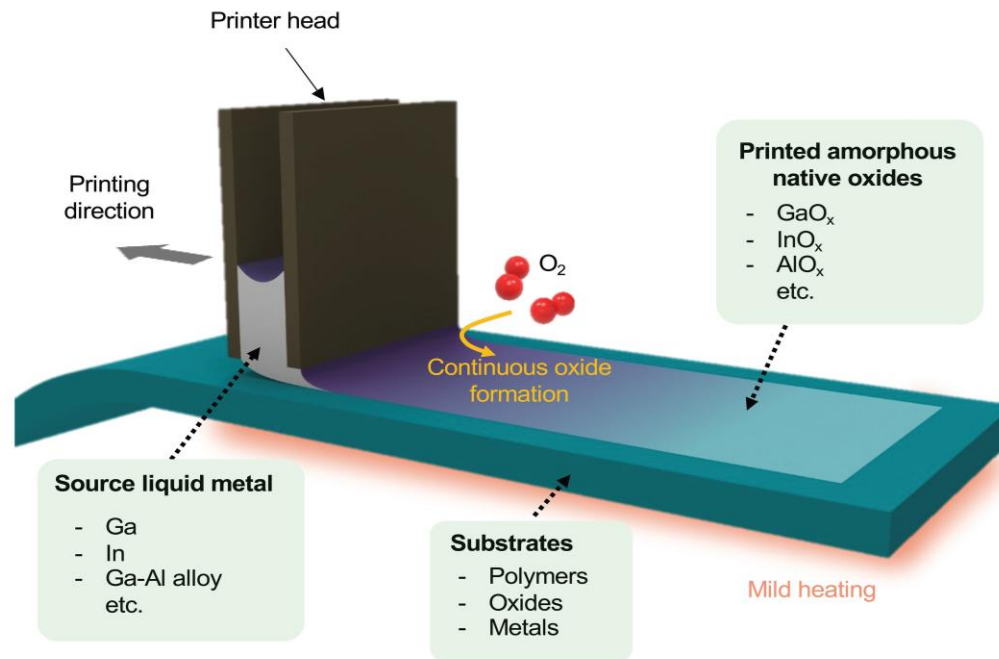


An element with natural solubility in both liquid metals and nonmetals/metalloids, acts as a bridge or carrier between the two insoluble phases.

Solubility criteria: least 10 at% at below 1500 °C.



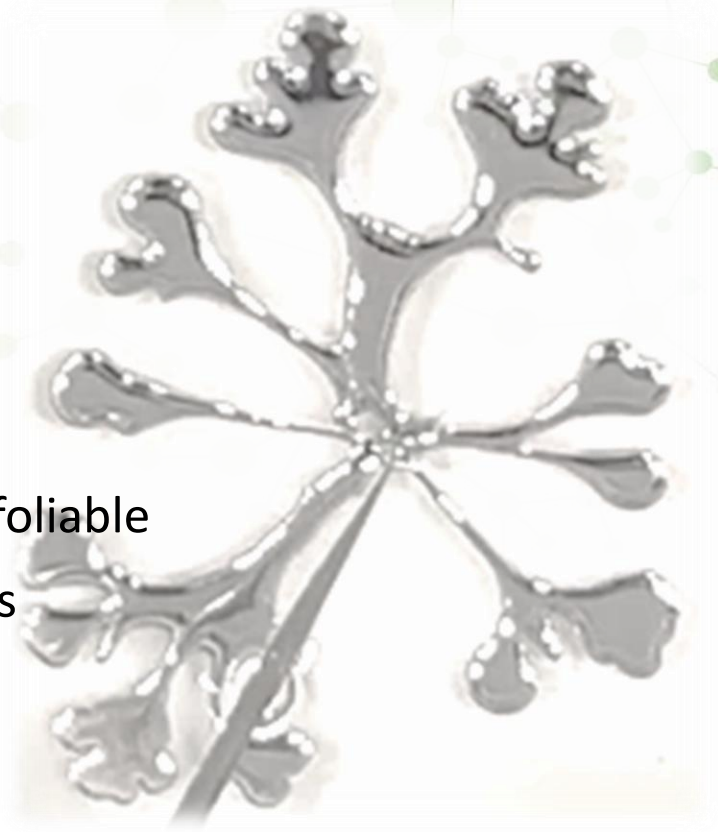
Large Scale Production of 2D Materials by LM



Summary

Advantages of the liquid metal process for the synthesis of nanoparticles and doping of 2D materials:

- Low or moderate working temperature
- Simultaneous functionalization
- Vacuum free
- Large smooth atomically thin sheets
- Thickness control
- Access to single atomic layers that are not intrinsically layered and exfoliable
- Introducing dopants at desired concentrations and avoiding impurities
- No need for extremely clean environments (cleanrooms)
- No need for sophisticated and costly instruments and processes
- Adoptability with low temperature roll to roll processes



Thank You

