

Materials by Design and Advances in Photovoltaic R&D



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UNSW Seminar

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Photovoltaics Summary

Rapid progress has been made in PV but **we aren't done yet**

Solar energy can deliver low-carbon energy to mitigate Climate Change BUT further advances are needed for TWs

- Grid parity \leq 2020
- Systems approach (module, BOS/soft costs, reliability, grid integration)
- Policy/markets; Utility models, Financing

Beyond Grid Parity with significant further cost reductions (2-3¢/kW-hr)

- Next-Gen technologies: new materials, concepts and processes for high efficiency, low cost, AND manufacturability
- Novel processing technologies (low CAP-EX mfg)
- Mitigate devaluation of solar at high penetration
- Grid flexibility, energy mix, and low cost energy storage

Solar energy can also provide power to the underserved

- Multi-scale approaches to energy systems
- Distributed and dispatchable energy, microgrids, storage

On the Path TO SunShot

ON THE PATH TO SUNSHOT:
EXECUTIVE SUMMARY

Solar Energy Technologies Office
U.S. Department of Energy



- Since 2011, costs down 65% and 70% towards **grid parity goals**
- 8 reports DOE and 4 National Labs (NREL, Berkeley, Argonne, Sandia)
 - Lessons Learned; Challenges/Opportunities

PHOTOVOLTAIC EFFICIENCY, RELIABILITY, AND COSTS

ADVANCING CONCENTRATING SOLAR POWER TECHNOLOGY

U.S. SOLAR MANUFACTURING

INTEGRATING HIGH LEVELS OF SOLAR INTO TRANSMISSION

INTEGRATING HIGH LEVELS OF SOLAR INTO THE DISTRIBUTION SYSTEM

FINANCING SOLAR

UTILITY REGULATION AND BUSINESS MODEL FOR FINANCIAL IMPACTS

ENVIRONMENTAL AND PUBLIC HEALTH BENEFITS

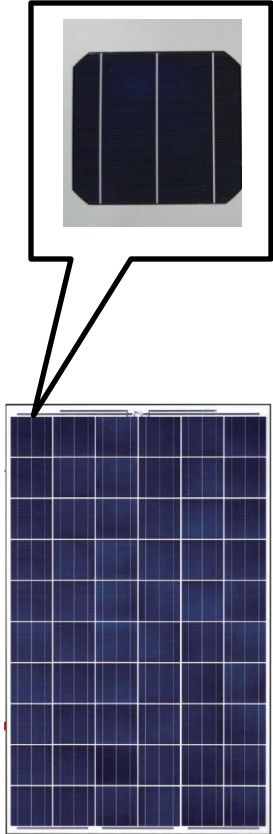
On the Path TO SunShot

- Continued innovation in system-level S&T critical for grid parity and beyond
- Need to pursue multiple strategies to maintain the value (costs + benefits) of solar
- Increasing grid flexibility, next-gen power electronics and other strategies could enable 25% solar
- Monetizing environmental benefits could add ~3.5¢/kWh to the value of solar energy

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NREL Solar R&D: Materials, Cells, Modules, Systems



Cost, Performance

Manufacturability

PV and System Reliability

Analysis

Balance of System & Soft Costs

Grid Integration

Energy Storage

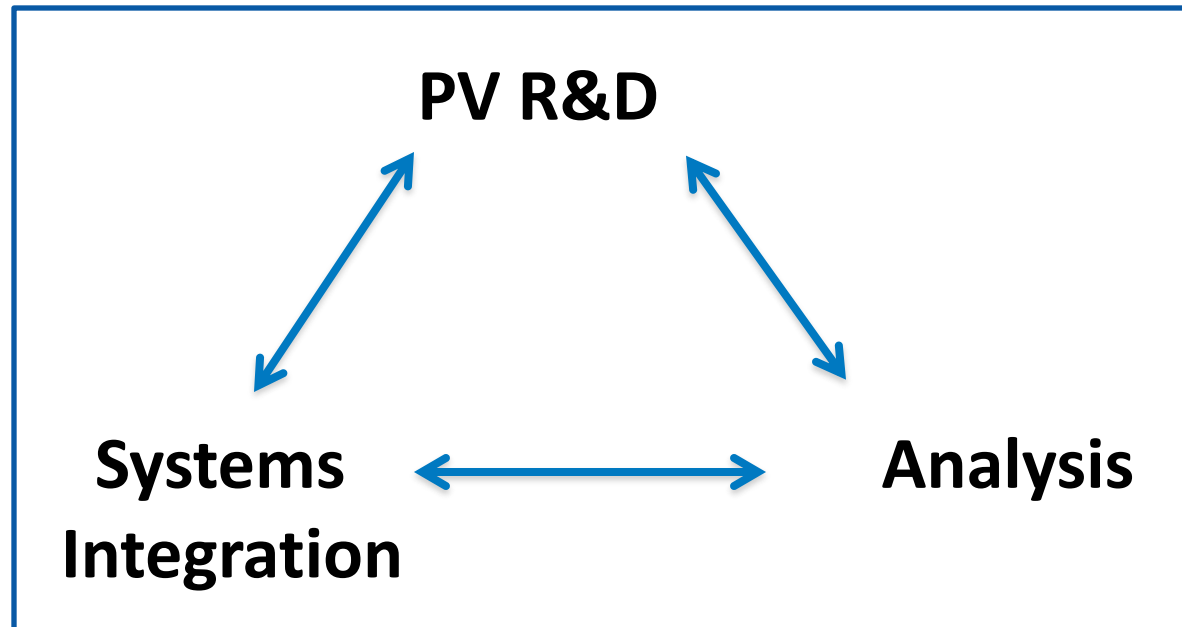


NREL Solar R&D: Materials, Cells, Modules, Systems

Understand limitations and enhance performance in current systems

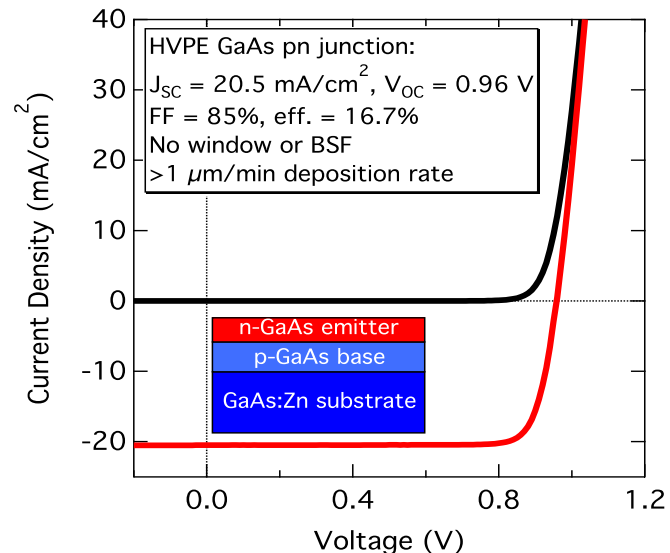
Accelerate emerging concepts

Develop next generation concepts and materials



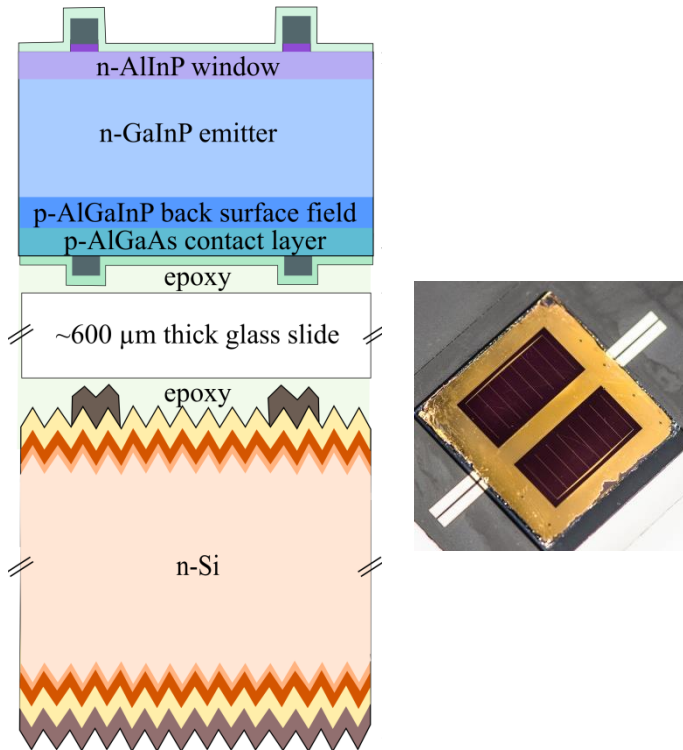
Hydride Vapor Phase Epitaxy for GaAs

Now >23% GaAs homojunction cell
(no cladding layers) :



- Dual chamber HVPE reactor for Ga, In, As, P alloys
 - full 3D computational fluid dynamics (CFD) modeling
- Produced epitaxial GaAs materials at growth rates exceeding $1.8 \mu\text{m}/\text{min}$
- Can produce flat, parallel, low-defect homo- and hetero-interfaces
- Demonstrated very high metal utilization ($\sim 70\%$ for Ga)

Development of World Record GaInP/Si Dual-Junction, One-Sun Solar Cell

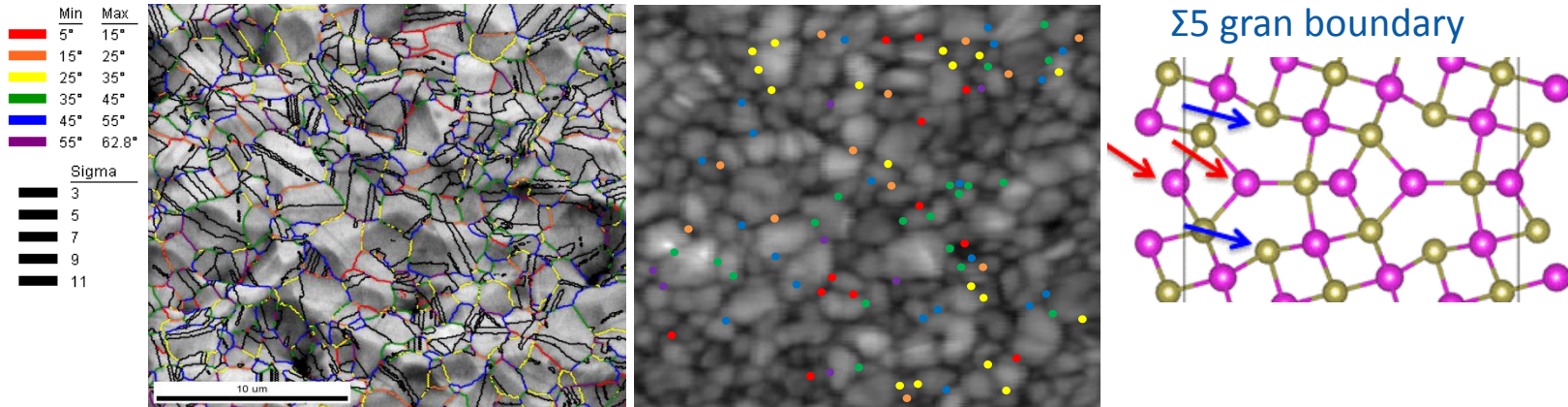


1.8-eV GaInP top junction
with a silicon bottom
junction, with a four-
terminal interconnection

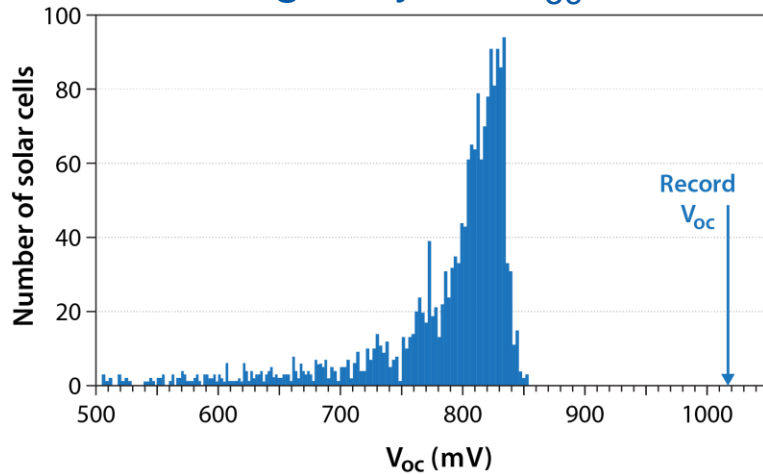
- A two-junction structure with a silicon bottom junction
- New device structure combining a III-V GaInP top junction and a silicon bottom junction,
- Demonstrated a world record 29.8% efficiency – significantly exceeding the best conventional silicon efficiency of 25.6%.
- Four-terminal structure allows ease of construction, and optimal energy production under real-world operating conditions.
- Developing an improved, manufacturable bonding

CdTe Technology

Cataloging the role of GBs, surfaces and bulk defects



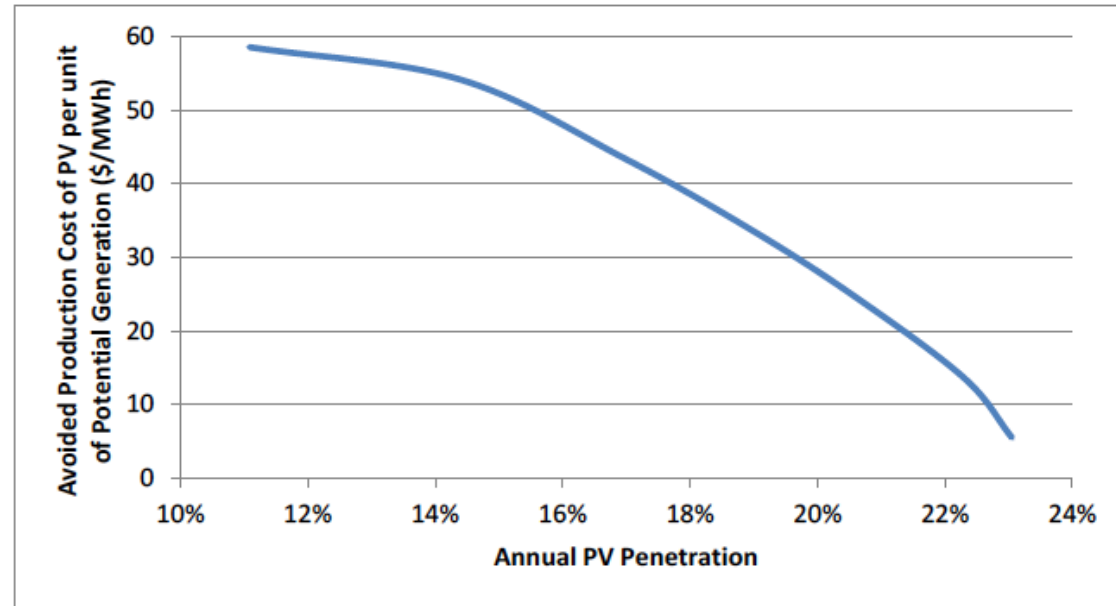
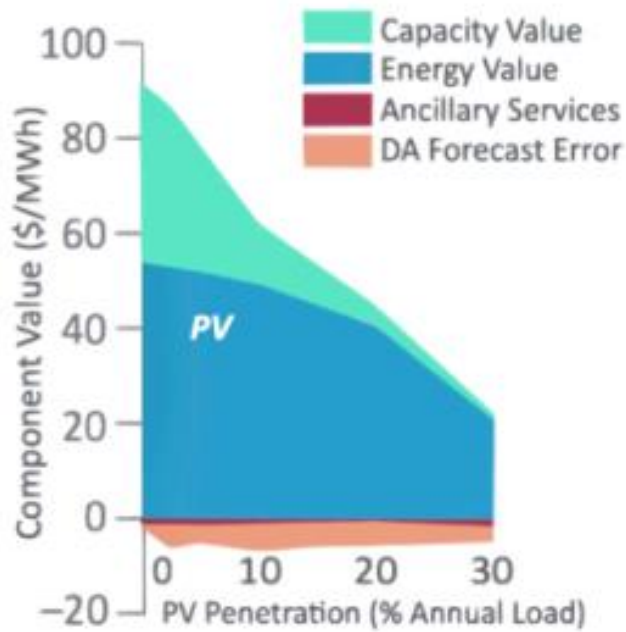
Overcoming 20-year V_{oc} barriers



Histogram of V_{oc} values for about 2200 polycrystalline CdTe devices

- Worked w/o universal $CdCl_2$ treatment
- Switched to anion Group V doping
- Shifted to Cd-rich stoichiometry to Improve lifetime by removing Te_{Cd} antisites, and hole density by placing P on Te sites.

Solar Devaluation with Increasing Deployment



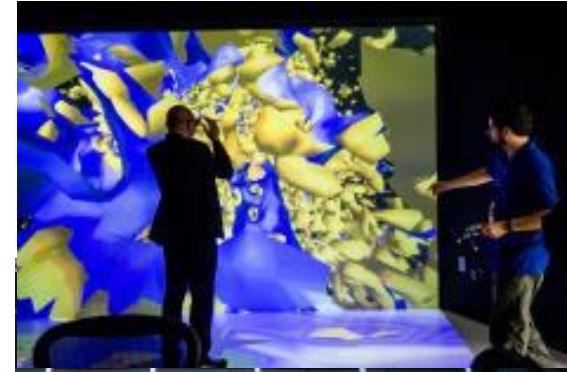
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Mills, Wiser, LBNL, 2012

Denholm, NREL, 20126

Grid Modernization Lab Consortium

- Grid Modernization Laboratory Consortium involves 14 DOE national laboratories and industry, academia, and state and local government partner
- Energy Systems Integration Facility
 - Multiple parallel AC and DC experimental busses (MW power level) with grid simulation
 - “Hardware-in-the-loop” simulation capability to test grid scenarios with high penetration of renewables
 - Peta-scale high-performance computing and data management system
 - Virtual utility operations center and visualization rooms

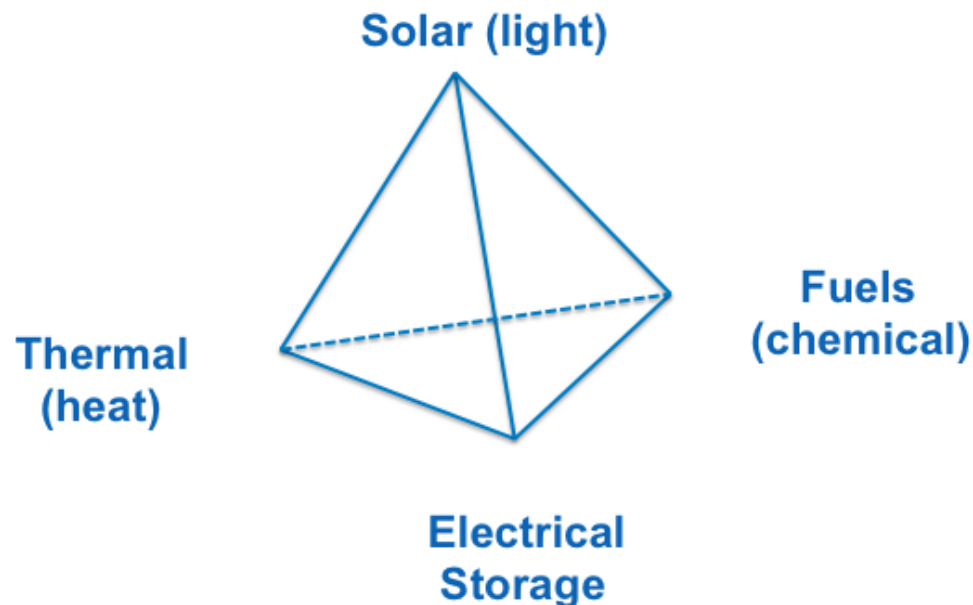


NREL Energy Systems
Integration Facility
(ESIF)
Research and Testing

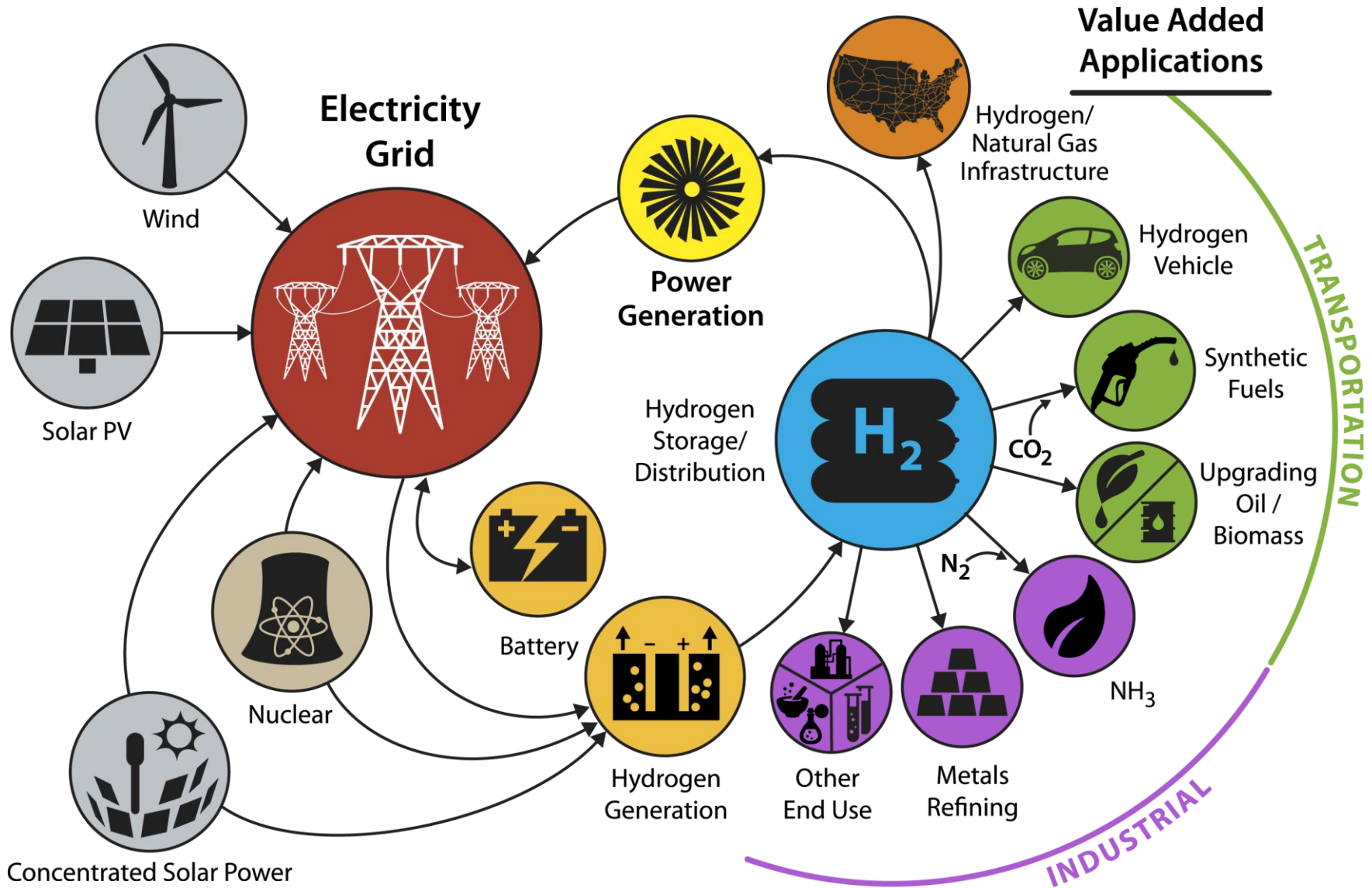


Grid Modernization Lab Consortium

- Flexible Grid
- Vehicles to Grid, Buildings
- Water purification/desalination
- Fuels, Chemicals
 - CO_2 reduction; $\text{N}_2 \rightarrow \text{NH}_3$; C



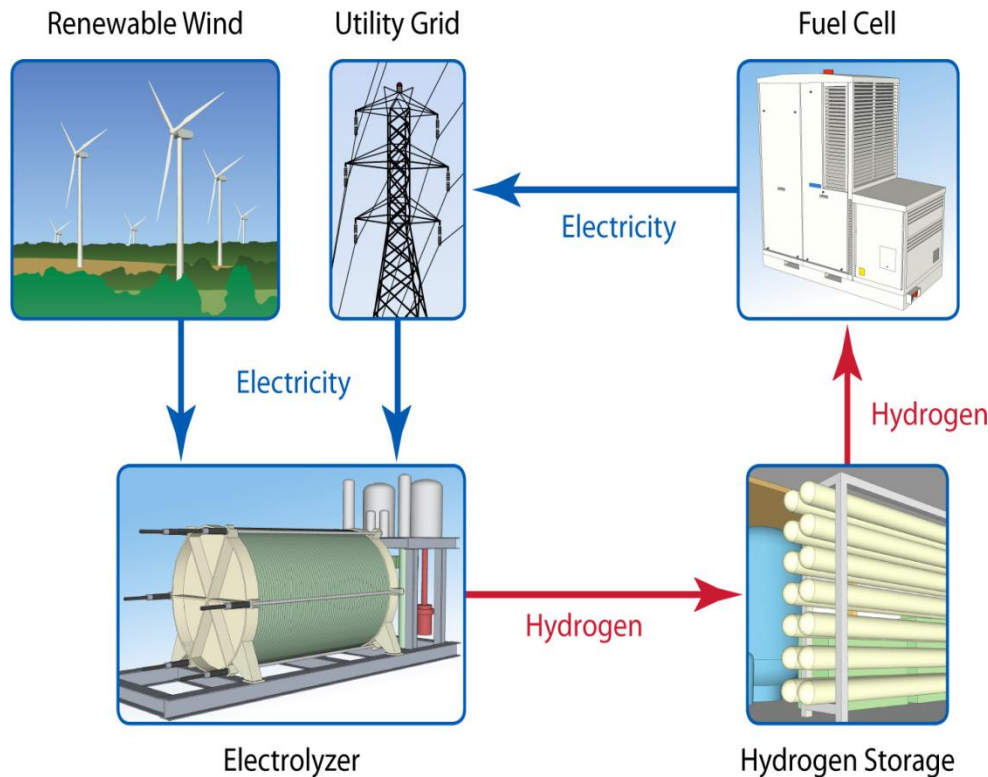
Conceptual H₂@Scale Energy System



*Illustrative example, not comprehensive

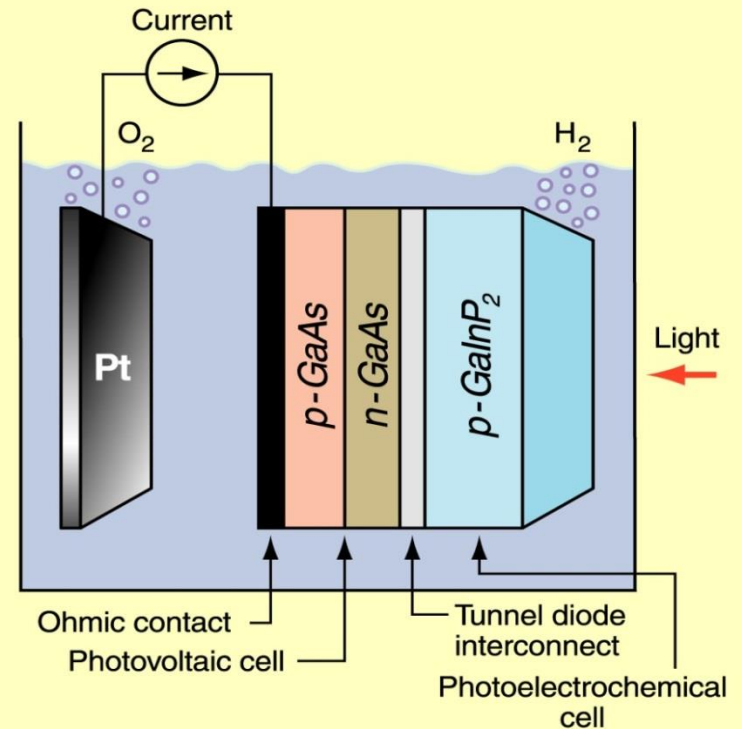
Solar Hydrogen Generation for Energy Storage

PV-Electrolysis



Photoelectrochemical Water Splitting

Novel cell uses light to produce H_2 at 12.4% efficiency



Note: *n* and *p* refer to *n*- and *p*-type semiconductors

Credit: Adapted with permission from Science, copyright 1996 AAAS

World Record: Photoelectrochemical (PEC): $\text{H}_2\text{O} \rightarrow \text{H}_2 + \frac{1}{2}\text{O}_2$

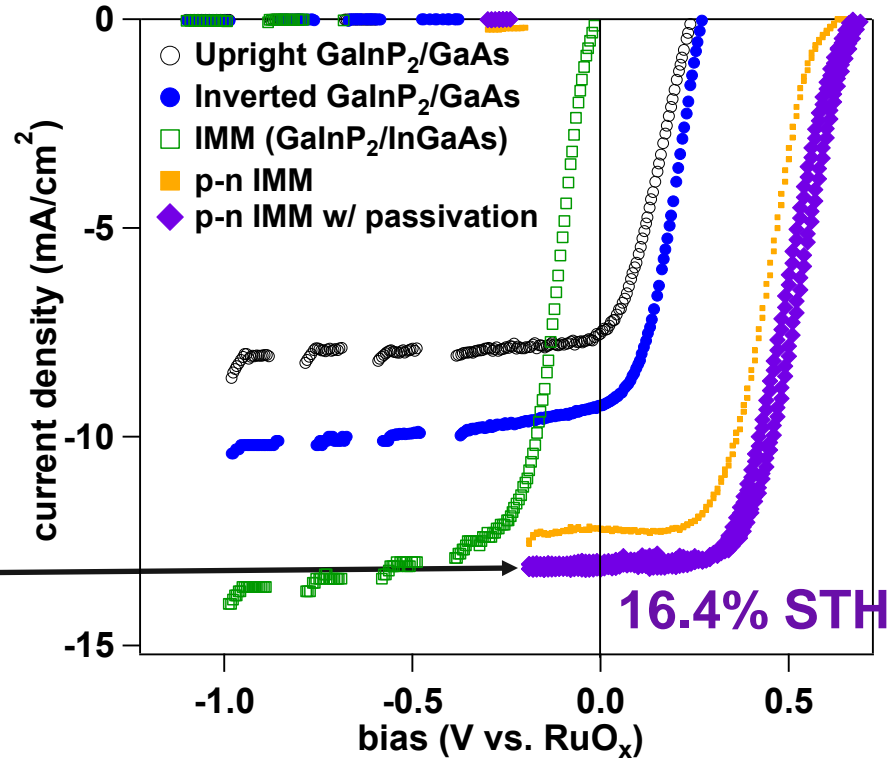
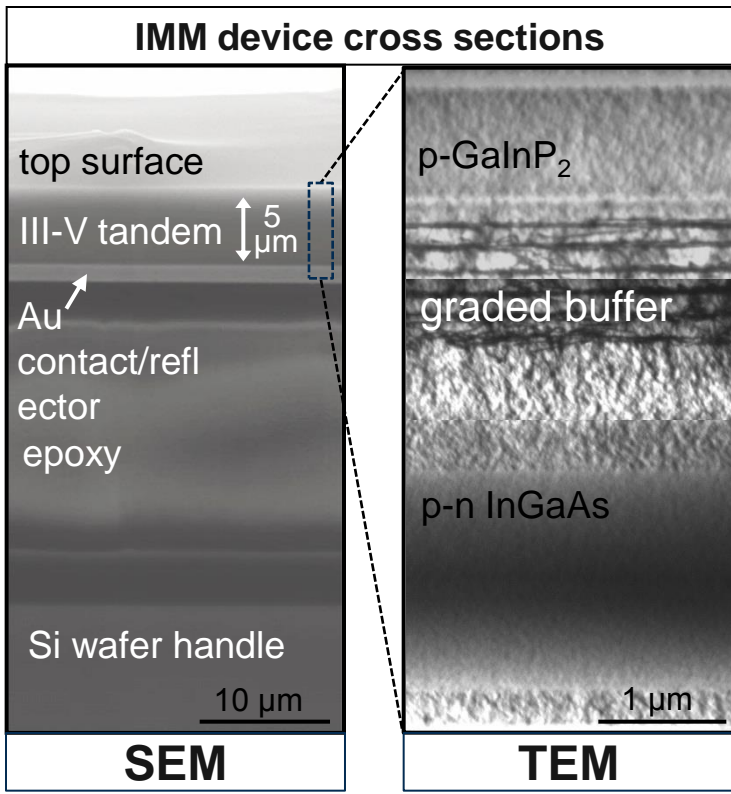
Technology

- **Inverted metamorphic multijunction (IMM) PEC device** enables more ideal bandgaps
- Grown by **organometallic vapor phase epitaxy**
- Incorporates **buried p/n GaInP₂ junction** and **AllnP passivation layer**

Solar-to-hydrogen Efficiency

16.4%

Benchmarked under outdoor sunlight at NREL



Credit: NREL

New ultrafast laser spectroscopy technique uncovers how photoelectrodes produce solar hydrogen from water

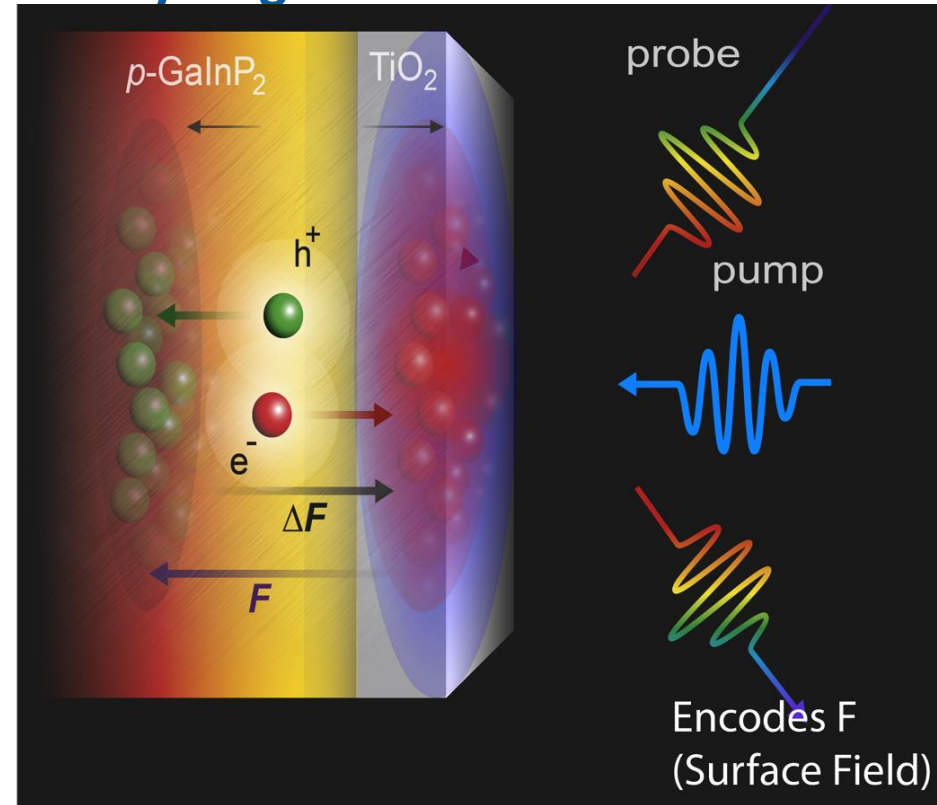
Semiconductor photoelectrodes convert solar energy directly into chemical fuels

NREL's new probe measures transient electrical fields and shows how semiconductor junctions convert sunlight to fuels

The field formed by the TiO_2 layer drives electrons to the surface where they reduce water to form hydrogen.

The oxide prevents photocorrosion by keeping holes away from the surface

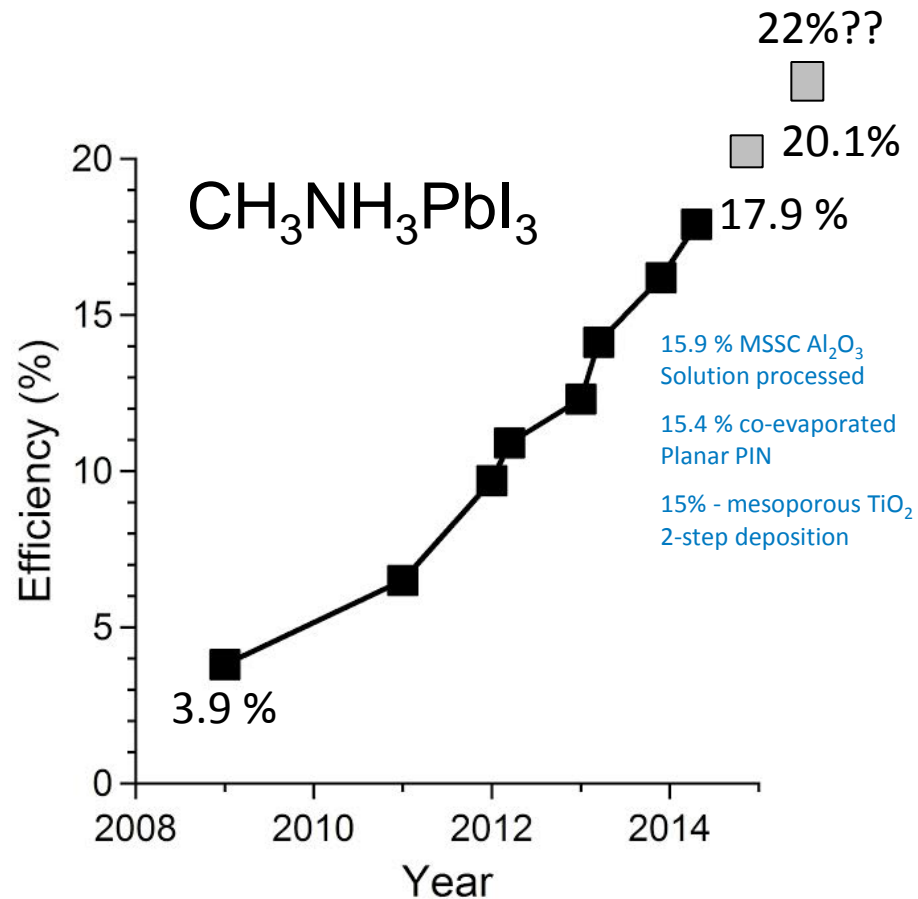
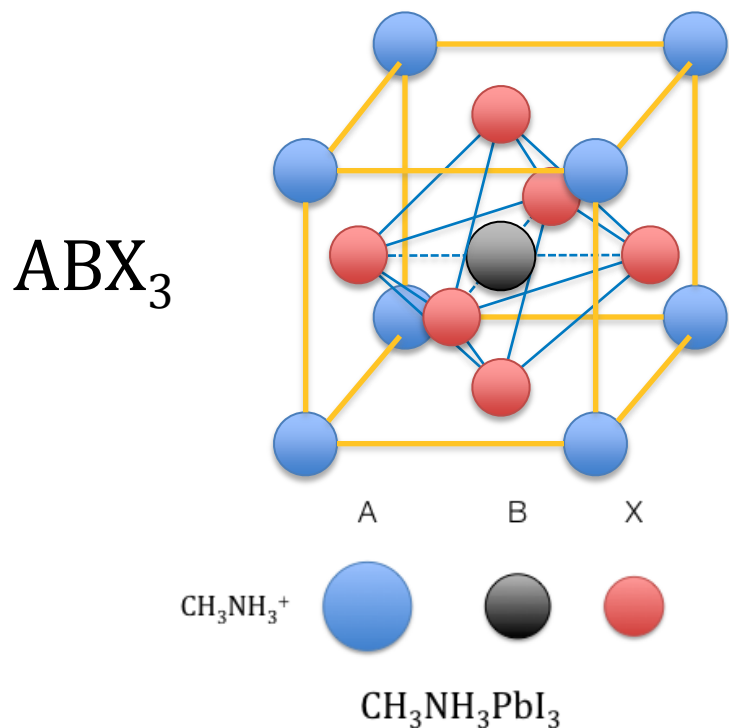
This new understanding will lead to more stable and efficient solar fuel generators



The transient photoreflectance (TPR) technique measures short-lived electrical fields that arise due to charges generated by light that are driven in opposite directions by the properties of the interface.

Ye Yang et al, Science **350**, 1061-1065, (2015)

Perovskites



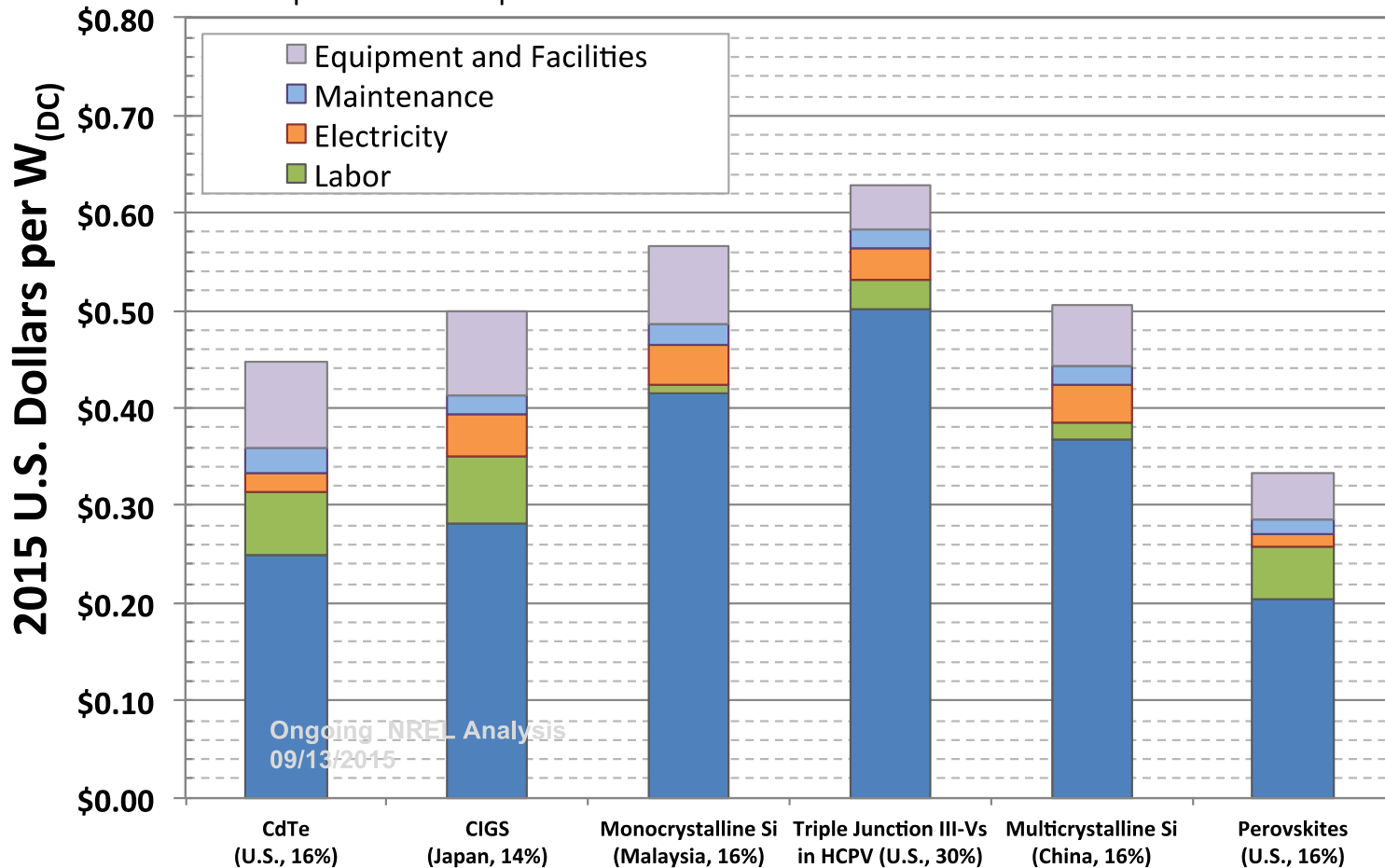
- Minority carrier diffusion lengths $> 1\mu m$ in thin films, $175\mu m$ in single xtal c
- Monomolecular recombination lifetimes of 280 ns
- Minority carrier mobilities $\sim 10\text{ cm}^2/V/s$ are reasonable
- High $\epsilon_r = 60-70$; Low $m^* = 0.1, 0.16$

Molecular approaches to solution-processable, defect-tolerant GaAs

Potential Costs

16% Perovskite compared to other PV

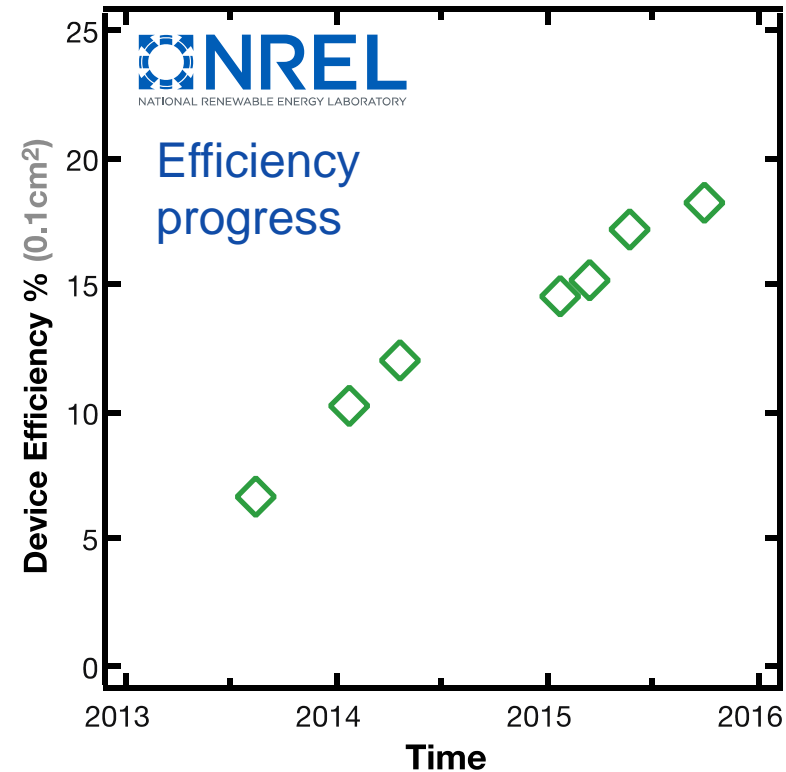
Aggregated Production Costs for Selected Photovoltaics Technologies
Representative Inputs Collected and Reviewed Over 2014 and 2015.



Stable perovskite PV meets 2020 targets (Woodhouse/NREL)

NREL R&D Themes

- › Basic understanding of photophysics & transport
- › Theory and modeling
- › Discovery
- › Device fabrication and characterization
- › Synthesis & processing
- › Interfaces
- › Device operation & physics
- › Stability and degradation mechanisms



Fundamental Perovskite work at NREL

Hot carrier dynamics

- Phonon bottle neck carrier cooling rate ($\sim 3x$ more efficient than GaAs) - *Nature Photonics* (2015)

Role of excitons

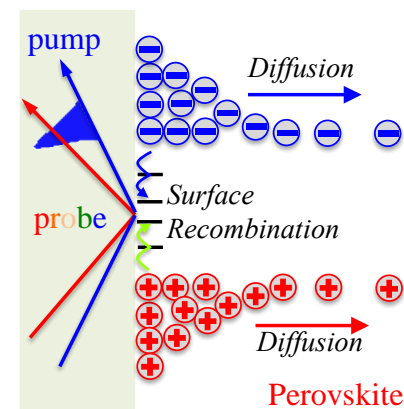
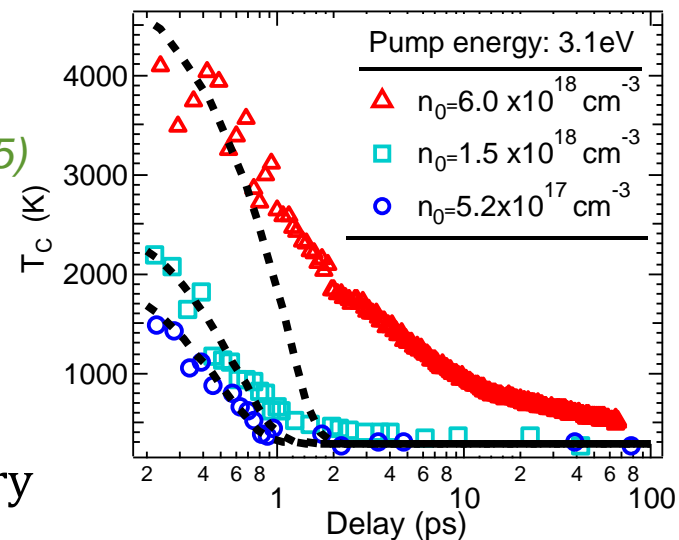
- Excitons enhance absorption and modify recombination *J. Phys. Chem. Lett.*, 6, 4688-4692, 2015

Surface recombination

- Intrinsic surface recombination velocity is very low *Nature Comm*, 2015, 6, 7961
- Difference in single crystal and thin film surface recombination velocities
- Grain boundaries impact on recombination

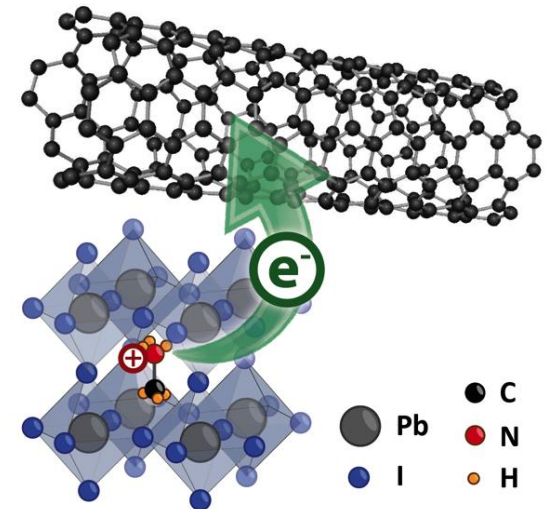
Interface charge transport

- Substrate controlled electronics (Kahn/Princeton)
- SAM layer for enhanced charge separation (Snaith/Oxford, Ginger & Jen/UW, Friend/Cambridge)

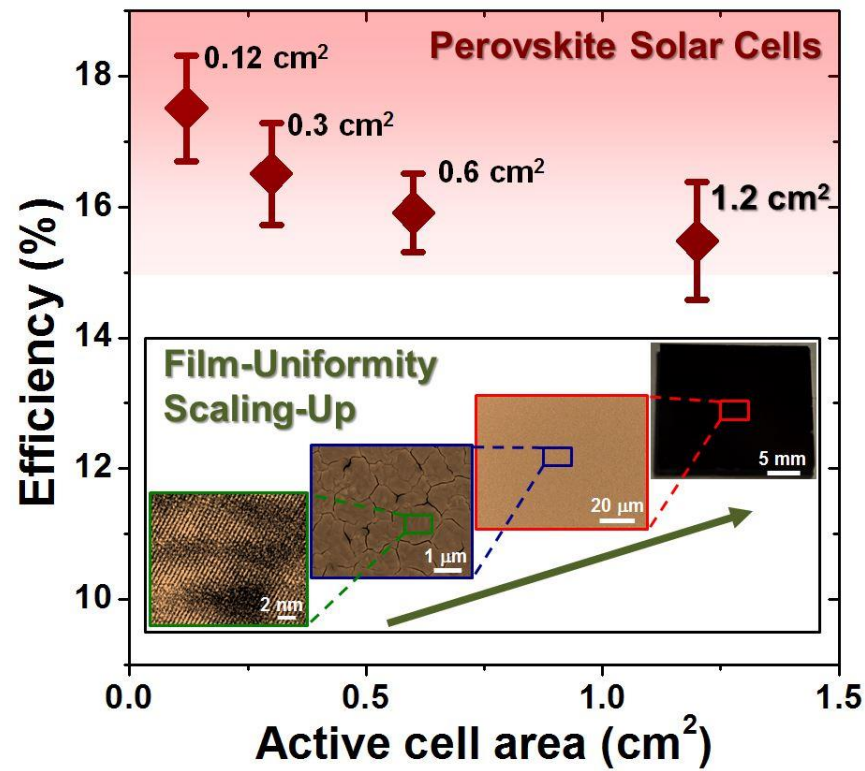
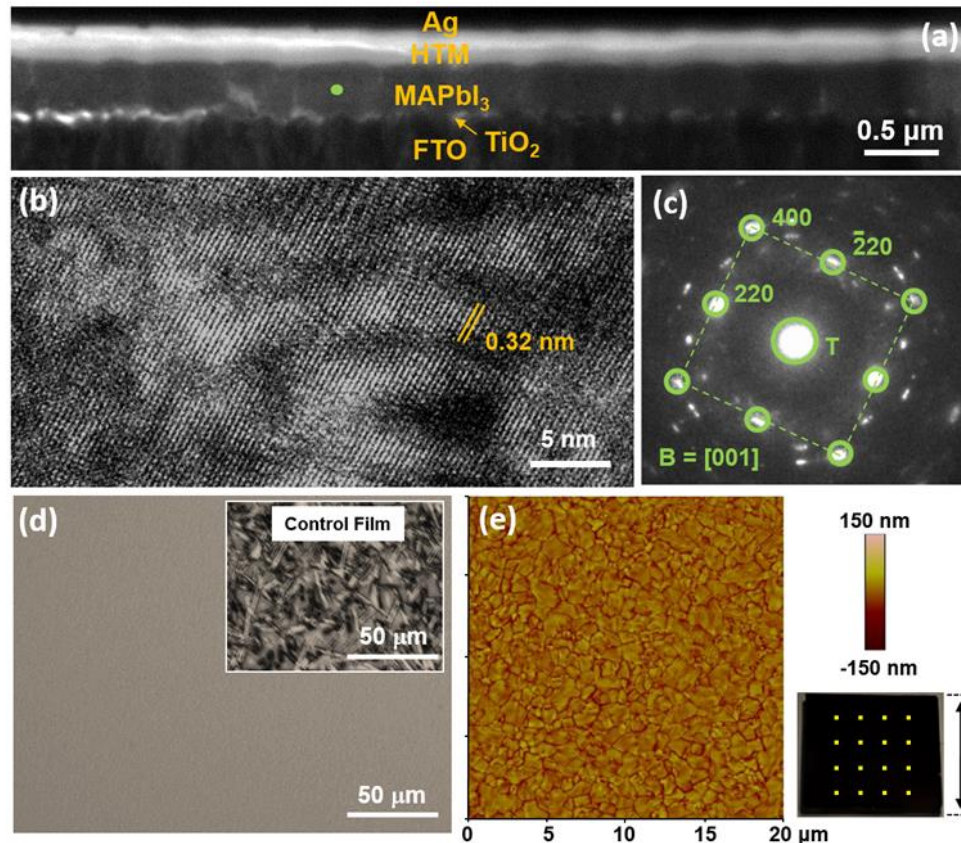


Some Perovskite Advances

- Band bending at model SWCNT:perovskite interface using PES. Ultrafast spectroscopy shows efficient photoexcited hole extraction
- J.Phys. Chem. Lett (2016)
- Developed new tools and techniques to evaluate absorber structure as function of processing. Using quantitative x-ray diffraction at SLAC showed that high efficiency device structures have a large amount of material in amorphous phase.



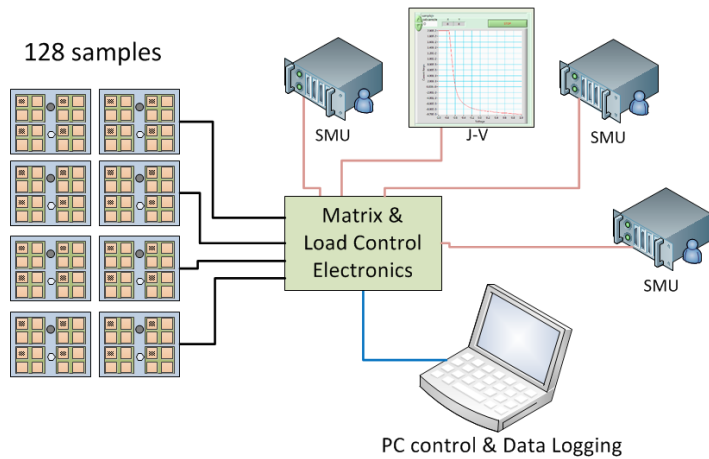
Square-Centimeter Solution-Processed Planar MAPbI₃ PSC with PCE >15%



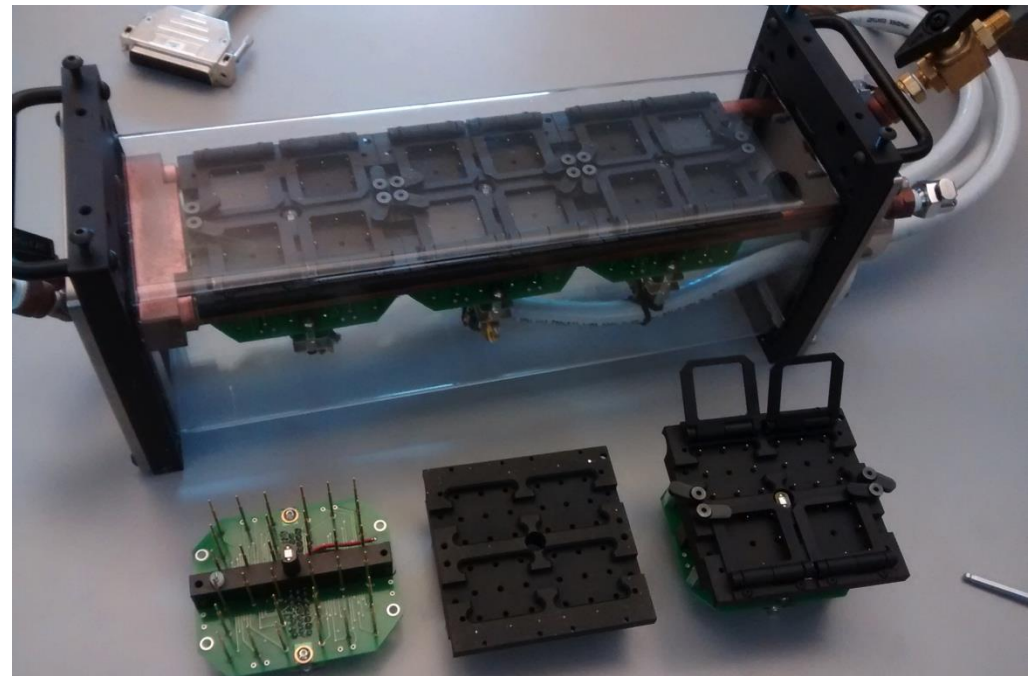
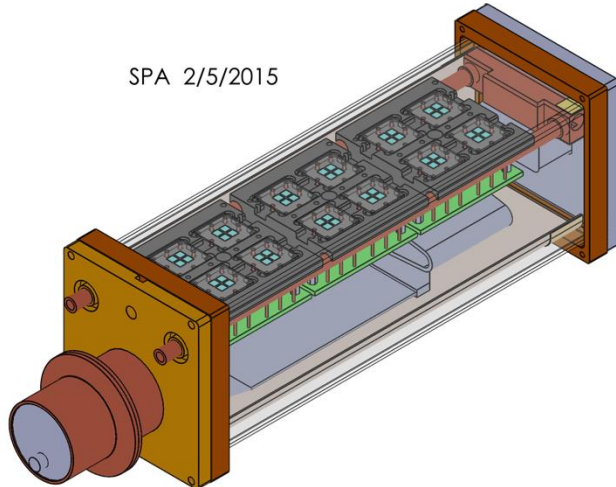
Novel solution chemistry for uniform, high-crystallinity, planar perovskite films with high-aspect-ratio grains over a square-inch area; and >15% efficiency PSC with 1.2 cm² active area

New Device Level Stability Capabilities

Functional device studies using combinatorial device testing rig
New stability parameter analysis systems



SPA 2/5/2015



- › **Flow cell geometry, controlled temperature, humidity and atmosphere**



Structural Materials

lightweight alloys & composites
cars, buses, wind turbines fuel tanks...

advanced optical absorbers
PV, solar fuels, solar purifiers...

Optical Materials

efficient solid-state converters
thermoelectrics, LED lighting...

high temp materials
CSP, combustion, geothermal...

advanced membranes
batteries, fuel cells, purifiers...

WBG semiconductors
power electronics, grid integration..

energy storage materials
thermal, thermochemical.

electrode materials
batteries, fuel cells, electrolyzers...

Thermal Materials

alternative fuels
advanced combustion, bio/synthetic fuels...

non precious catalysts
fuel cells, electrolyzers, reformers...

Electronic Materials

Chemical Materials

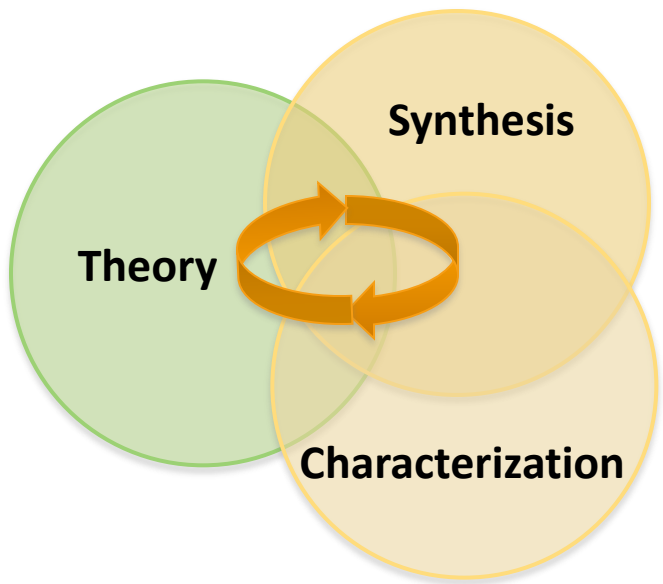
Numerous materials classes are vital to "Clean Energy Technologies"

1. Design and discover new energy-relevant materials with targeted functionalities.
2. Develop foundational theoretical, synthesis, characterization tools.
3. Incorporate functional metastable materials into MbD.
4. Develop a systematic theory-driven approach to guide synthesis.

Address four Critical Gaps limiting Materials by Design

1. Multiple-Property Design
2. Accuracy and Relevance
3. Metastability
4. Synthesizability

CNGMD Team Integrates Theory, Experiment, Data



LBL-Berkeley – Gerd Ceder (Chief Theorist)
Kristin Persson
NREL – Stephan Lany
CSM – Vladan Stevanovic
MIT – Alexie Kolpak

NREL – David Ginley (Chief Experimentalist)
Andriy Zakutayev
CSM – Brian Gorman
MIT – Tonio Buonassisi
Harvard – Dan Nocera
Roy Gordon
OSU – Janet Tate
SLAC – Mike Toney

NREL – Bill Tumas (Director), John Perkins (Program Integrator)

Theory Tools Development: DFT Improvements

● GW Corrections for bandgaps

- One empirical parameter per TM atom, good transferability to ternaries

S. Lany, *Phys. Rev. B* **87**, 085112 (2013)

● Fitted Elemental Reference Energies (FERE) for heats of formation

- $\mu^0(\text{FERE}) = \mu^0(\text{GGA+U}) + \delta\mu^0(\text{FERE})$
- FERE reduces the mean average error (MAE) to 0.054 eV/atom \approx 1 kcal/mol

Stevanovic et al. *Phys. Rev. B* **85**, 115104 (2012)

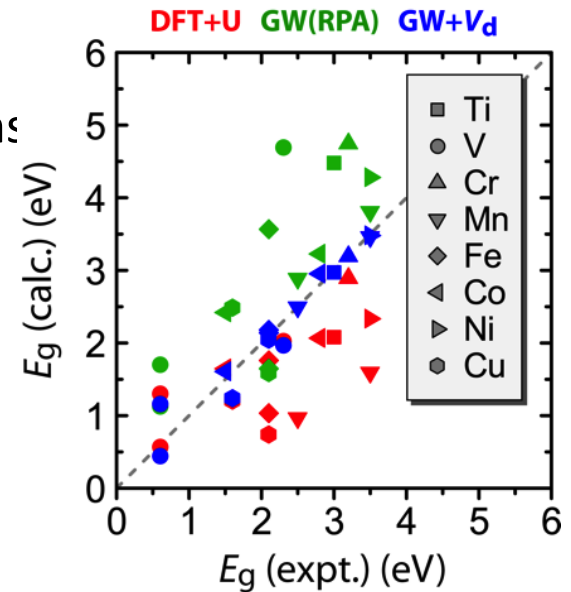
● SCAN functional to reproduce basic MnO_2 properties

- First functional to simultaneously reproduce strong covalent-like bonds and weaker long-range interactions
- J. Perdew (Temple), CCDM EFRC

Kitchaev et al. *Phys. Rev.* **93**, 045132 (2016)

● Neural Networks for large-scale Defects

A. Kolpak et al., in progress



Materials.NREL.gov

High-throughput calculations

DFT level (atomic structure and total energy)

ca. 20,000 crystalline ordered materials

- Repository of atomic structures
- Thermochemistry and stability

GW level (electronic structure)

ca. 250 semiconducting and insulating materials

- So far: Mostly oxides, chalcogenides, nitrides
- Direct and indirect band gaps
- Band-edge shifts wrt DFT (defects, IP/EA, band offsets)
- Effective masses, density of states
- Optical properties, dielectric function, absorption spectra

Required elements subset: At most these elements Exactly these elements At least all these elements
Formula: exactly this formula

Required elements: Forbidden Elements:

Restrictions:

(Restriction info: energy (total energy, per atom, eV/Atom), gap (bandgap, eV), gapd (direct bandgap, eV), netcharge (net charge), pressure (final pressure, Kbar), stable (qhull stability)
Example: gap > 0.2 and gap < 0.8 and netcharge = 0 and stable = true

Standards: (Examples: fere, gwvd, post-lopt) [Standards help](#)

Keywords: (Examples: bandgap, enthalpy, ggau, ggauwf, gwvd, hse, hsefere, hsewf, vexp)

View: General Thermochemistry GW_Bandgap (Changing the View and clicking Submit will reset the standards and columns.)

Show columns: id sorted formula ΔH_f (eV/atom) Stb ΔH_{Gnd} ΔH_{Decomp}
 ICSD SG final SG E_{tot} (eV/Atom) minID ΔE_{cbm} ΔE_{vbm} ϵ_c (IP) ϵ_c (TDDFT)
 ϵ_c (post-lopt) E_g (eV) $E_{g,d}$ (eV) netCharge pressure mag mom numSpin icstdId
 lastName standards keywords parents

Sort order:

Hide dups: (Hide duplicate FERE entries for each formula and ICSD space group.)

Output format: HTML CSV

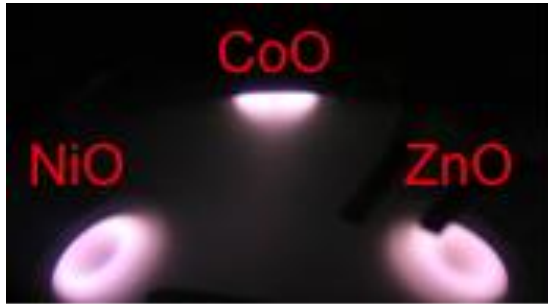
Submit Query

Total matches: 238 Page length: 100 Page:

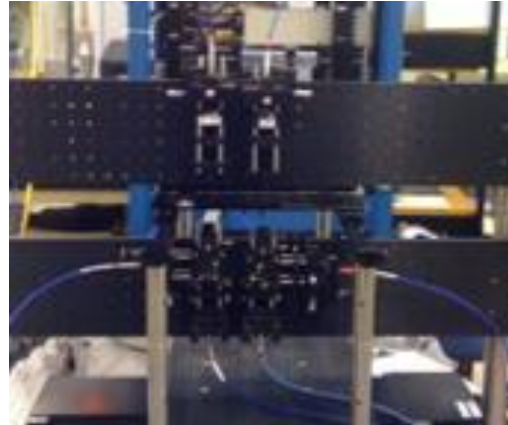
id	sorted formula	final SG	ΔE_{cbm}	ΔE_{vbm}	ϵ_c (TDDFT)	E_g (eV)	$E_{g,d}$ (eV)	numSpin	parents	Family
8298	Al2 Cd O4	227	1.138	-1.420	2.960	5.716	5.716	1	[8982]	family 8027
9404	Al2 Co O4	227	1.602	-1.455	2.913	6.240	6.334	2	[7725]	family 8595
8686	Al2 Cu O4	141	-0.297	-1.422	3.308	2.291	2.810	2	[8699]	family 6992
7915	Al2 Mg O4	227	1.404	-2.004	2.530	8.896	8.905	1	[8837]	family 8983
8048	Al2 Mn O4	227	1.259	-1.719	2.831	6.157	6.157	2	[8622]	family 9022
8404	Al2 Ni O4	74	1.840	-1.205	3.047	5.676	5.676	2	[9310]	family 8217
9497	Al2 O3	167	1.619	-1.883	2.681	9.753	9.753	1	[7984]	family 8662
8146	Al2 O4 Zn	227	1.393	-1.696	2.767	7.554	7.554	1	[8767]	family 9110
8783	Al2 S3	167	0.248	-1.317	5.836	4.042	4.392	1	[8923]	family 7153
8038	Al2 Se3	9	0.504	-1.141	4.821	3.720	3.733	1	[9181]	family 9146
8192	Al As	216	-0.080	-0.674	7.764	2.029	3.237	1	[8876]	family 7982
9008	Al Cu O2	166	0.993	-1.079	3.898	4.385	5.077	1	[10357]	family 8973
7829	Al N	186	0.834	-1.269	3.694	6.345	6.345	1	[8714]	family 8898
8581	Al N	216	0.566	-1.262	3.752	5.152	6.268	1	[8471]	family 6936
9322	Al N	225	1.021	-1.092	4.267	6.731	7.909	1	[7518]	family 8536
7942	Al P	216	-0.004	-1.092	6.599	2.659	4.519	1	[8441]	family 8899
10491	As Ga	216	0.538	-0.528	10.607	1.625	1.625	1	[10444]	family 9770
10824	As In	216	0.268	-0.400	16.819	0.484	0.484	1	[11172]	family 10144
10728	B2 O3	144	1.701	-2.168	2.364	10.349	10.662	1	[13405]	family 10914
8515	B2 O	225	0.588	0.859	2.560	3.260	3.260	1	[04251]	family 8220

NREL High-Throughput Experimental Tools

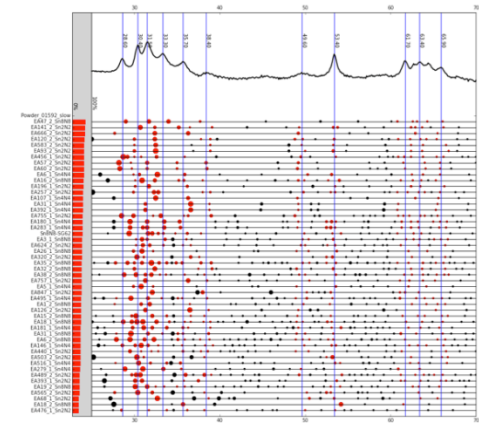
Thin Film Deposition



Property Mapping



Analysis & Visualization



4 Dedicated PVD System

- 3 RF sputtering
- 1 PLD
- Composition Gradients
- Thickness Gradients
- Temperature Gradients
- Ar, N, O, Ar/H₂S gasses
- Atomic S & N sources

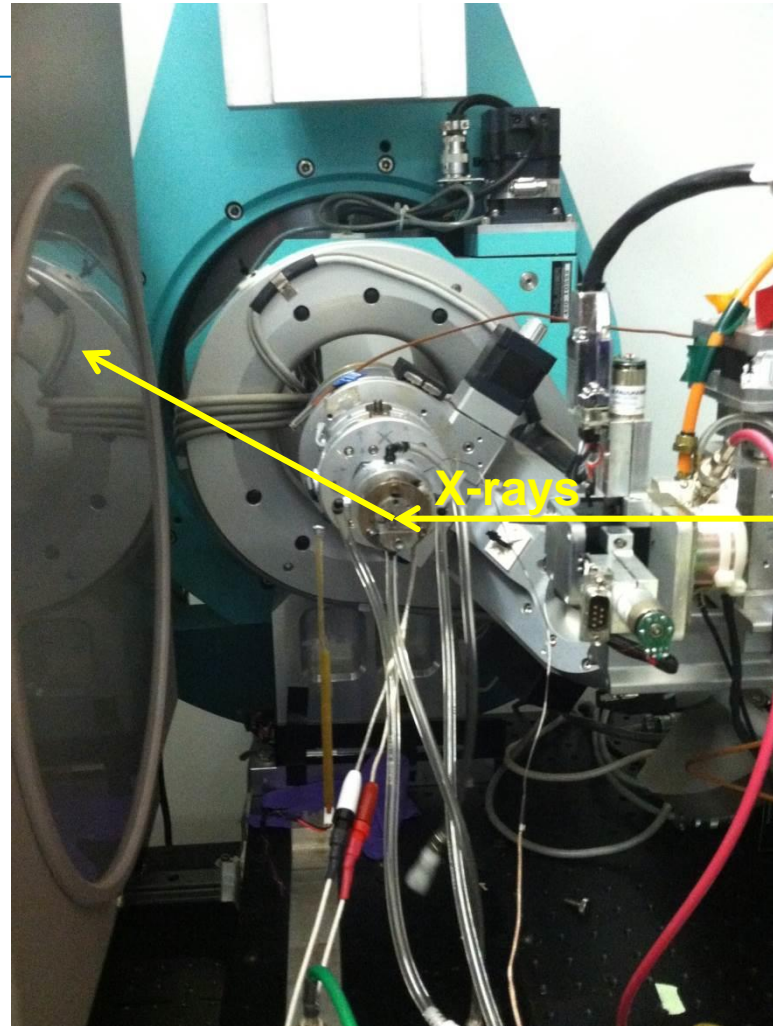
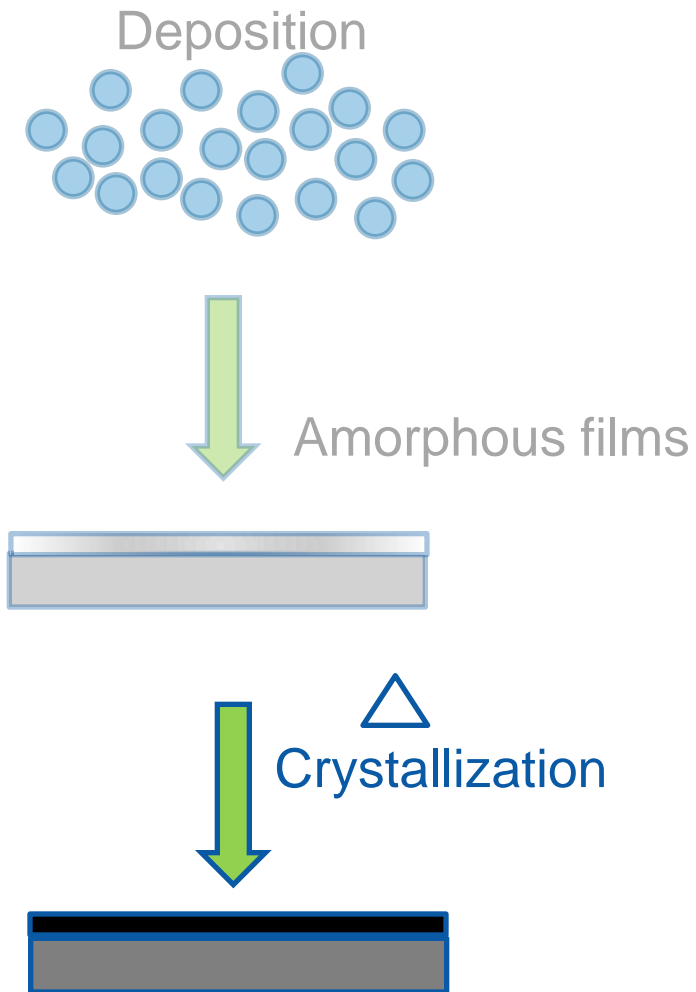
15 Mapping Tools

- Composition (XRF,RBS)
- Structure (XRD, Raman)
- Transport (4pp, Seebeck)
- Optical (UV-Vis, IR,PL)
- Surface (KP, XPS/UPS)
- Microscopy (SEM, AFM)

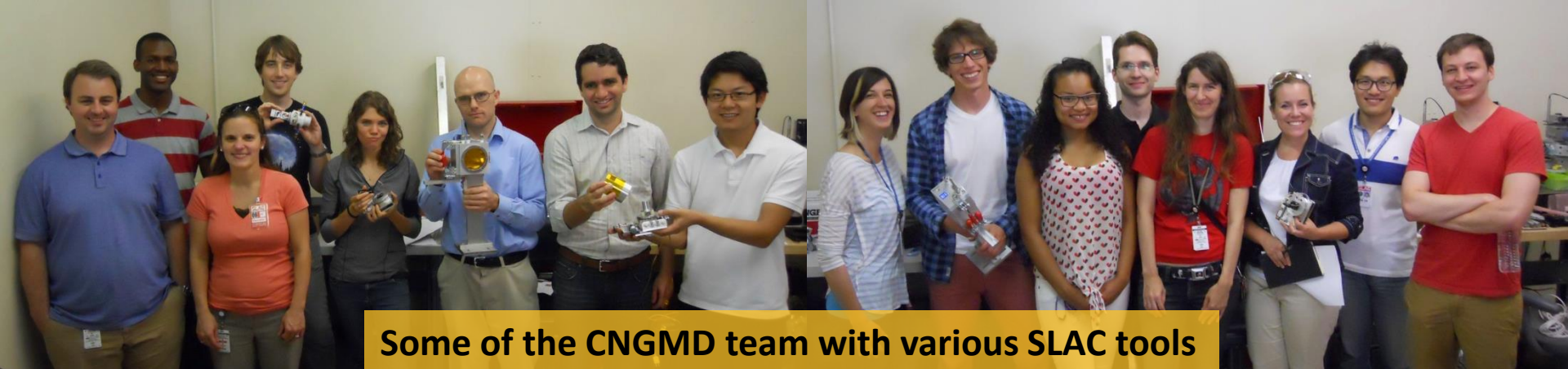
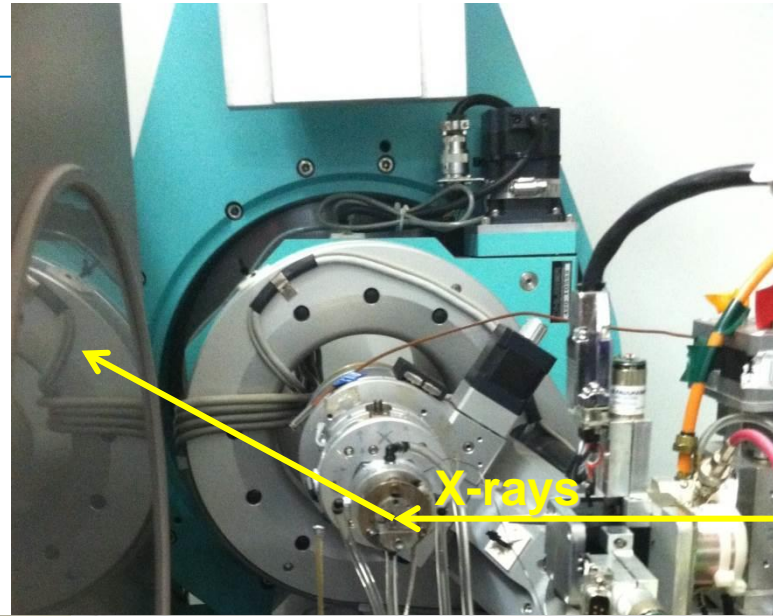
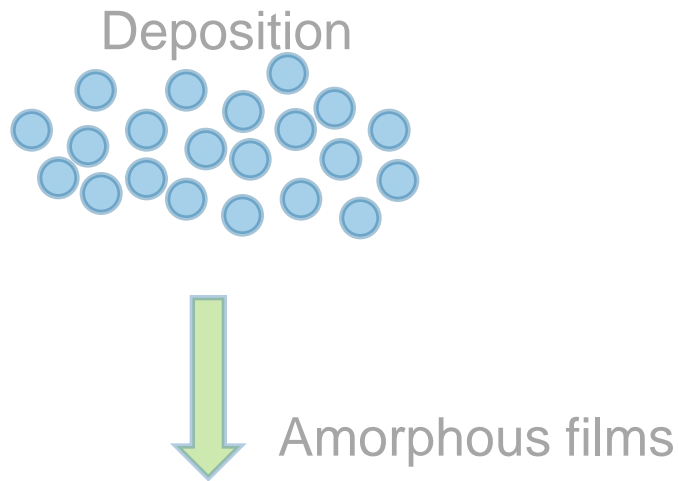
Advanced Data Tools

- NREL Data Network
- Igor PRO framework
- Extensible
- User-assisted analysis
- Data mining/analysis

In-Situ Tools at SLAC: In-situ crystallization of amorphous films



In-Situ Tools at SLAC: In-situ crystallization of amorphous films



Two Main Approaches to Materials by Design

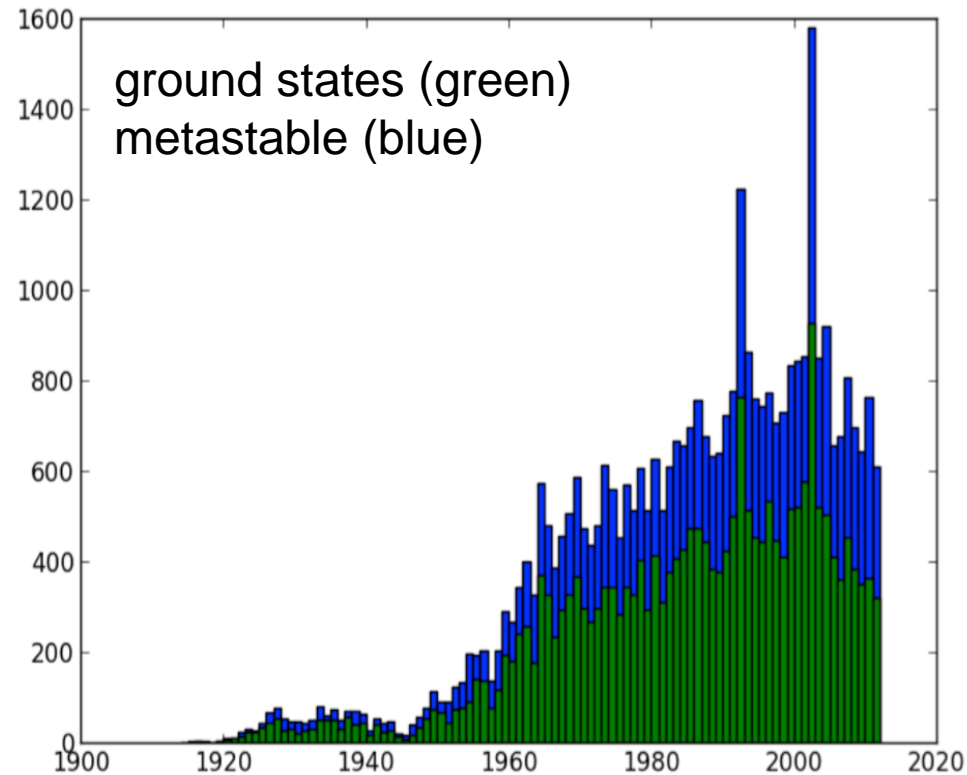
Design by Design Principles

- Many material systems with known structure and composition (e.g. ICSD)
- **Functionality unknown**
- Search via design principles for targeted functionalities

Missing Materials

- Many material systems, but **structure unknown**
- Many (~ 50–100) possible configurations, requiring energy minimization and stability analysis
- Target properties: first existence, then other properties

Metastable Compounds



G. Ceder, and K.A. Persson
C. Wolverton, et al. *Journ of Mater*, 65, 1501 (2013).

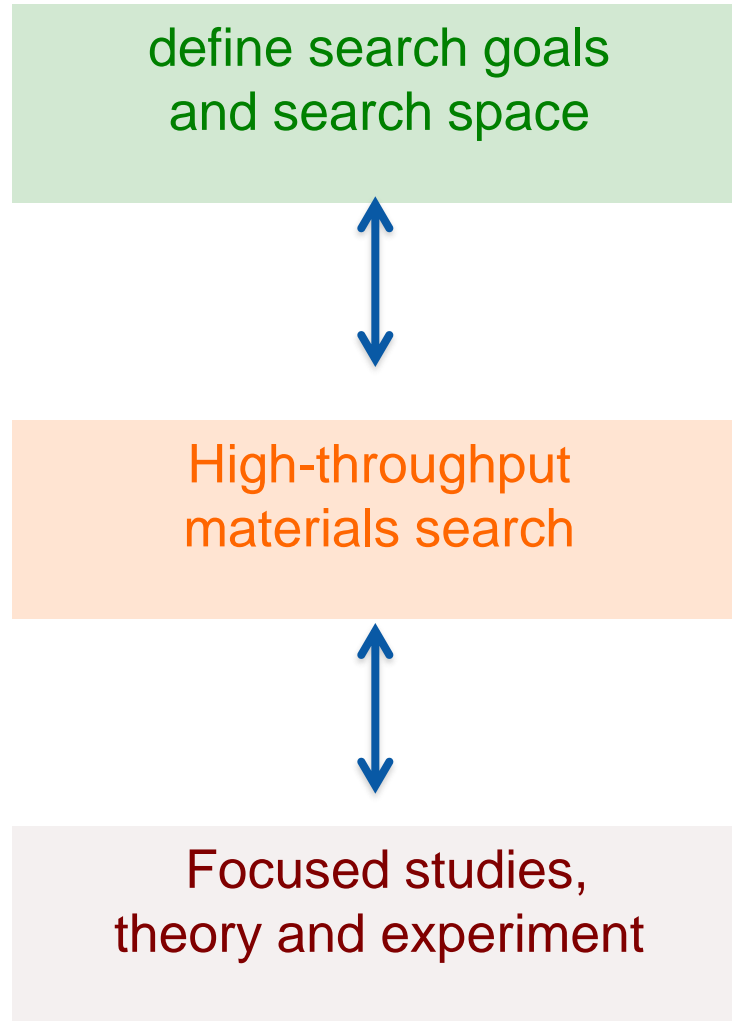
Two Main Approaches to Materials by Design

Design

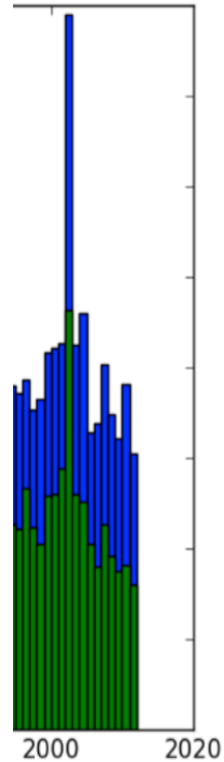
- Material known (e.g. steel)
- **Fun**
- Search target

Mission

- Material structure
- Material composition
- Target properties

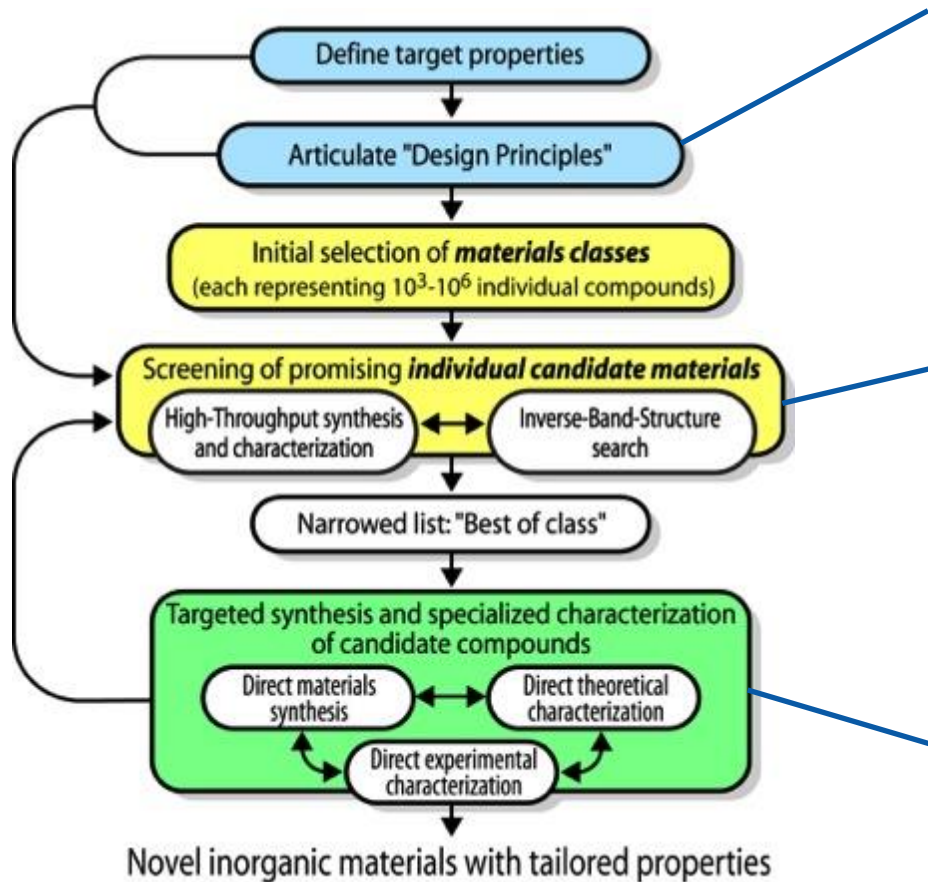


ids



Implementing Inverse Design

Designing p-Type Ternary Oxides

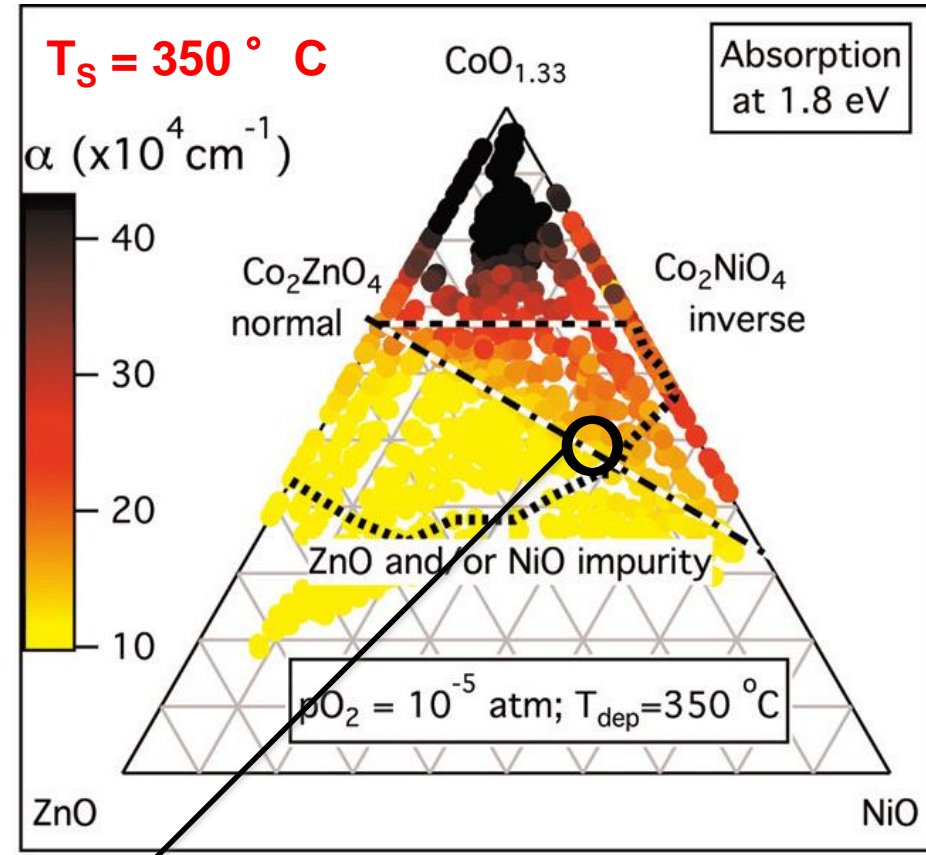
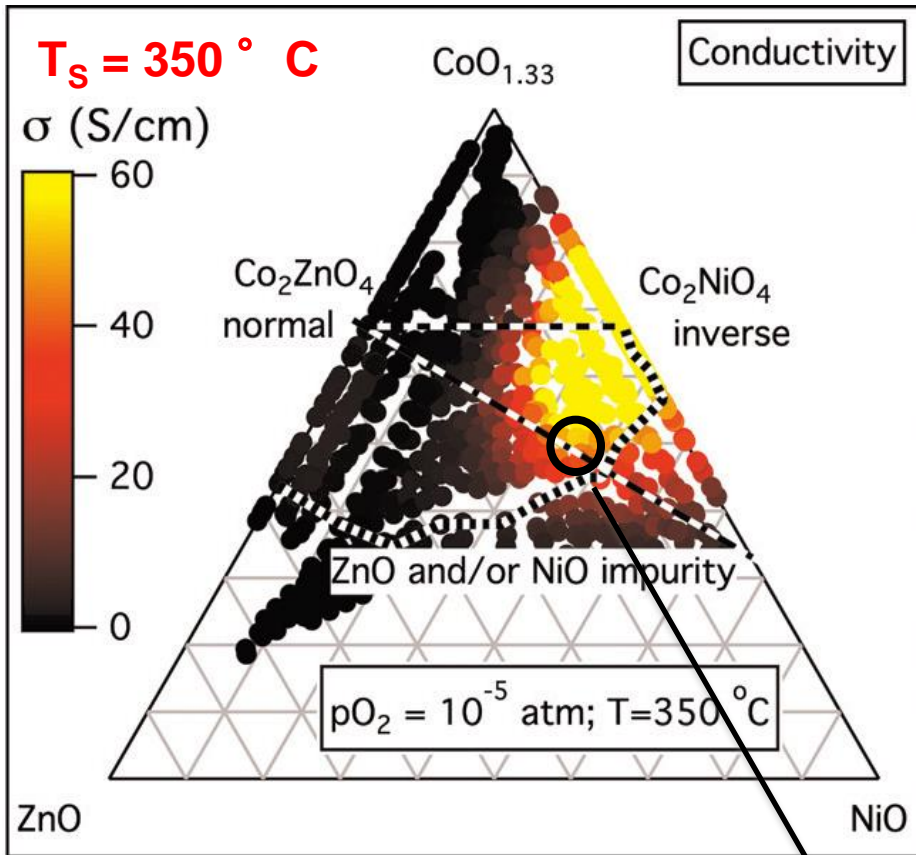


DEVELOP p-type TCO design principles

SEARCH A₂BO₄ w.r.t. design principles

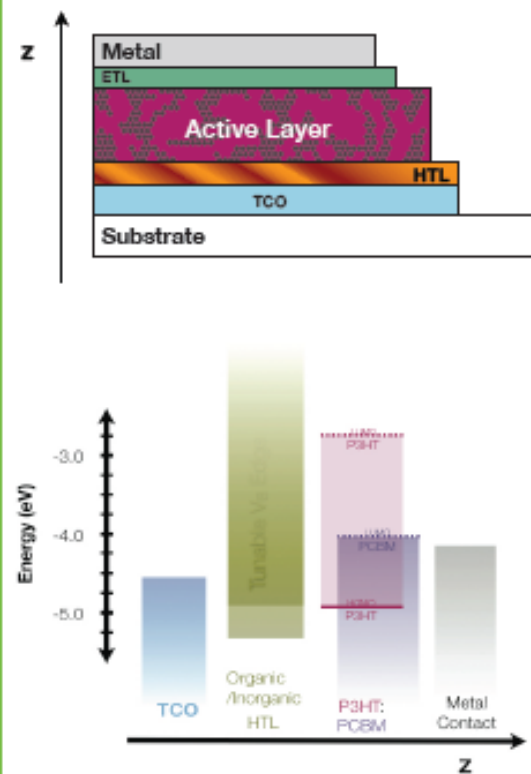
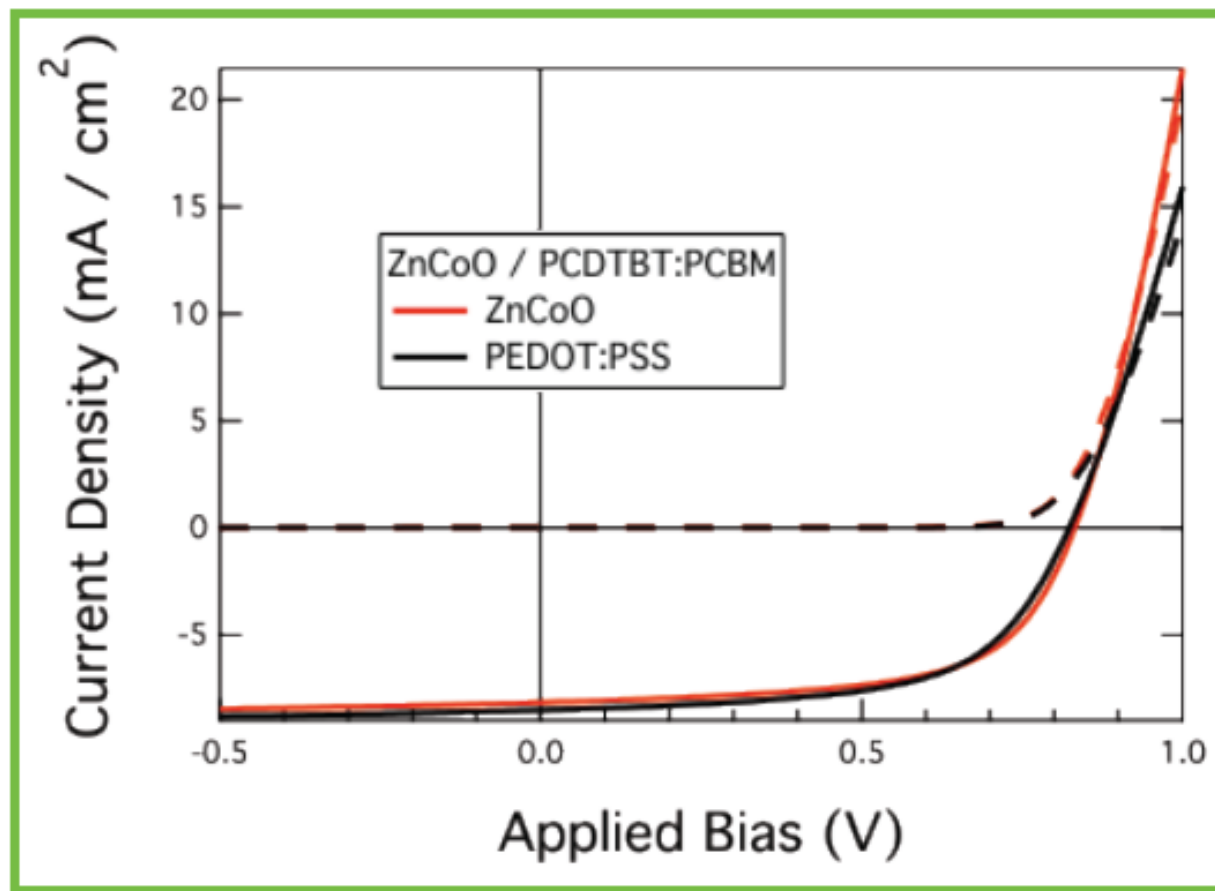
IMPROVE Co₂ZnO₄ based on design principles

Optical and Electrical: Region of Interest



Optical **and** Electrical properties optimized in composition region

Zn-Co-O as a hole transport layer (HTL) for OPV



- Performance comparable to PEDOT:PSS on the first try

"Zn-Ni-Co-O wide-bandgap p-type conductive oxides with high work functions"

A. Zakutayev, J. D. Perkins, P. A. Parilla, N. E. Widjonarko, A. K. Sigdel, J. J. Berry and D. S. Ginley, MRS Communications (2011)



High-throughput Discovery of New A_2BX_4 Compounds

(A,B)

hydrogen 1 H 1.008																	helium 2 He 4.0026				
lithium 3 Li 6.941	beryllium 4 Be 9.0122															boron 5 B 10.81	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg 24.305															aluminum 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.38	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80				
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc 98	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.32	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.6	iodine 53 I 126.90	xenon 54 Xe 131.29				
cesium 55 Cs 132.91	barium 56 Ba 137.33	lanthanum 57-70 * 71 Lu 174.967	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po 209	astatine 85 At 210	radon 86 Rn 222				
francium 87 Fr	radium 88 Ra	actinides 89-102 ** 103 Lr	rutherfordium 104 Rf	dubnium 105 Db	seaborgium 106 Sg	bohrium 107 Bh	hassium 108 Hs	meitnerium 109 Mt	unbinilium 110 Uun	ununtrium 111 Uuu	unquadrium 112 Uub	unseptennium 114 Uuq									

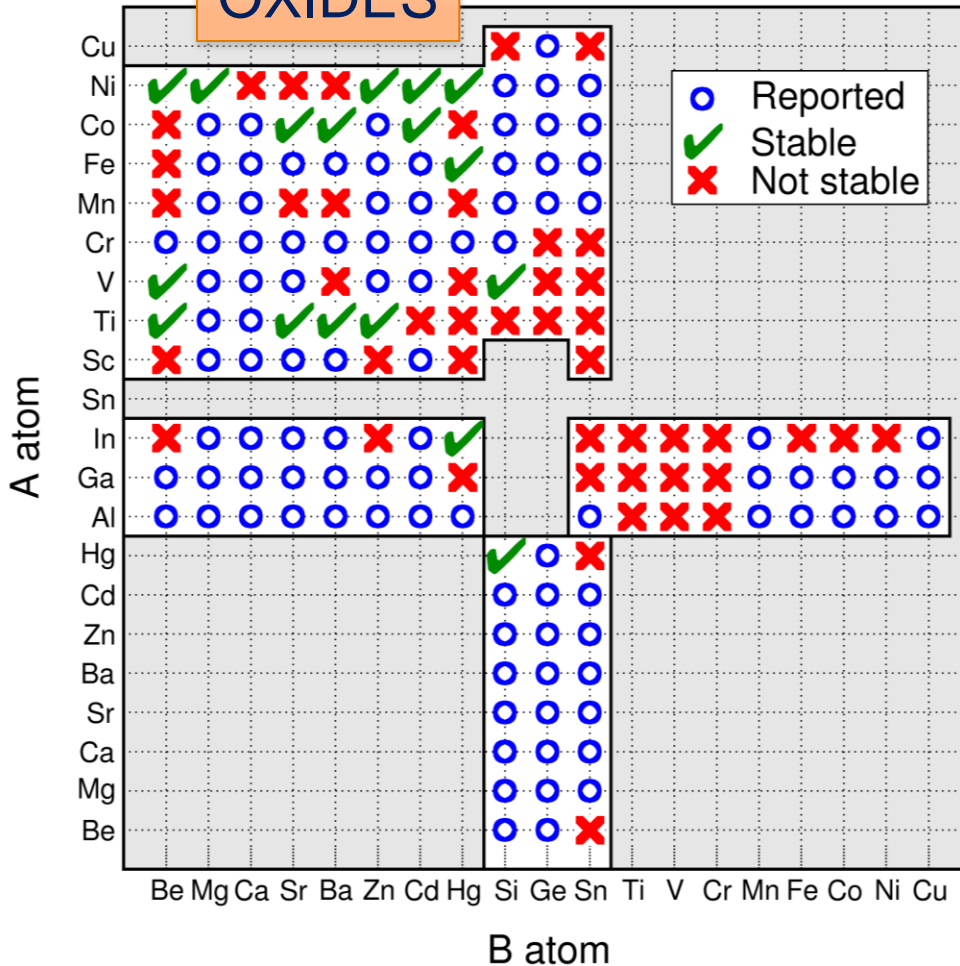
Rules:(1) only one transition metal at a time
(2) respect possible oxidation states

Total 656 possible combinations
250 are reported
406 are not reported (“missing compounds”)



Predicted New A_2BO_4

OXIDES



7 already predicted by
Hautier *et al.*, Chem. Mater., 2010

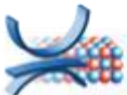
Out of 63 missing oxides
46 not stable
17 stable

Newly predicted:

Hg_2SiO_4 In_2HgO_4
 Ti_2BeO_4 Ti_2SrO_4
 Ti_2BaO_4 Ti_2ZnO_4
 V_2BeO_4 V_2SiO_4

A_2BX_4 search:

~80000 individual total-energy
calculations
(incl. structures and magnetic
configurations)



CID Predicted Ternary Materials

A₂BX₄ materials main group and 3d elements:

Out of 684 variations, 429 are unreported

100 predicted stable, 11 undetermined, and 318 predicted not stable

X. Zhang, V. Stevanovic, M. d'Avezac, S. Lany, and A. Zunger, Phys. Rev. B, 86, 014109 (2012)

ABX materials with 8 electrons:

Out of 714 variations, 488 are unreported

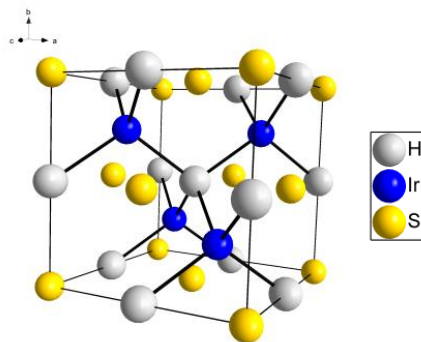
235 predicted stable, 18 undetermined, and 235 predicted not stable

X. Zhang et al., Adv. Funct. Mater. 22, 1425–1435 (2013).

Identification of ABX ternary materials

HfIrSb, ZrRhBi, ScRhTe, TaCoSn, TaIrGe, VIrSi, VRhSi and HfRhP have been shown to crystallize in their predicted crystal structure.

Example:
HfIrSb
F-43m

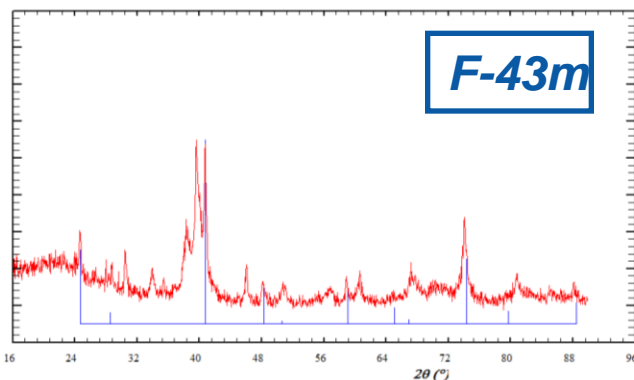


Predicted crystal structure

The symmetry of a predicted stable compound makes possible:

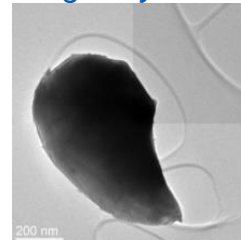
1) Simulation of diffraction pattern

2) Fast identification in the experimental pattern

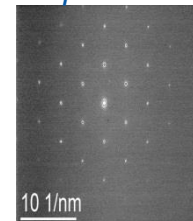


With Confirmation By Electron diffraction

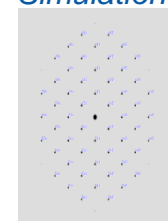
Single crystallite



Experiment



Simulation



(110) zone

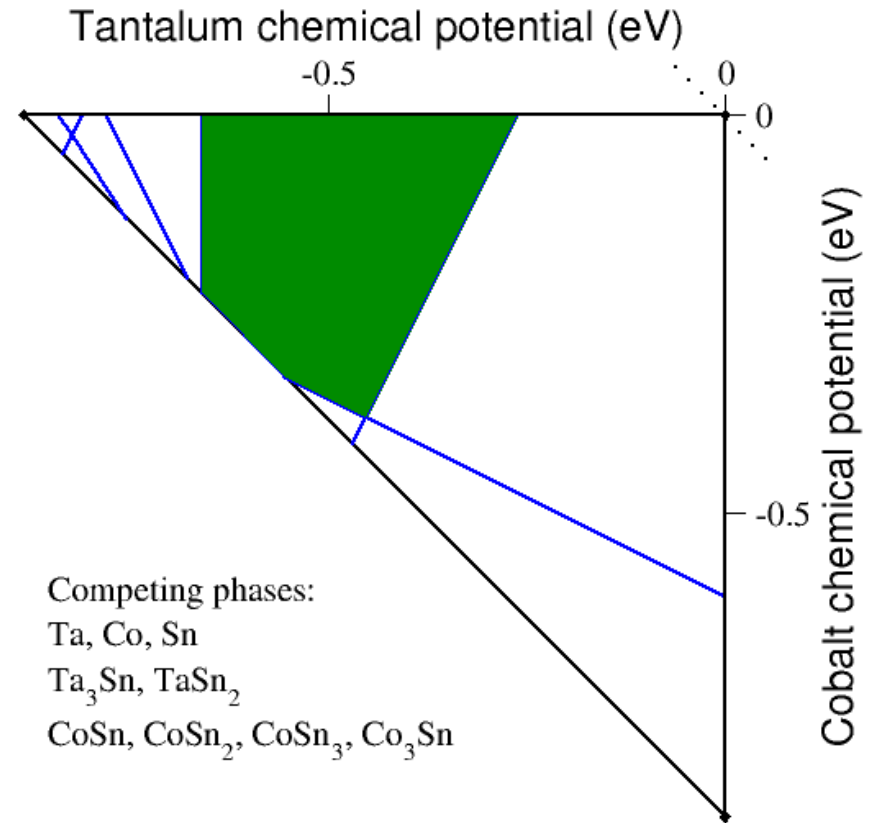
Fast identification in multiphasic sample

Missing TaCoSn Compound

Not known in ICSD or ICDD

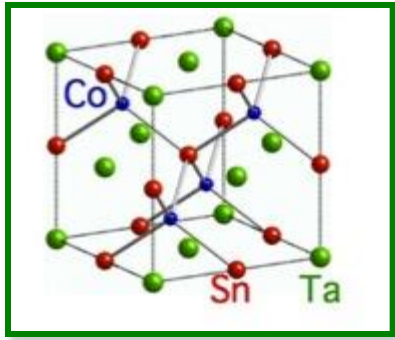
Large stability range

Predicted to have semi-conducting gap ~ 1.3 eV
(GGA + U)

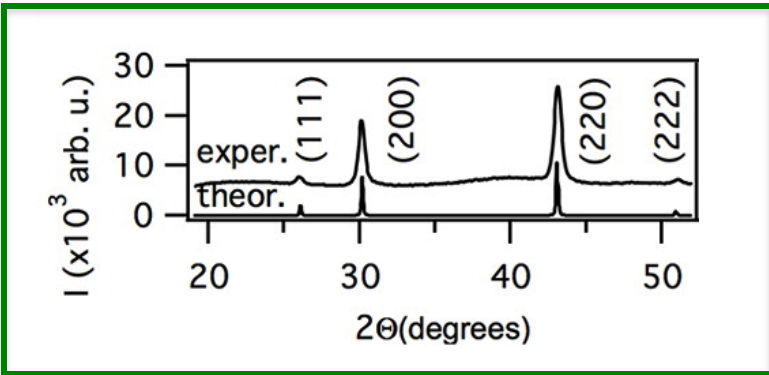


Validation: growth of new TaCoSn

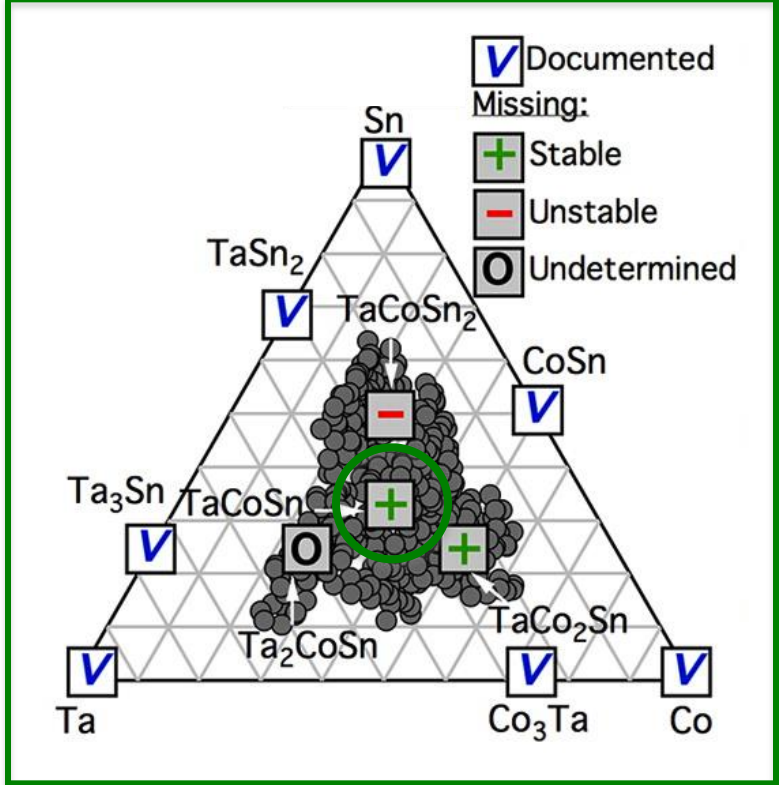
Predicted Structure



XRD: Predicted & Measured



TaCoSn Grown

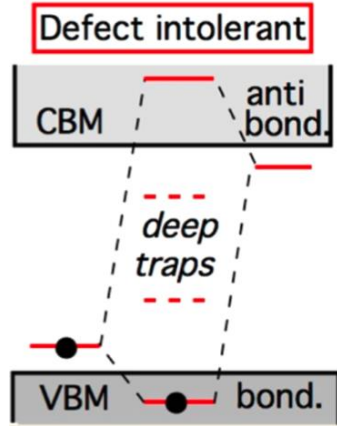


Zakutayev et al. *J. Am. Chem. Soc.*, 2013, 135, 10048

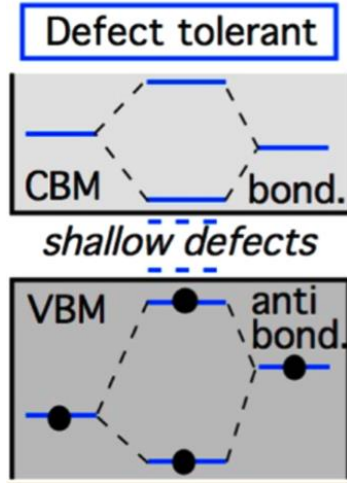
Perovskite Search: Proxy for Transport/Defect Tolerance

- Minority carrier lifetimes challenging for both computation and experiment
- The concept of defect tolerance can be used as a proxy (qualitative)
- Defect tolerance is a consequence of the electronic structure

Classic III-Vs and II-VIs are defect intolerant: GaAs, InP, GaN, ZnO,...



Electronic structure of a defect tolerant material

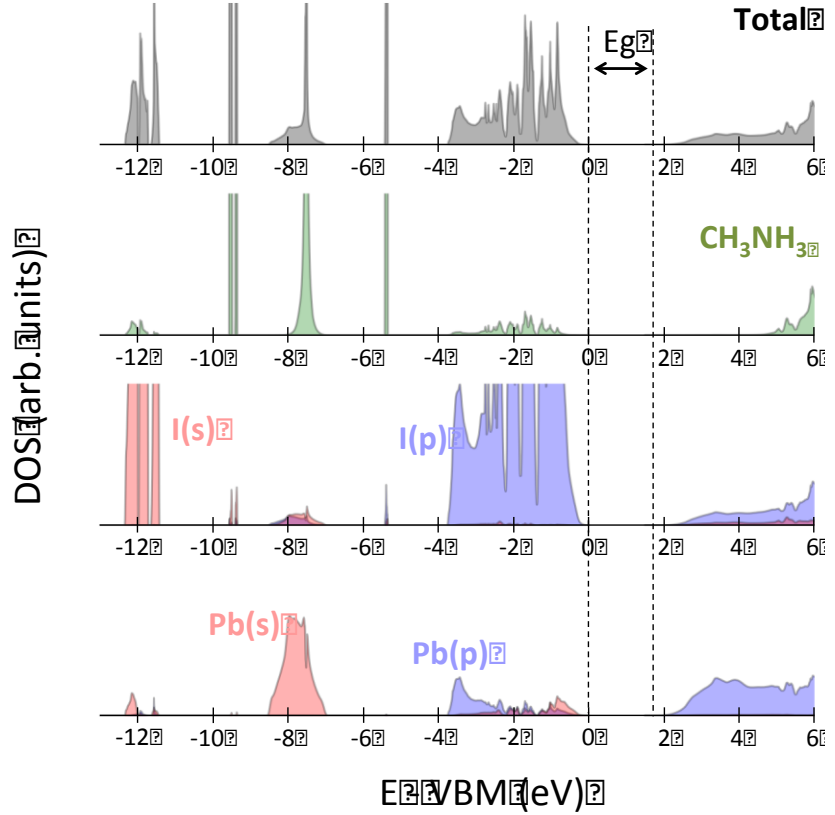
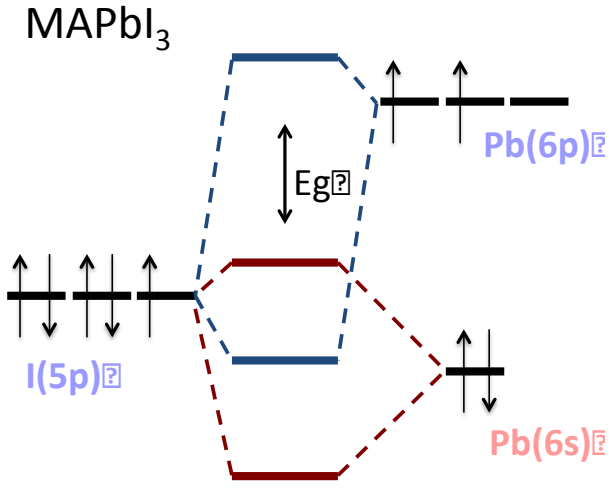


Zakutayev et al. *J. Phys. Chem. Lett.* 5 (2014)
 Brandt et al. *MRS Communications* 5, 265–275 (2015)
 S.B. Zhang, et al., *Phys. Rev. B* 57, 9642 (1998)

Electronic structure of MAPbI₃

In MAPbI₃:

- Pb 6s orbitals provide antibonding character to the VBM (s-p repulsion)
- With spin-orbit coupling, conduction band is more disperse

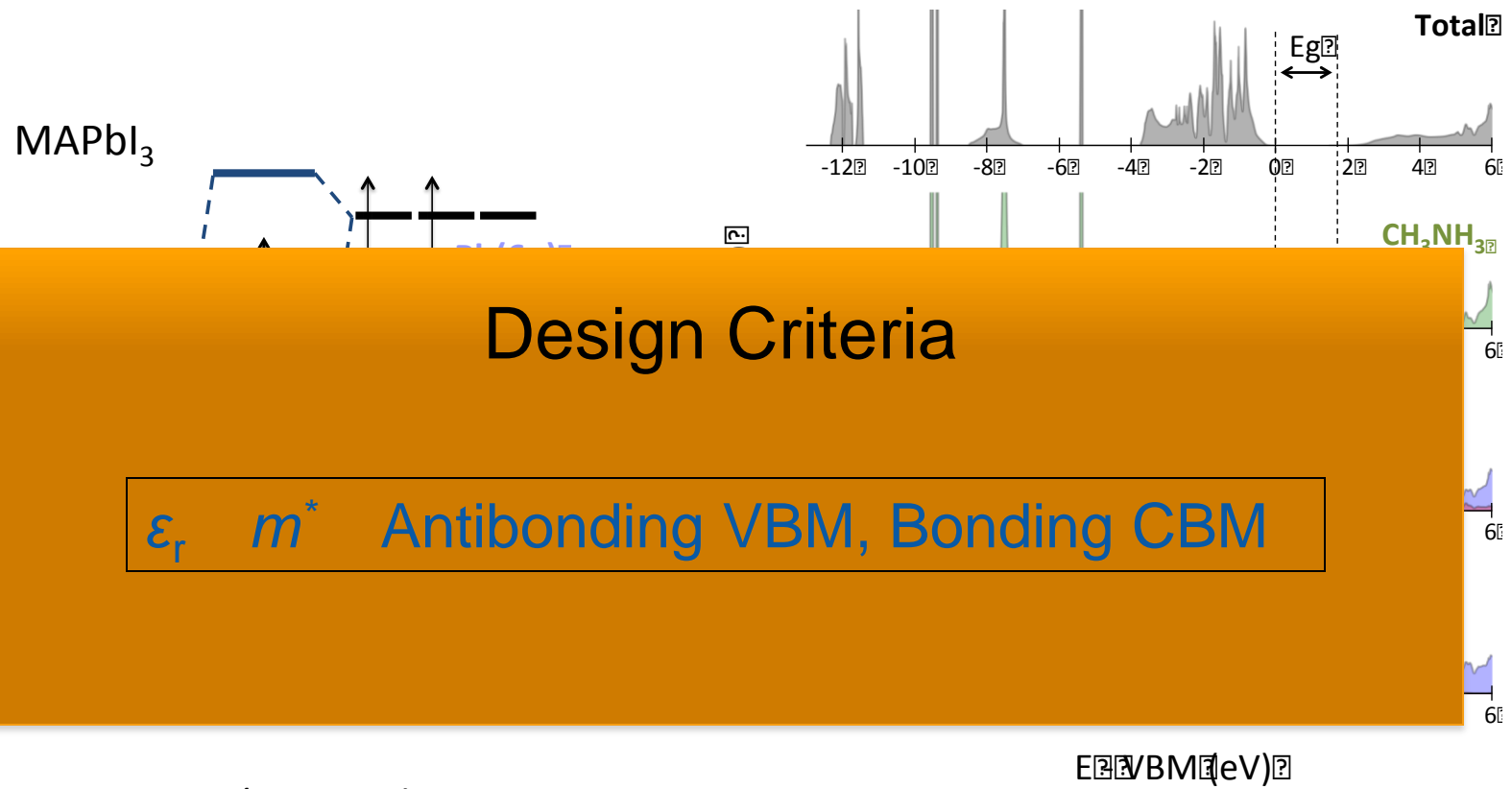


R.E. Brandt, V. Stevanović, D.S. Ginley, T. Buonassisi,
MRS Communications **5**, 265–275 (2015)

Electronic structure of MAPbI₃

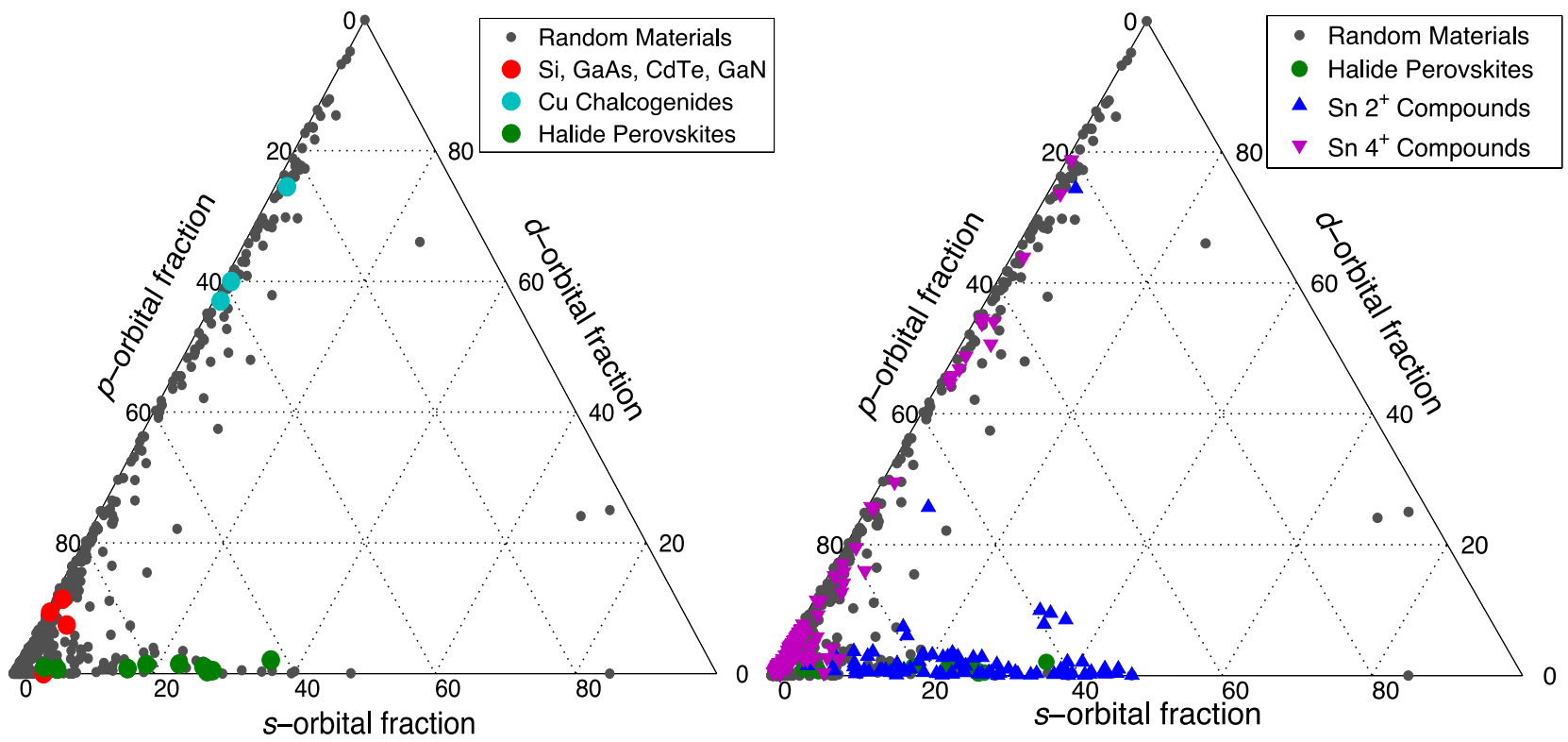
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- With spin-orbit coupling, conduction band is more disperse



R.E. Brandt, V. Stevanović, D.S. Ginley, T. Buonassisi,
MRS Communications **5**, 265–275 (2015)

Search 27,000 Inorganic Materials for s-VBM



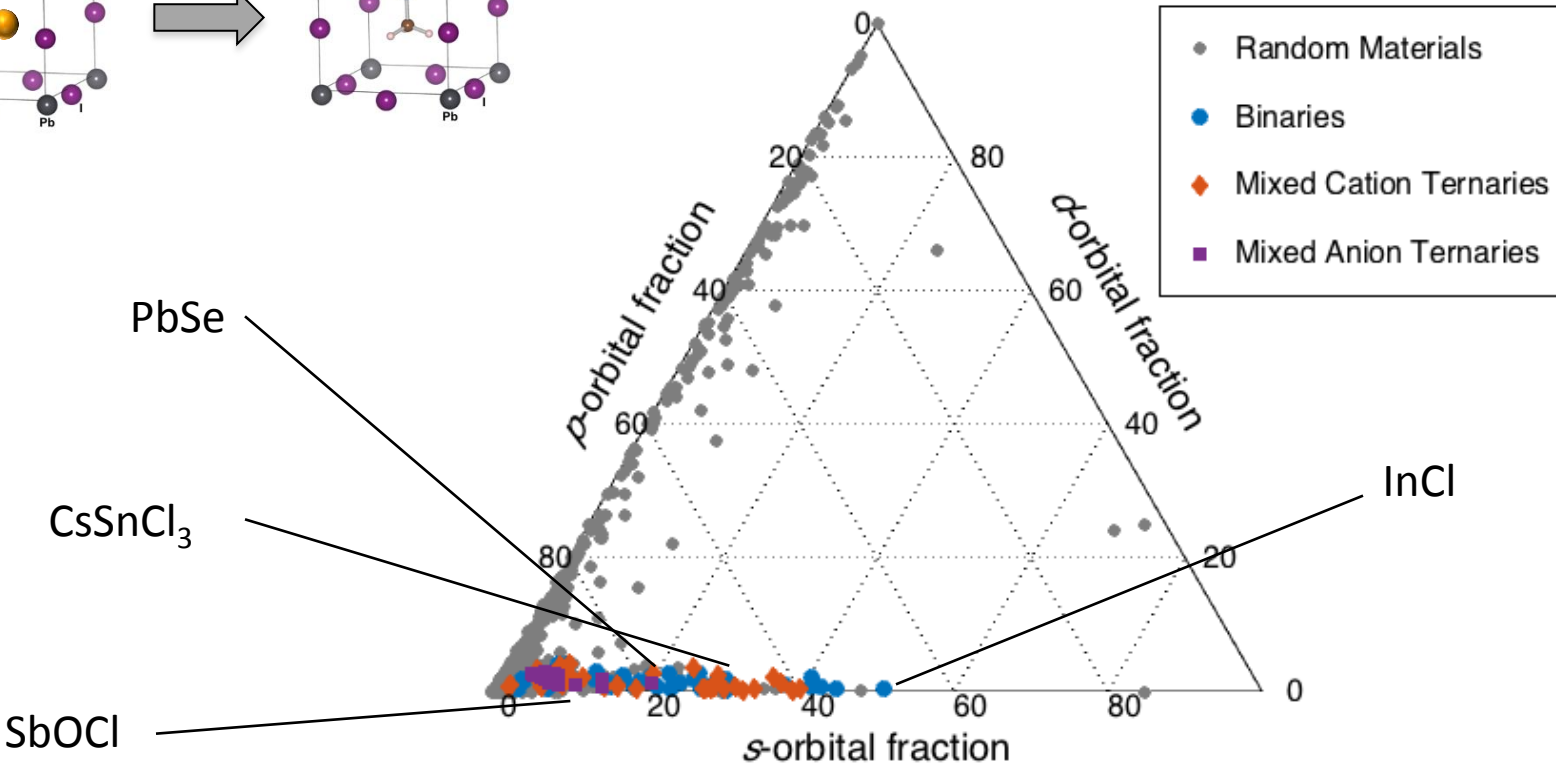
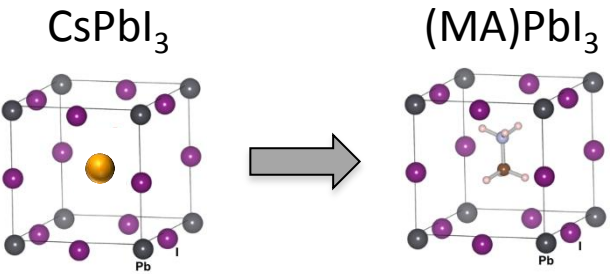
R.E. Brandt, V. Stevanović, D.S. Ginley, T. Buonassisi, *MRS Comm.* **5**, 265 (2015)

A. Jain, S.P. Ong, G. Hautier, *et al.* *APL Materials* **1**, 011002 (2013)

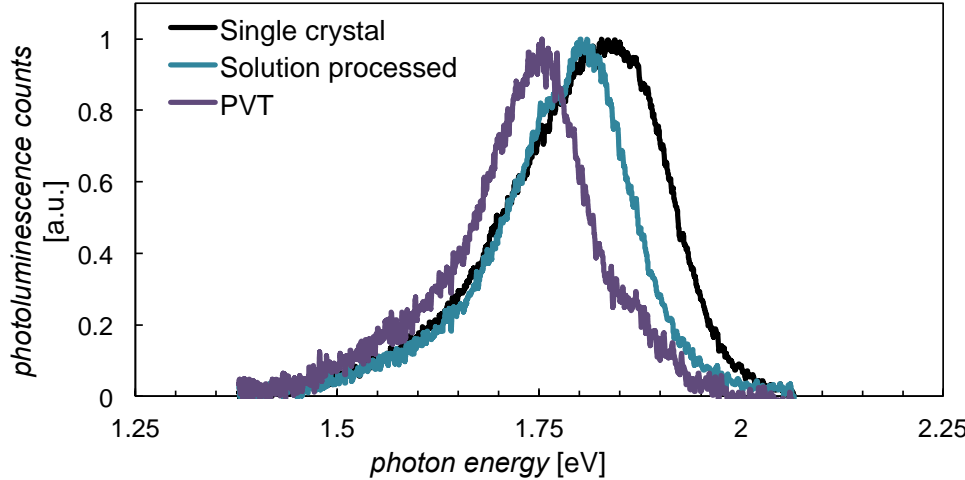
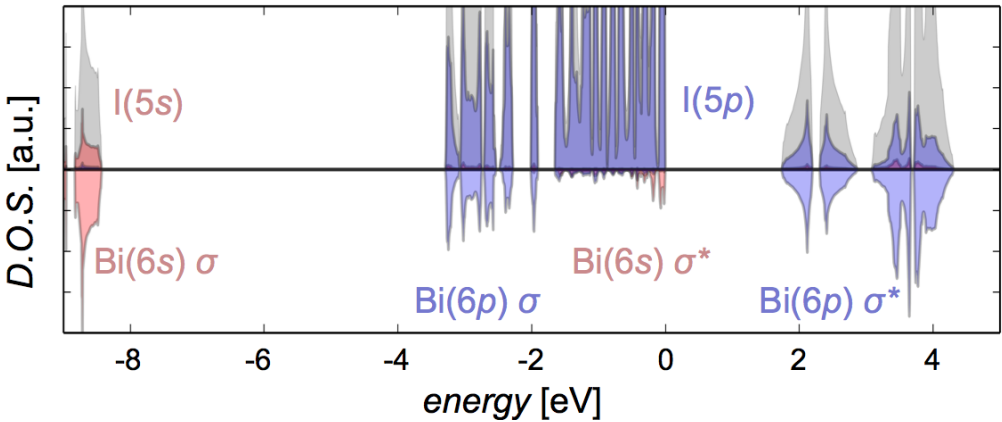
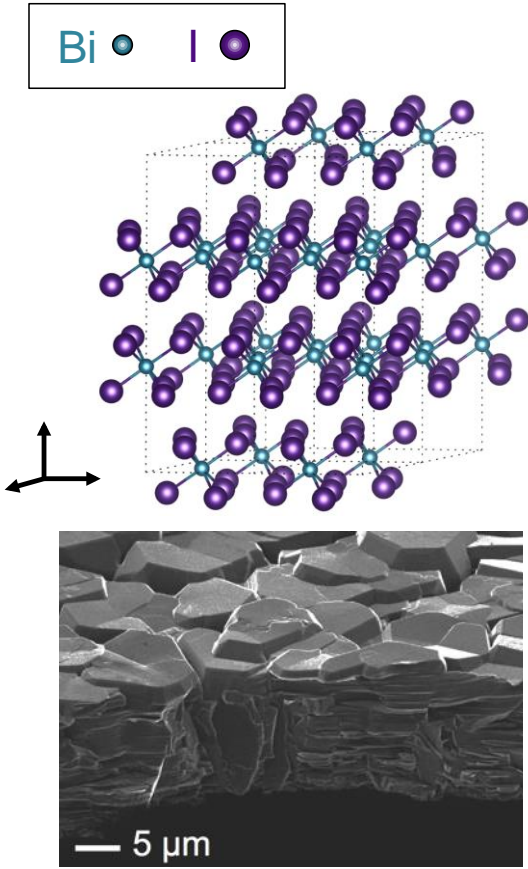
www.materialsproject.org

Multiple Material Classes Identified

- Compounds with “lone-pair” cations: **In⁺**, **Sn²⁺**, **Sb³⁺**, **Tl⁺**, **Pb²⁺**, **Bi³⁺**
- Building libraries of hybrid materials through inorganic analogues



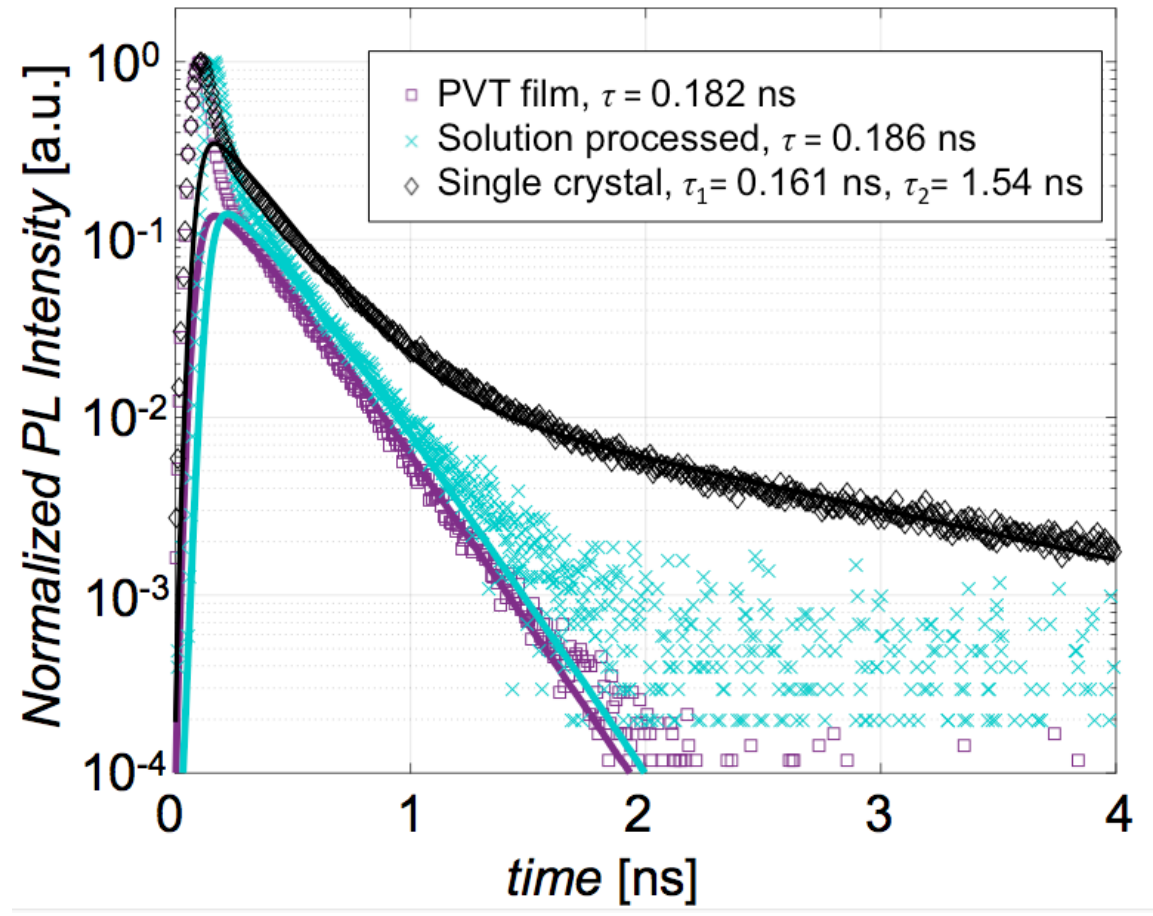
Bismuth Triiodide (BiI₃)



- First films synthesized exhibited room-temperature photoluminescence

R.E. Brandt *et al.*, *J. Phys. Chem. Lett.* **6**, 4297 (2015).

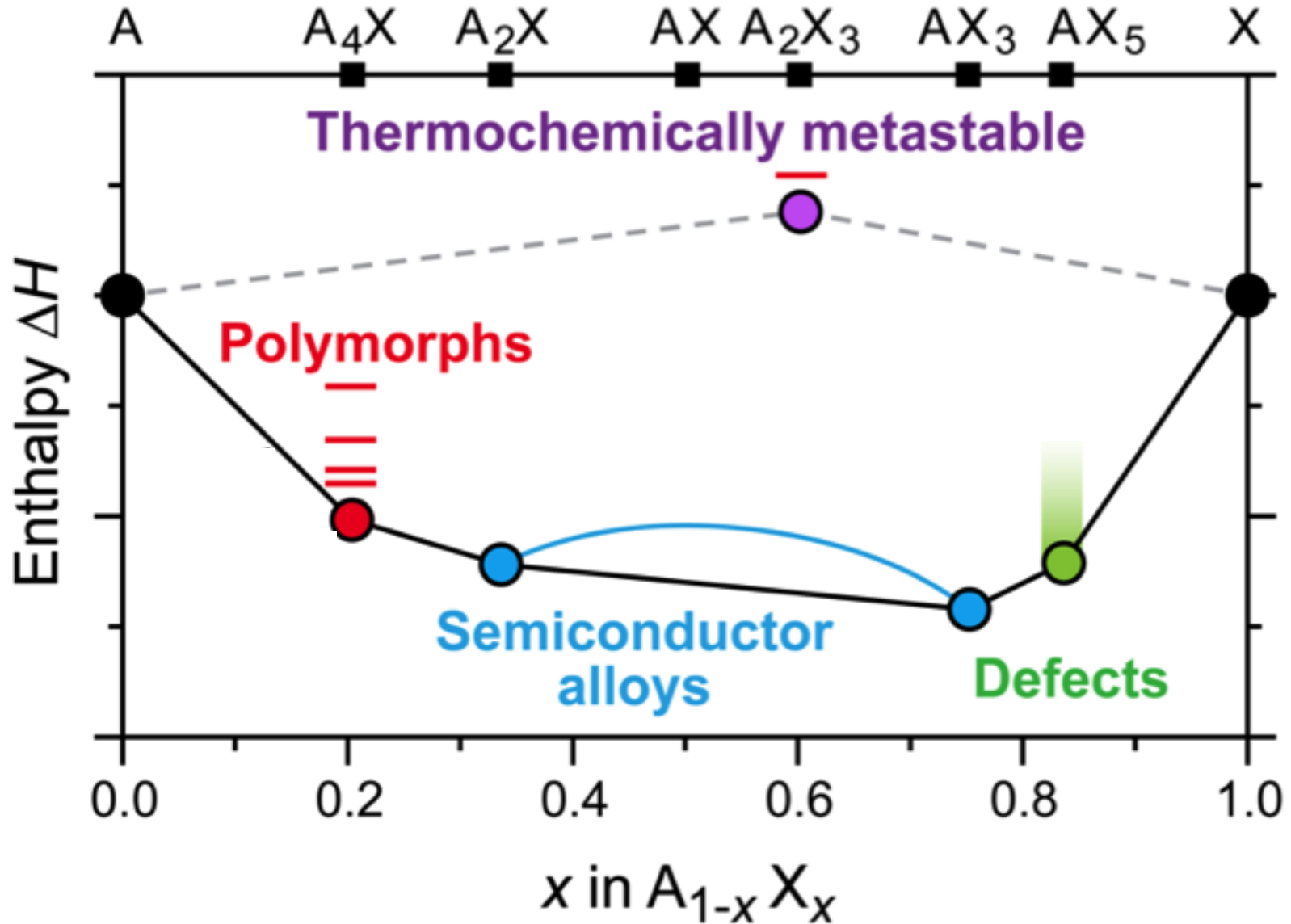
BiI_3 – Carrier Lifetime Measurements



- Informed new design criterion – purity of materials and growth environments

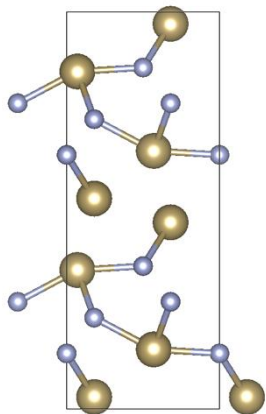
R.E. Brandt *et al.*, *J. Phys. Chem. Lett.* **6**, 4297 (2015).

Incorporating Metastability



Search for New Nitrides

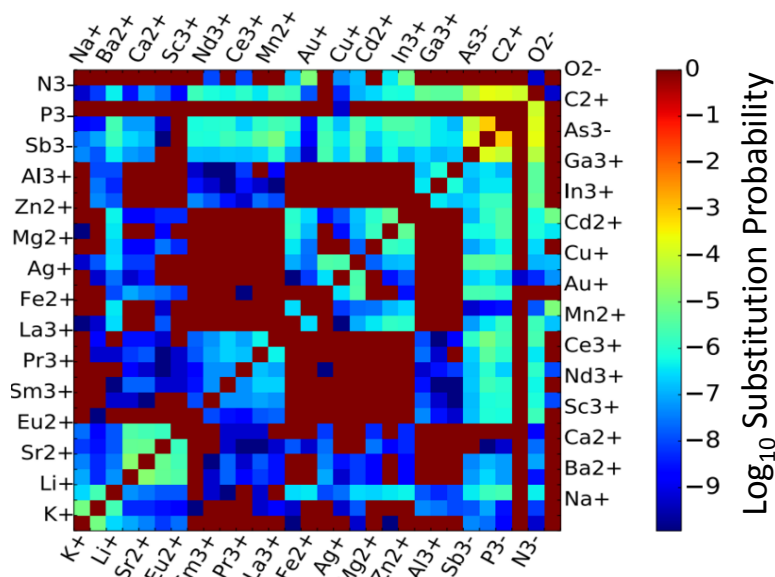
Known Compounds



OsN

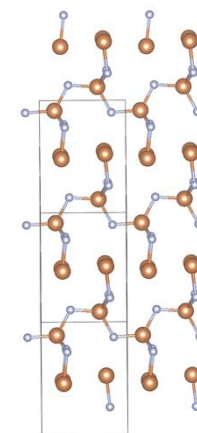
ICSD #167514

Data-Mined Ionic Substitution



Hautier, G, et al. *Inorganic Chemistry* (2010)

Suggested Compounds



SbN

Predicted Structure

Train data-mining algorithm on known Oxides+Pnictides

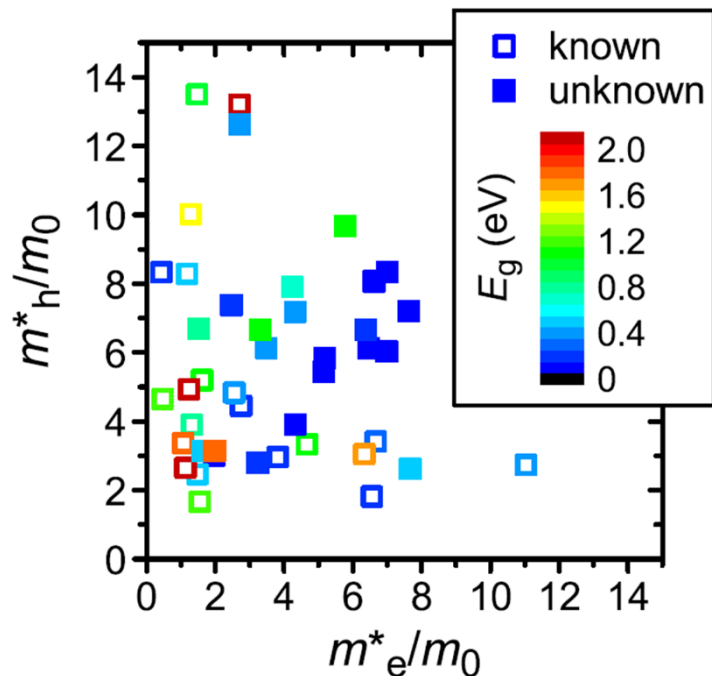
Cross-validated by “predicting” known nitrides. **~80%** chance of recovery

492 suggested binary nitrides
(Alkali, Transition, Main Group)

Candidate Stable Phases

Co₂N, CoN, Cr₃N₂, Cr₃N₄, CrN, Hf₃N₄, Nb₂N, SbN, Sr₂N, Ta₂N, TeN₂, V₂N, V₃N₂, VN, Zn₃N₂

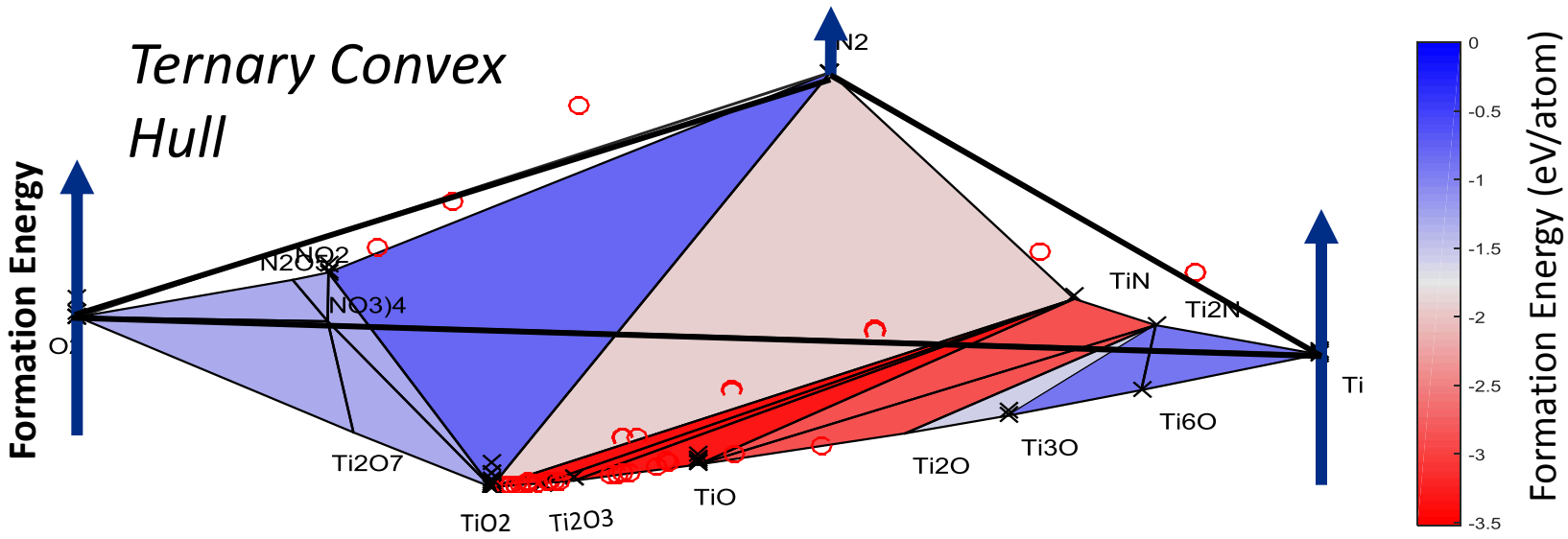
Pathway to new nitrides – from search to application



From design & discovery to properties search

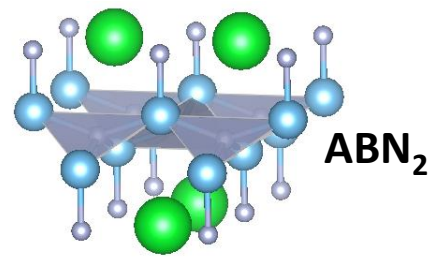
- We calculate **optoelectronic properties** of newly predicted nitrides.
- Promising semiconductor nitrides *accessible via sputtering* will be further screened using higher levels of theory (G_0W_0)
- **Down-selection** for experimental synthesis and characterization

In progress: Ternary Nitride Search

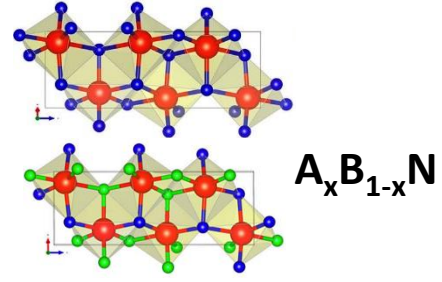


4000 potential ternary nitrides

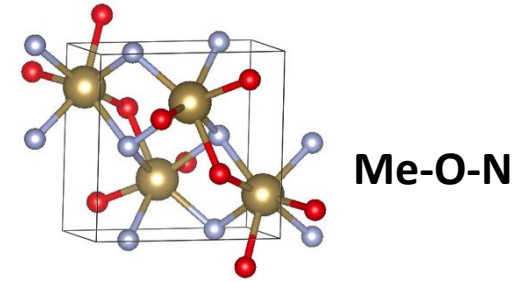
Ternary Nitride Phases



Alloyed Binary Nitrides

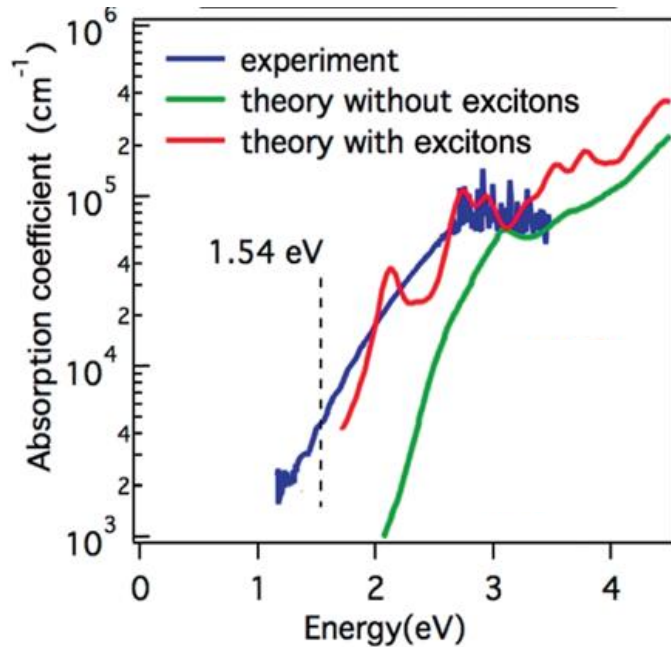


Metal Oxynitrides



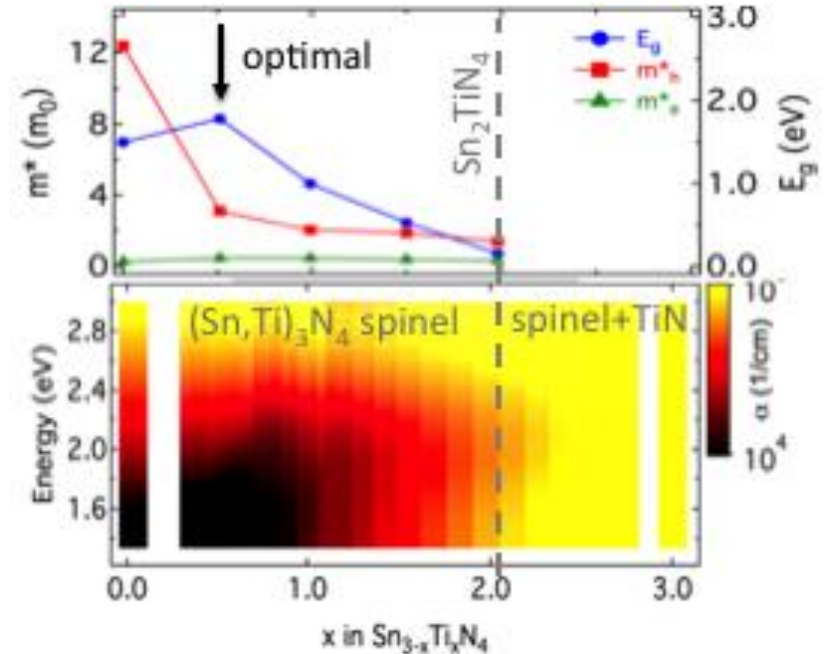
Enormous exploration and design space for new nitrides!

Entry point material: metastable Sn_3N_4



- Sn_3N_4 is a potential PEC material:
 - good optical absorption ($E_g=1.6$ eV)
 - suitable VB position for H_2O oxidation
 - n-type conductivity (light electrons)
- **However:** large hole effective masses

Improved material: metastable $\text{Sn}_{3-x}\text{Ti}_x\text{N}_4$



- $\text{Sn}_{3-x}\text{Ti}_x\text{N}_4$ has better properties than Sn_3N_4
 - Theory: lighter m_h , same m_e , and lower E_g
 - Experiment: strong optical absorption
 - Also: SnTi_2N_4 is a new spinel nitride!!!
 - Structure: pure by XRD, spinodal by TEM

C. Caskey et al, *J. Mater. Chem. C*, 3, 1389(2015)

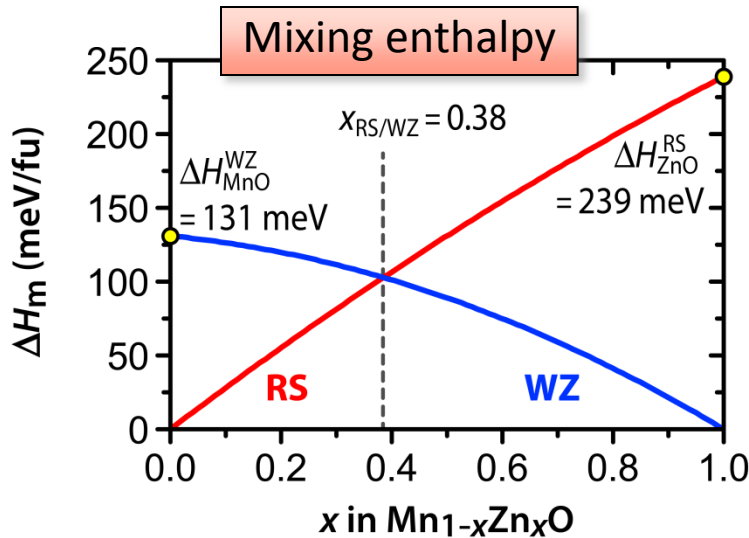
Realization of tetrahedral MnO by alloying with ZnO

ZnO has tetrahedral wurtzite structure similar to zinc-blende

Mixing enthalpy

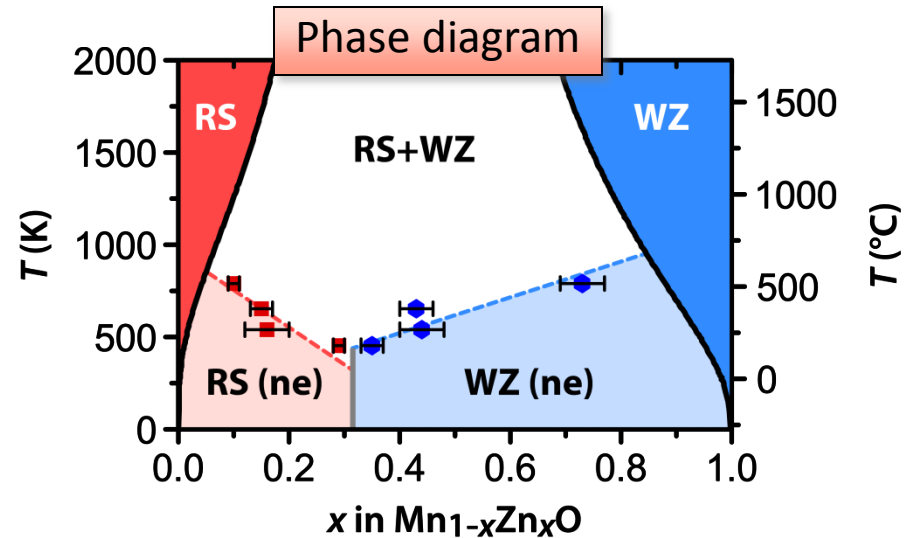
$$\Delta H_{mix}(x) = (1-x) \cdot H_{MnO} + x \cdot H_{ZnO} + \beta \cdot x(1-x)$$

• $\beta = 46$ meV (RS), $\beta = 94$ meV (WZ)



T-x phase diagram

- Common tangent construction
- Ideal solution model for entropy



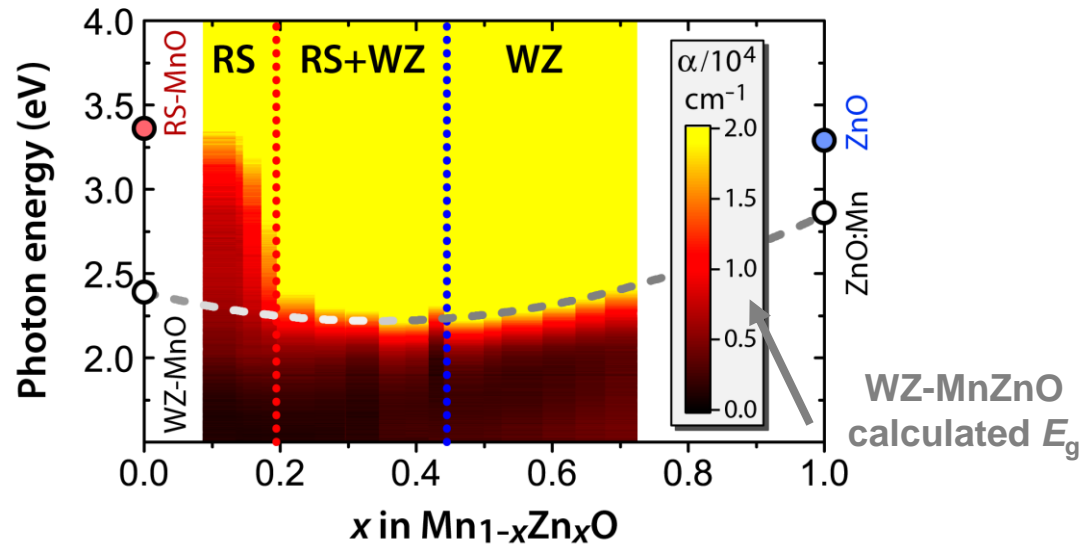
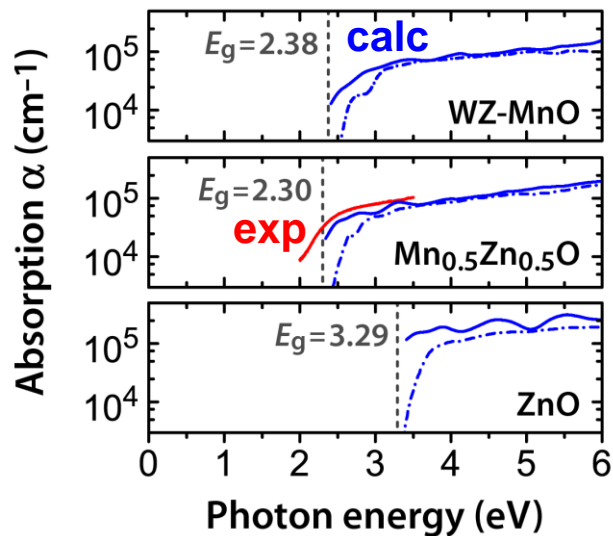
Theory

Phase transition predicted at $x = 0.38$
 However, desired alloy composition is deep inside miscibility gap

Experiment

Realization of single-phase WZ MnZnO by non-equilibrium PLD growth
 Predicted phase transition confirmed

Optical absorption: Theory vs experiment



Measured absorption coefficient α (contour plot) and calculated band gaps (dashed line)

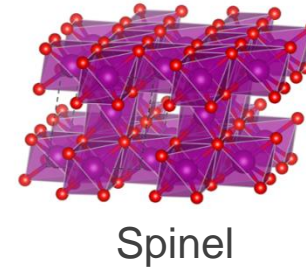
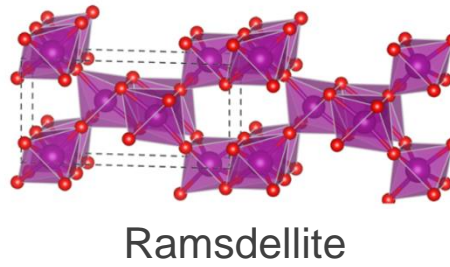
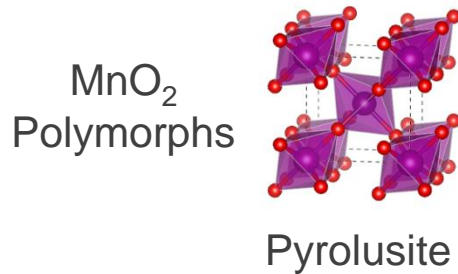
Metastable transition metal oxide alloy with unique semiconducting properties

- Band gap control through alloying
- Non-equilibrium growth via PLD
- PEC measurements
 - Band alignment, carrier transport

MOx Polymorphs

Manganese Oxides

- Over 30 known polymorphs
- Energy storage, catalysis, pigments

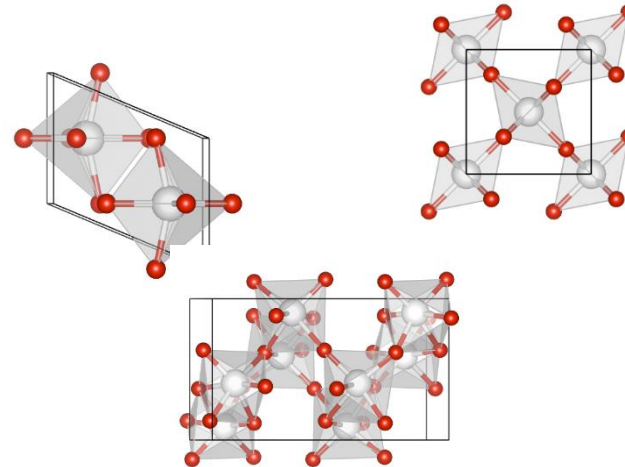


Vanadium Oxides

- Highly complex phase diagram
- Batteries, reagents, coatings

Titanium Oxides

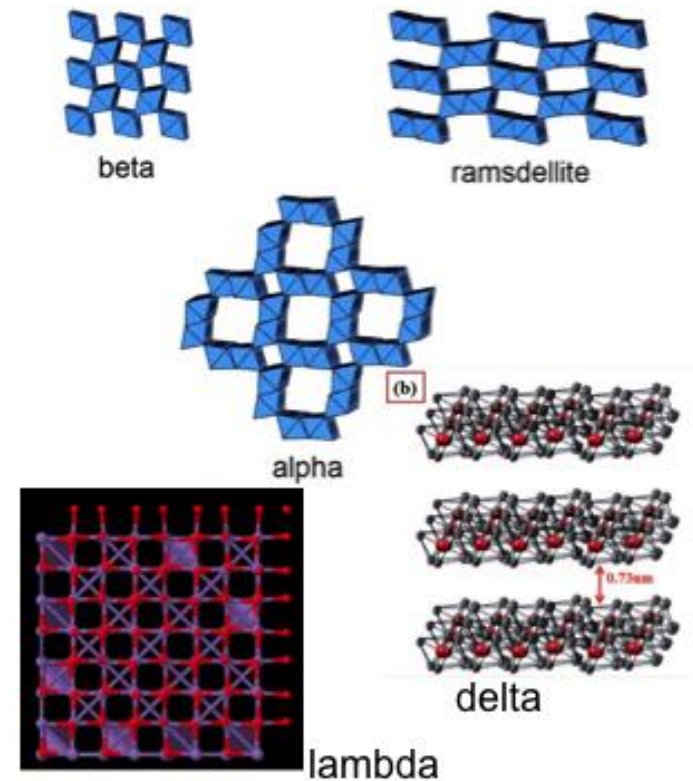
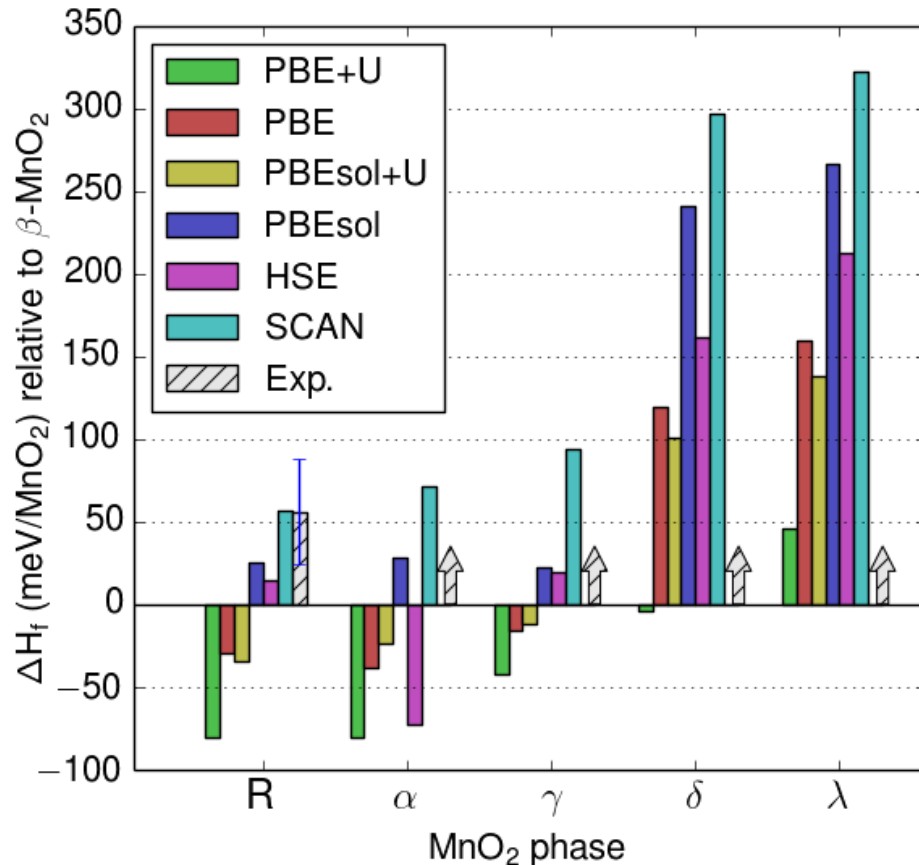
- Poorly understood nanoscale transformations, polymorphs
- Widely studied photocatalyst



Challenging Energetics for MnO₂

SCAN: new meta GGA
(Perdew et al.)

- All known DFT methods fail to reproduce basic MnO₂ properties
- First functional to simultaneously reproduce strong covalent-like bonds and weaker long-range interactions
- Formation energies of MnO₂ polymorphs are reproduced in SCAN

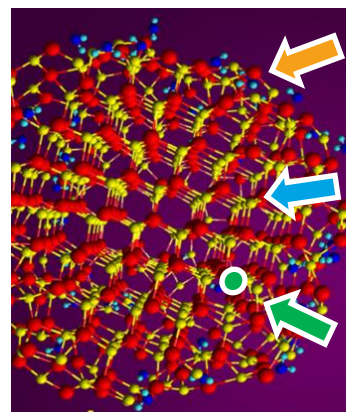


Kitchaev et al. *Phys. Rev B.* **93**, 045132 (2016)

Understanding of synthesis paths is required to rationally and effectively design metastable compounds

Controlled synthesis

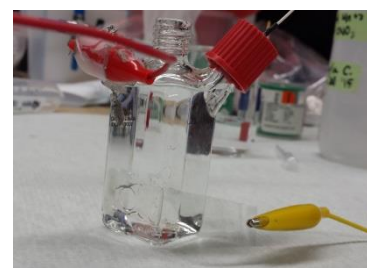
Nucleation



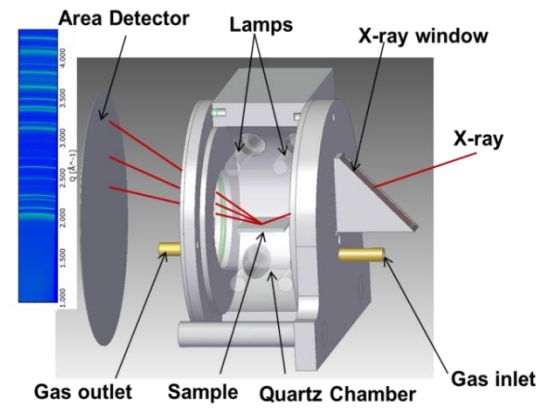
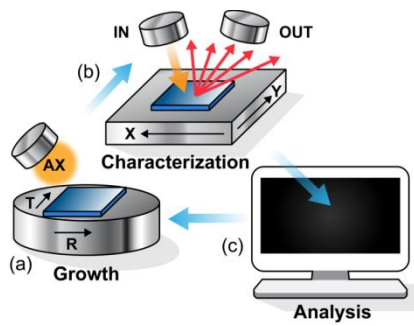
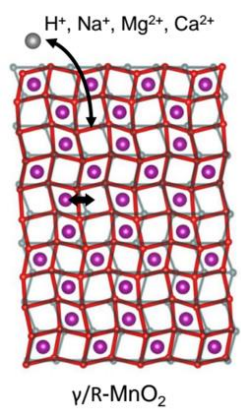
Surface Energy

Bulk Energy

Defect Energy



Bulk solid



In-situ experiments

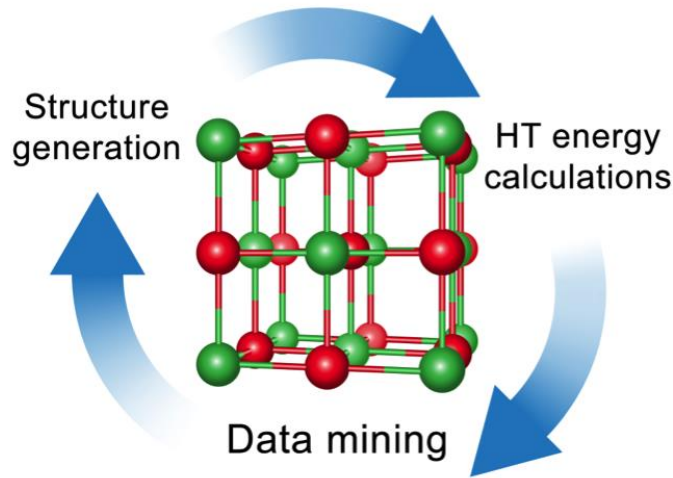
Apply the same fundamental science used to understand properties to understand synthesis

Polymorph selection

Need to understand thermodynamics in all relevant environments

1. Bulk energy of polymorphs
2. Intercalation of ions from solution
3. Surface energies in solution
Predicting nucleation behavior
4. Electrochemical transformations
5. Modeling solid-solid transformations
Kinetics of polymorph conversion
6. Substrate-controlled depositions

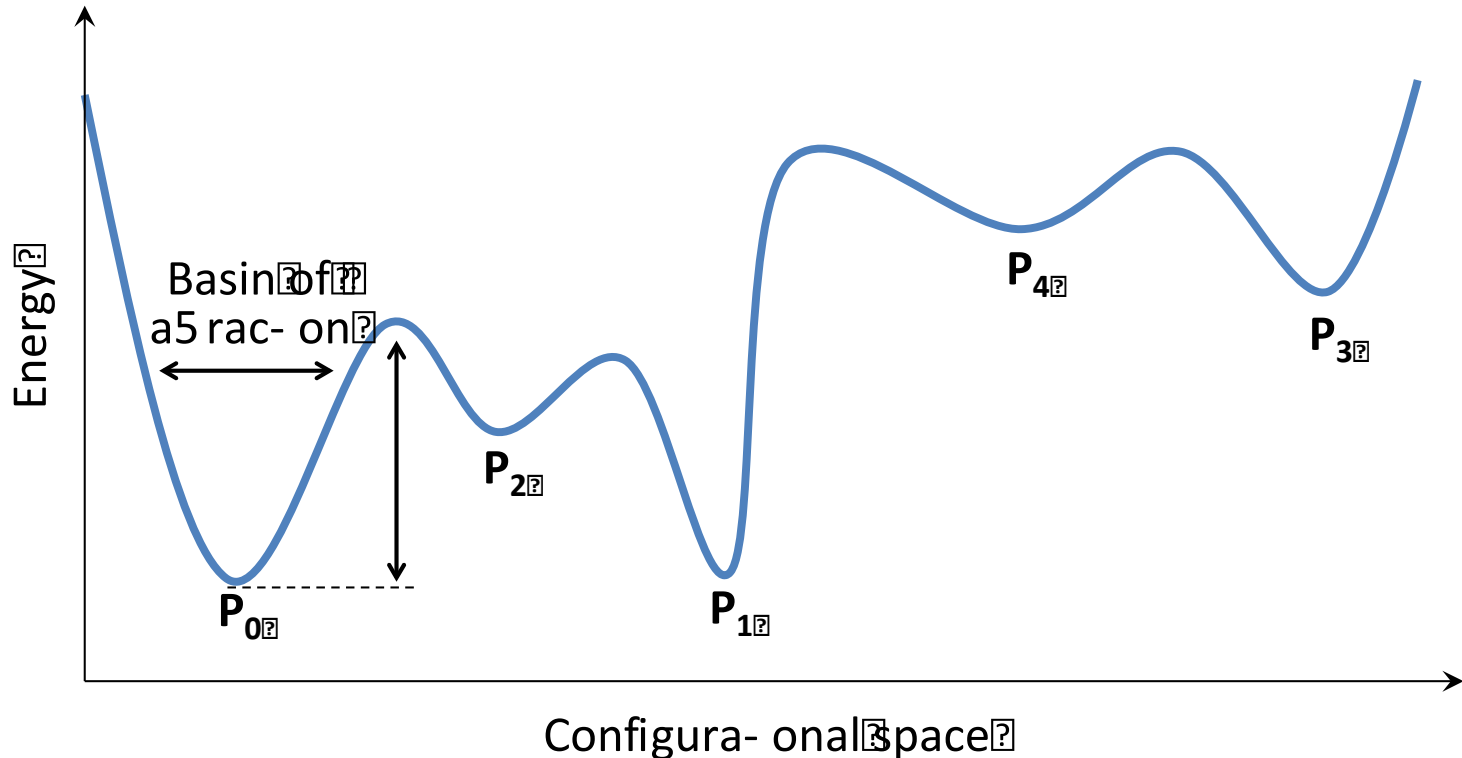
Polymorph Sampler



- Volume of configuration space (“width”) of local minima shown to correlate with realization of different polymorphs
- Random structure sampling followed by local DFT relaxations used to estimate the “width”
- All experimentally realized polymorphs appear as high frequency structures in random sampling
- Translates into a simple and elegant approach for predicting polymorphism - Polymorph sampler
- Easy to apply in a high-throughput fashion

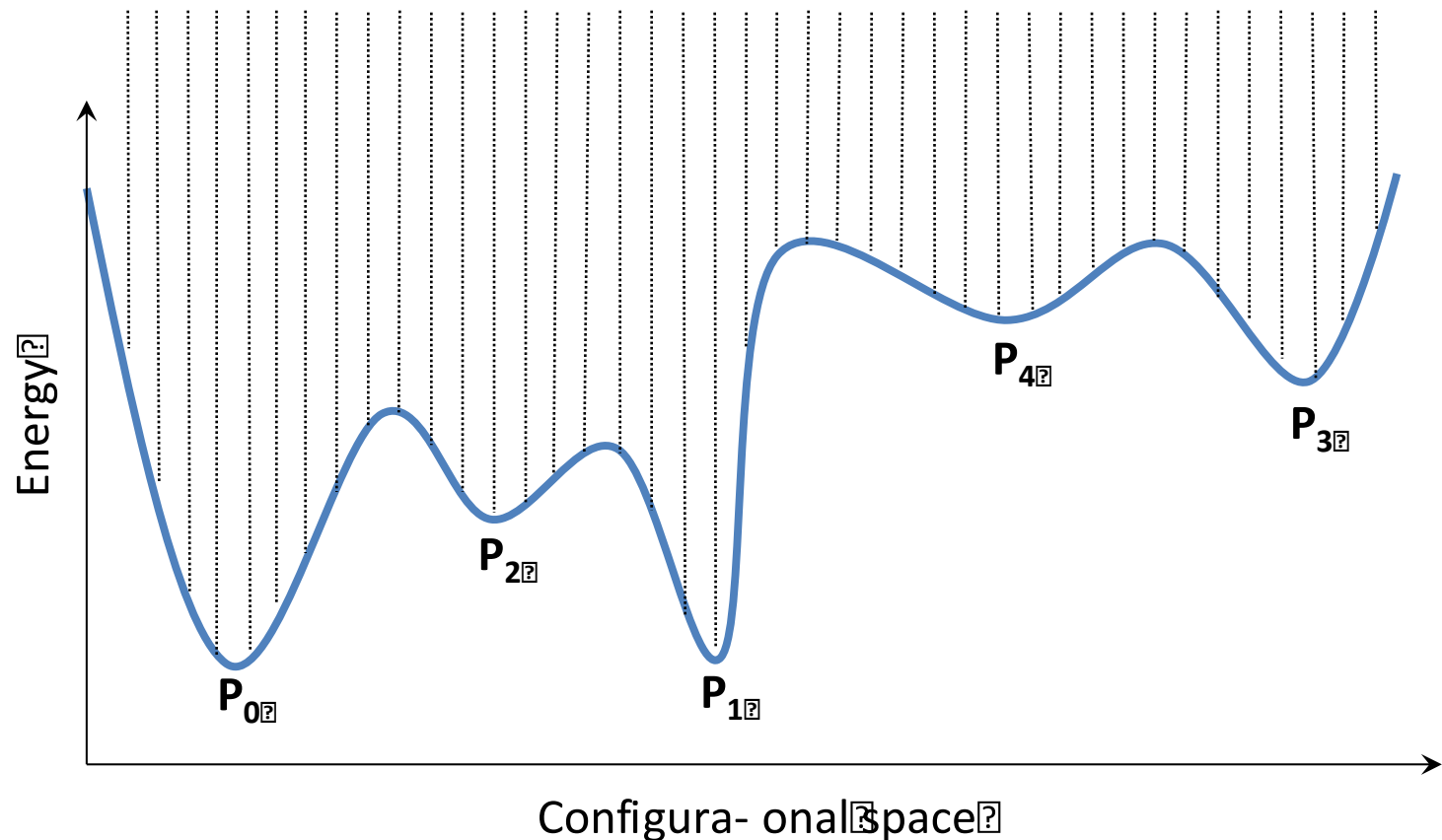
V. Stevanovic, *Phys. Rev. Lett.* 116, 075503 (2016)

Realizability from the “width” of local minima



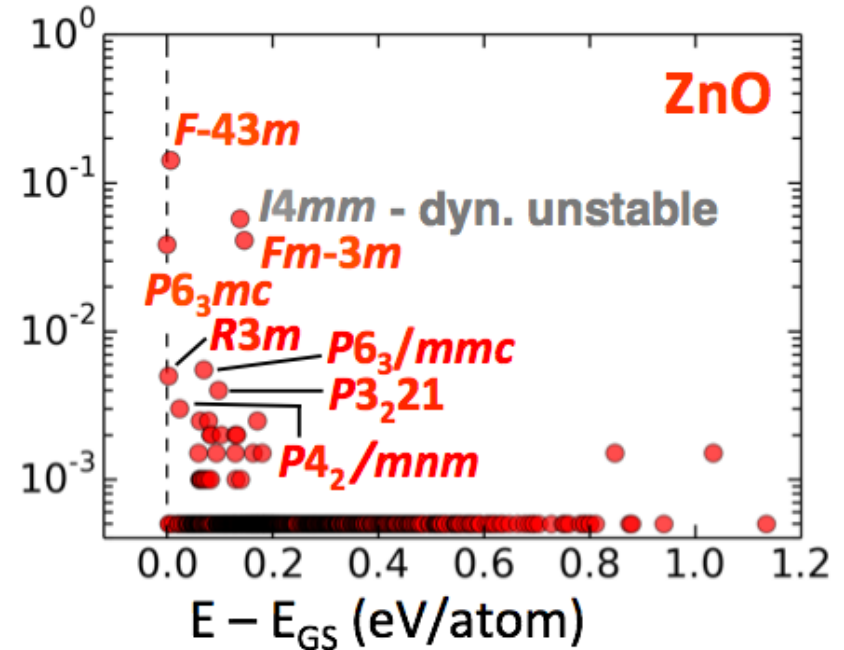
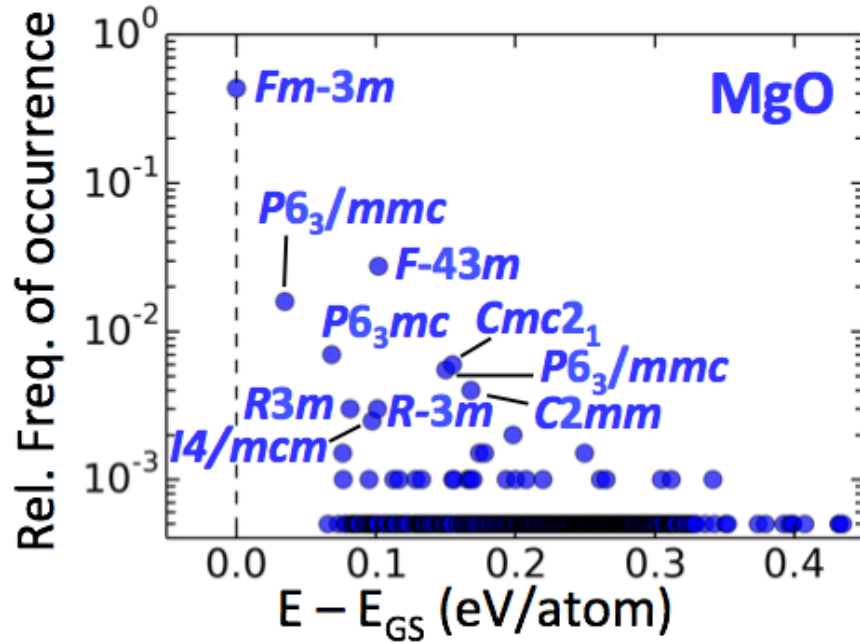
- For majority of systems, even well studied, only a relatively small number of polymorphs is known (experimentally realized)
- Theoretical predictions usually suggest a large number of low energy structures
- Our approach: “Width” of local minima matters, i.e. it is more probable to create exp. conditions that will prepare the system nearby larger (“wider”) minima

Random sampling to measure the “width”



- Random structure sampling followed by DFT relaxations can be used to measure/estimate the “width” of local minima
- Frequencies of occurrence in random sampling to assess the “width” of individual basins

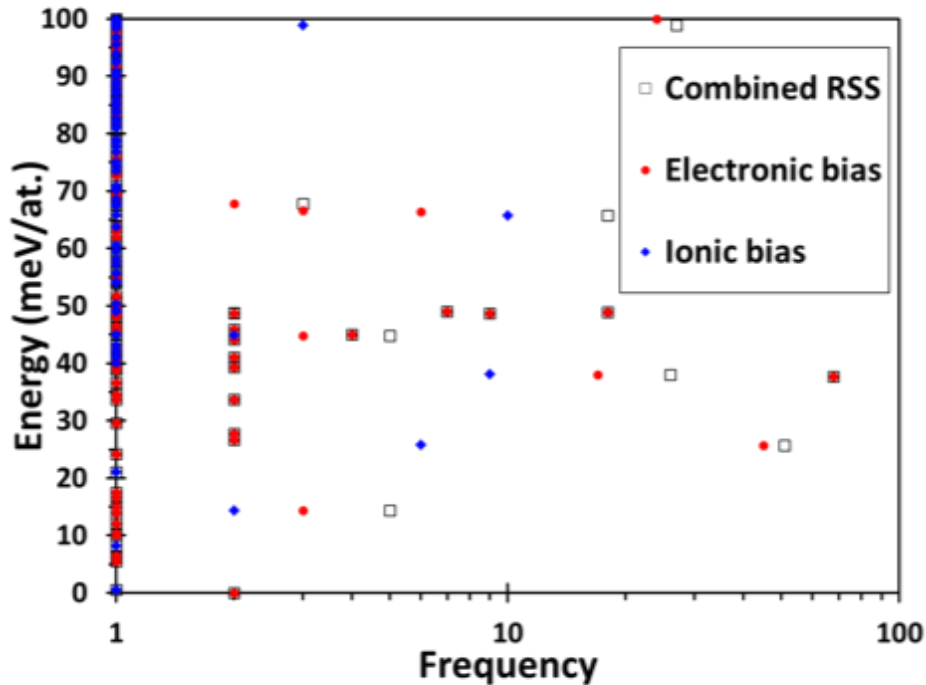
Validation of the hypothesis - MgO and ZnO



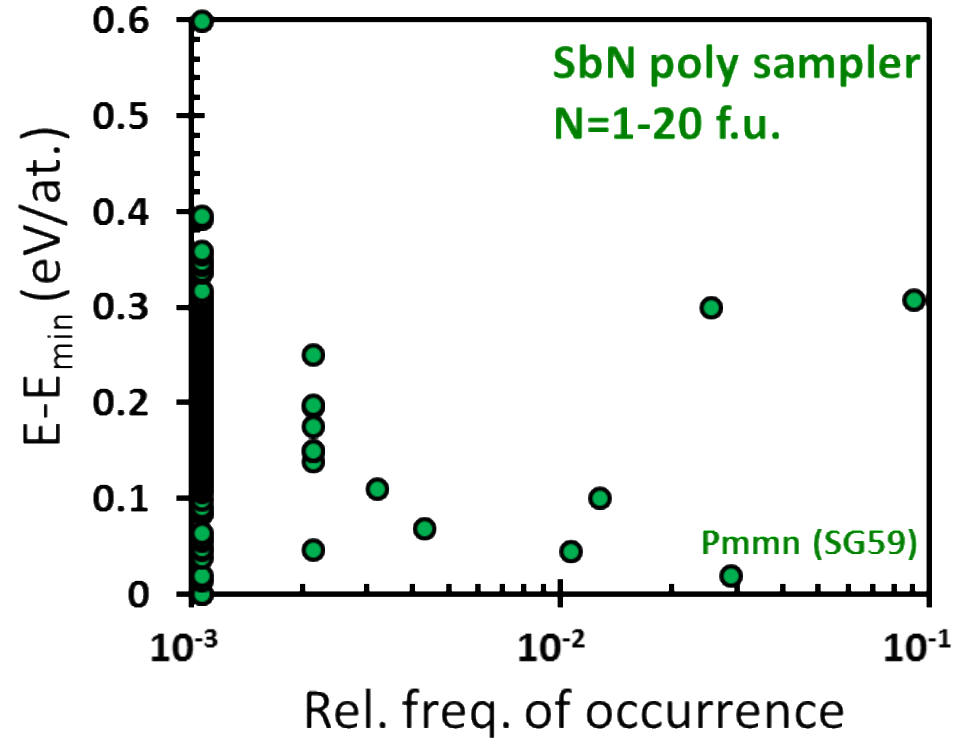
- 2,000 random structures per system
- All experimentally realized polymorphs appear as high freq. structures in random sampling
- RS MgO about 25 times more frequent than any other structure, indicates why is RS the only exp. realized MgO structure
- Experimentally realized ZnO structures are three top occurring in random sampling

CNGMD - Poly sampler applied in nitrides search

SnN



Sb(III)'Nitride'(SbN)'structures'



Paper submitted
(NREL/CSM/SLAC/LBNL/MIT)

Nitride search

Brief Summary: Materials by Design

Materials by Design has advanced considerably

- Increasing number of active centers
- Structural and functional materials
- Tools being developed, databases being developed

Integration (and iteration) of theory and experiment is critical

- Tool development
- Validation, accuracy

Materials Properties for have been predicted and confirmed

New materials have been predicted and synthesized

oxides, charge transport layers, absorbers, perovskite-analogs

Computational databases are becoming readily available, applied research can be built on top of these more basic science efforts

Theory and experiment can provide information on metastable systems, e.g. new nitrides, alloys, polymorphs

Predicting synthesis constitutes a grand challenge for materials science

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