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Analytic Description of Zinc-Blende Nanocrystals as Function of Size, Shape and Surface Orientation to Interpret Solid State Spectroscopy Data

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### Outline



- 1. Types of zb-NCs investigated, surface faceting vs. energy
- 2. Key Variables Significance and Usage
- 3. Graphical Description of zb-NCs: Symmetry Arguments
- 4. Hands-on Examples:
  - 4.1 Derivation of Number Series for 111-Faceted Octahedral zb-NCs
  - 4.2 A Glimpse on 111(dominant) -001 Faceted Quatrodecahedral zb-NCs

#### 5. Applications

5.1 The Ratio N<sub>IF</sub>/N<sub>bnd</sub>: Stress balance Between zb-NC and Embedding Matrix 5.2 The Ratio N<sub>IF</sub>/N<sub>NC</sub>: Interface Dipoles, Interface Charge Transfer (ICT) 5.3 The Ratio N<sub>IF</sub><sup>001</sup>/N<sub>IF</sub><sup>111</sup>: Detecting Quatrodecahedral NC Shape/Size by EPR 5.4 The Ratio N<sub>bnd</sub>/N<sub>NC</sub>: Response of zb-NCs to External Stress

6. Conclusions

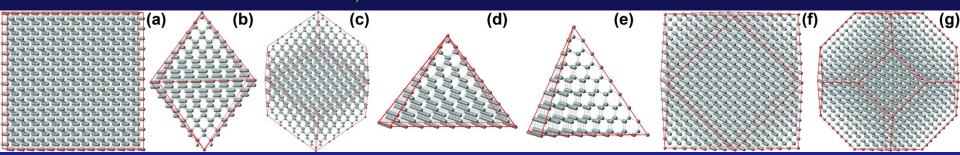
For details and respective number series, see

AIP Advances 6, 085306 (2016) [DOI: 10.1063/1.4960994] {open access}

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We focus on Si-based NCs; results are valid for zb- and diamond lattice NCs



zb-NCs described: (a) {001}-faceted cube, (b) {111}-faceted octahedra,
(c) {110}-faceted dodecahedra, (d) {111} pyramids with {001} base,
(e) {111}-faceted tetrahedra, (f) {001}- [dominant]/{111}-faceted quatrodecahedra, (g) {111}- [dominant]/{001}-faceted quatrodecahedra

TABLE I. Bond densities and free energies per square for low index Si facets. Bond density values taken from<sup>25</sup>, experimental surface energy values taken from<sup>26</sup>.

facet	surface bond	surface free		ve lowest surface energy,		
orientation	density $[\rm cm^{-2}]$	energy $[Jm^{-2}]$		/e 2 <sup>nd</sup> lowest surface energy		
{001}	$1.36 \times 10^{15}$	1.36	up to {433} face	ets <sup>[25,26]</sup> → {111}, {001} and		
{110}	$0.96 \times 10^{15}$	1.43		st relevant for NC surface		
{111}	$0.30 \times 10^{-10}$ $0.78 \times 10^{15}$	1.13	orientations	$\sqrt{6}$		
				$d_{\rm NC}[i] = \sqrt[3]{\frac{6}{\pi}} N_{\rm NC}[i] \times V_{\rm atom}$		
[25] J. Electrochem. Soc. 140, • NC size from atomic volume, $a_{NC}[i] = \sqrt{\pi} N_{NC}[i] \times V_{atom}$						
[26] PRL <b>70</b> , 1643 (1993) assuming spheres to compare different NC shapes						
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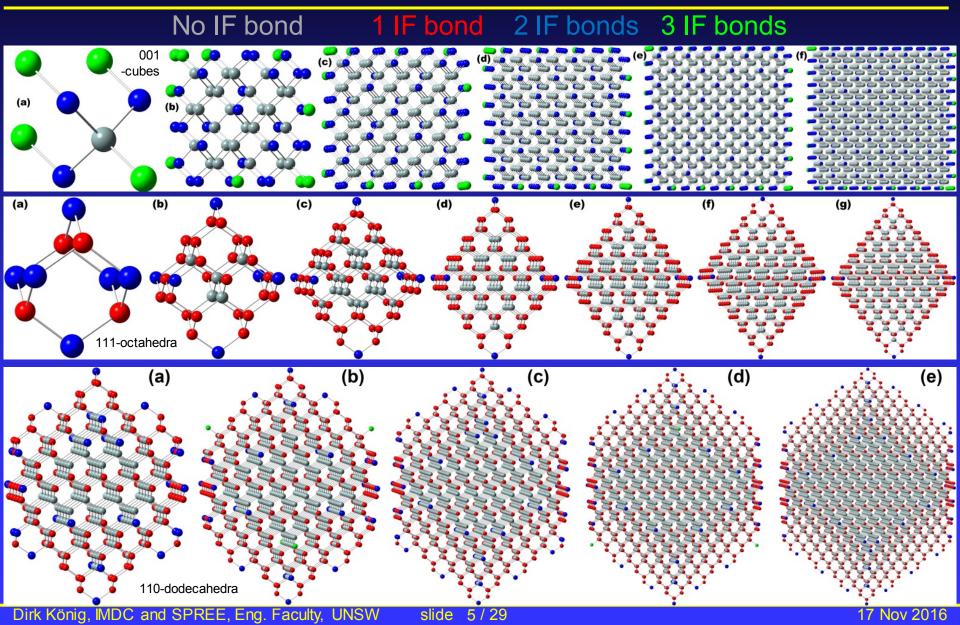
### 2. Key Variables Derived for zb-NCs: Significance and Usage



 Three numer series calculated over run index i: # N<sub>NC</sub>[i] – number of atoms forming zb-NC # N<sub>bnd</sub>[i] – number of bonds 111-Tetrahedron 001-111-Pyramid 111-001between zb-NC atoms Atom colours in zb-NCs refer to Quatrodecahedron 110-Dodecahedron # N<sub>IF</sub>[i] – number of bonds no one two three interface bonds per atom 2.0 from zb-NC to its environ-Atom 2.000 ment = interface bonds 1.995 Nanocrystal 1.990 1.985 further very useful results: 1.980 anocrystal shape 1 975 # N<sub>bpd</sub>[i] /N<sub>NC</sub>[i] – NC res-(facetting) 140 240 per 40 cubic (001) octahedral (111) ponse to external stress 1.5 dodecahedral (110) tetrahedral (111) like impurity doping 001-111pyramidal (001 base, 111) quatrodecahedral (001 [dominant], 111) 111-Octahedron Quatrodecahedron # N<sub>IF</sub>[i] /N<sub>hnd</sub>[i] – internal vs quatrodecahedral (111 [dominant], 001) external stress balance, Nanocrysta 10 diameter [nm] 100 e.g. NC shape as  $f(d_{NC})$  $N_{
m NC}^{
m cube}[i]=8\,i^3$ Atoms Forming Zinc-Blende Nanocrystal (zb-NC) # N<sub>IF</sub>[i] /N<sub>NC</sub>[i] – interface  $N_{ ext{bnd}}^{ ext{cube}}[i] = i ig[(4i-2)(4i-1)+1ig]$ Bonds between zb-NC Atoms charge transfer, impact on  $N_{
m IE}^{
m cube}[i] = 6\left[(2i-1)^2 + 3i - 1
ight]$ Interface bonds of zb-NC electronic structure of NC 001-Cube For any NC spectroscopy Raman, EPR, XRD, Al Dirk König, IMDC and SPREE, Eng. Faculty, UNSW 4 / 29 17 Nov 2016 slide



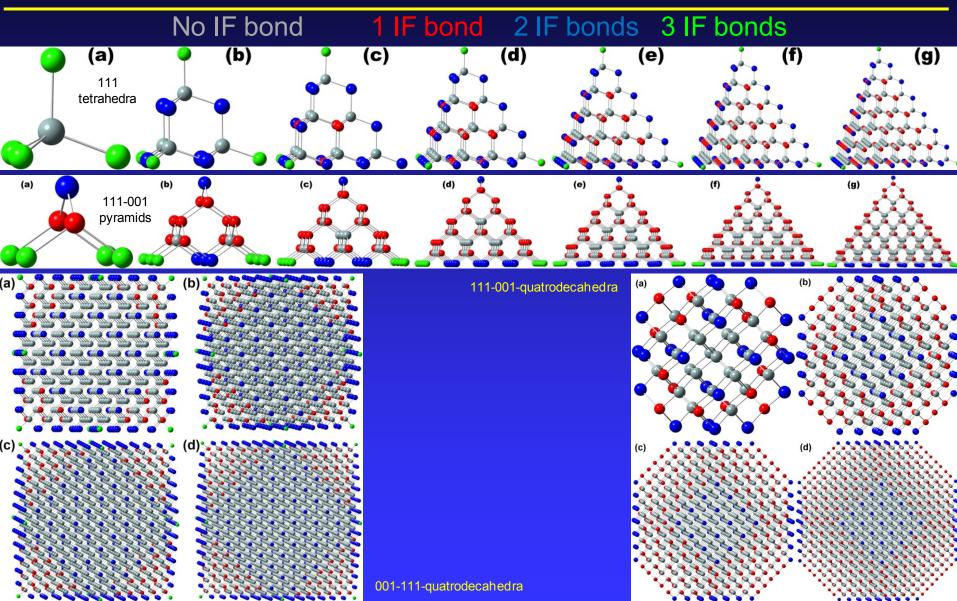
### 3. Graphical Description of zb-NCs: Symmetry Arguments





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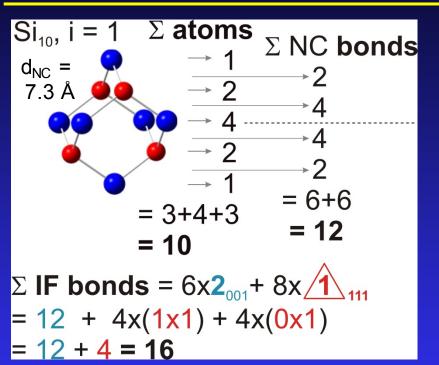


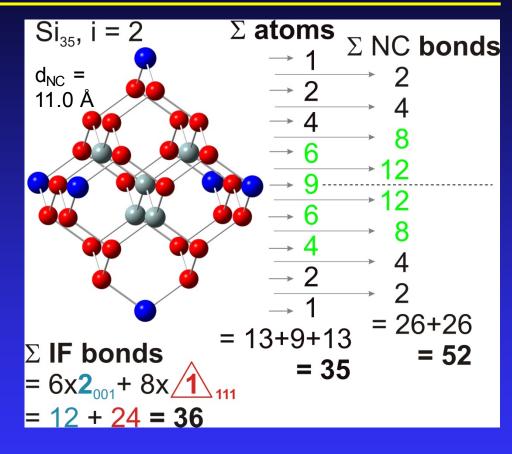
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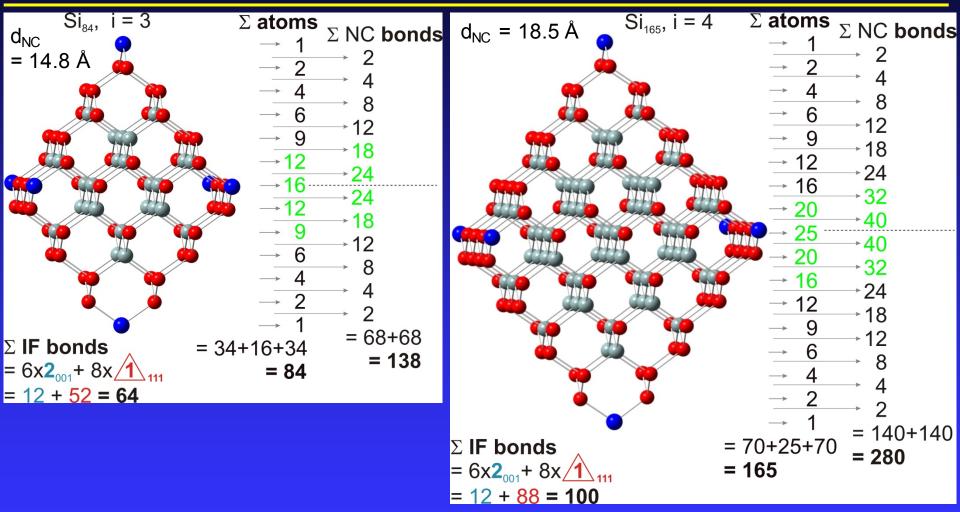
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4. Simple Example: Derivation of Number Series for {111} Octahedral zb-NCs



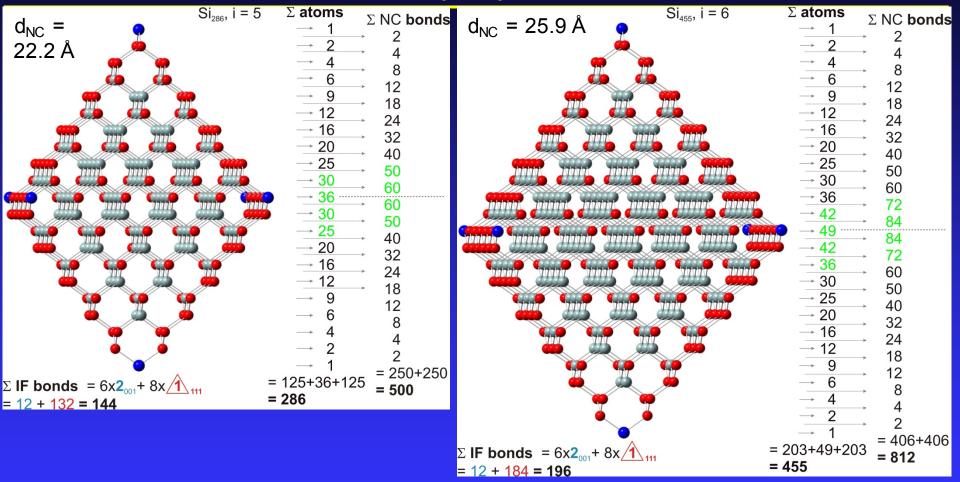


### 4. Simple Example: Derivation of Number Series for {111} Octahedral zb-NCs



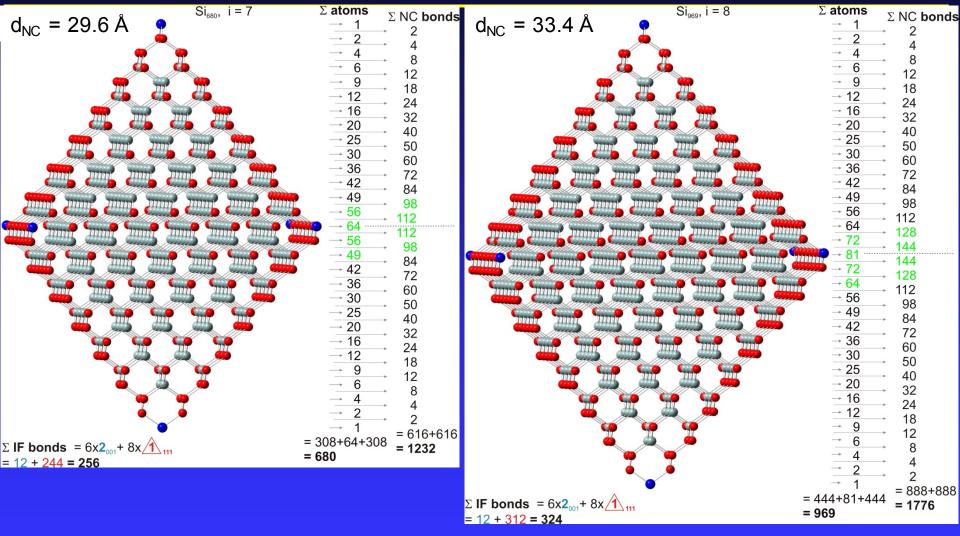
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## 4. Simple Example: Derivation of Number Series for {111} Octahedral zb-NCs





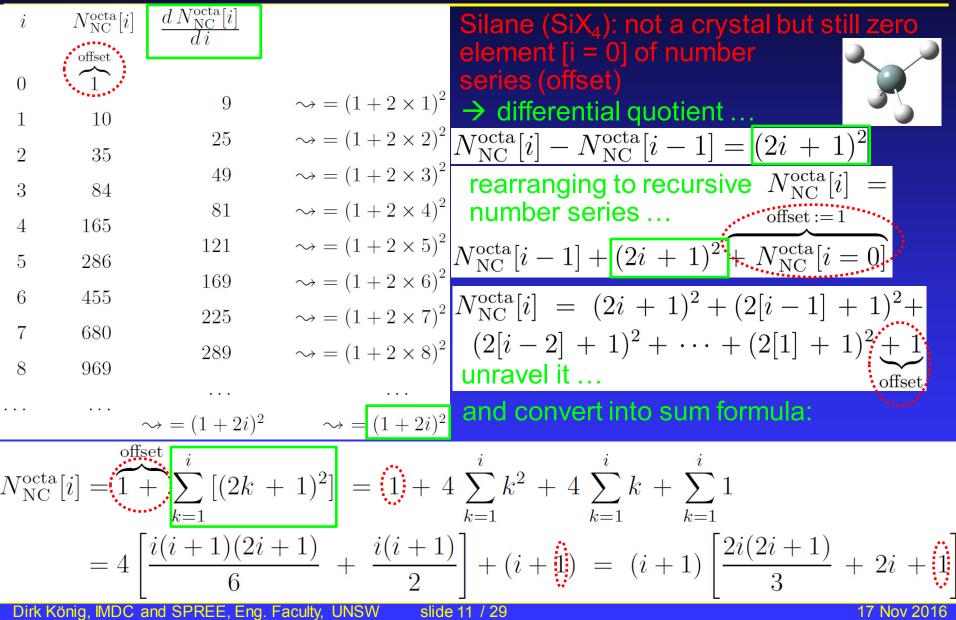
### 4. Simple Example: Derivation of Number Series for {111} Octahedral zb-NCs







### 4. Simple Example: Deriving the Number of NC Atoms for {111} Octahedral zb-NCs



# 4. Simple Example: Deriving the Number of Bonds Between NC Atoms for {111} Octahedral zb-NCs

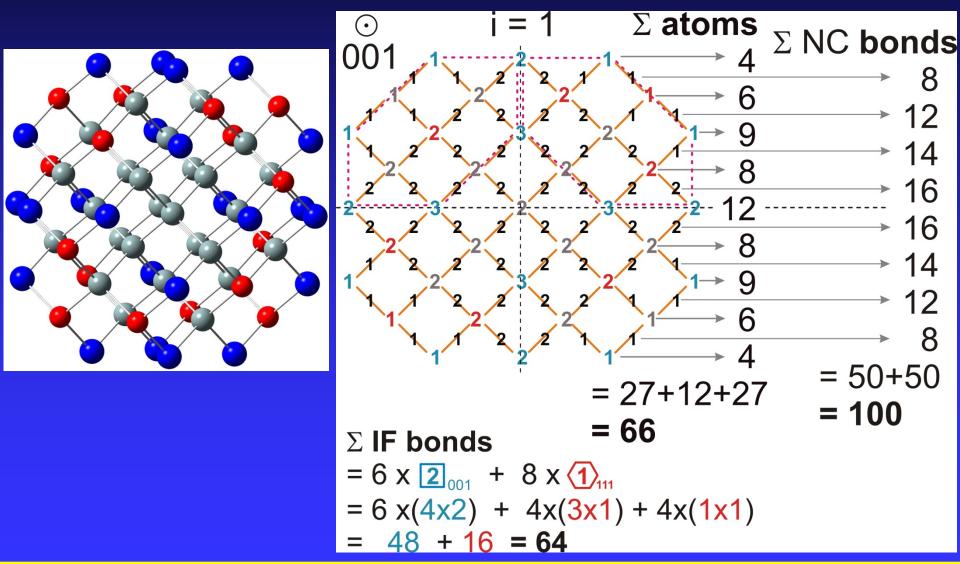




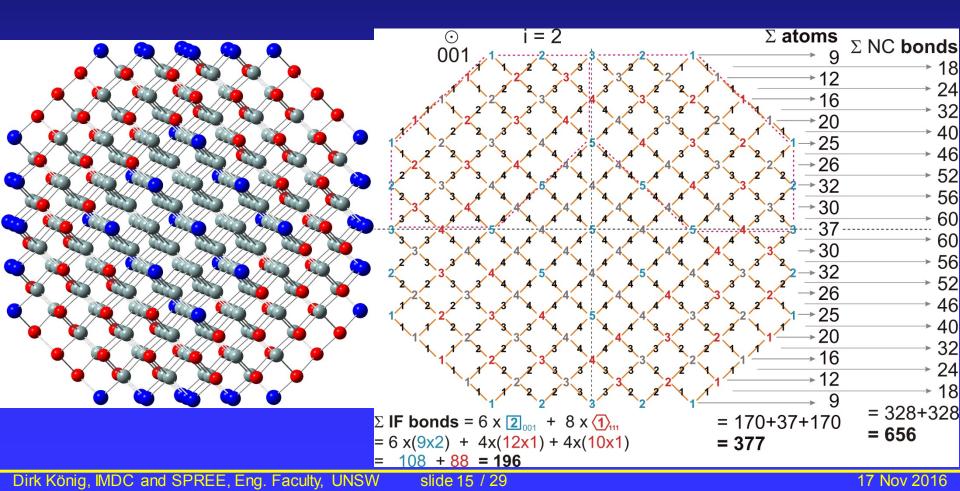
	Interfac	ce Bonds for {	111} Octahedral zb-NCs
i	$N_{\mathrm{IF}}^{\mathrm{octa}}[i]$		Silane (SiX <sub>4</sub> ): not a crystal but still zero
0	4	$\rightsquigarrow = 4 \ (0+1)^2$	element [i = 0] of the number series, having four IF bonds
1	16	$\rightsquigarrow = 4 \ (1+1)^2$	
2	36	$\rightsquigarrow = 4 \ (2+1)^2$	
3	64	$\rightsquigarrow = 4 \ (3+1)^2$	
4	100	$\rightsquigarrow = 4 \ (4+1)^2$	
5	144	$\rightsquigarrow = 4 \ (5+1)^2$	
6	196	$\rightsquigarrow = 4 \ (6+1)^2$	
7	256	$\rightsquigarrow = 4 \ (7+1)^2$	
8	324	$\rightsquigarrow = 4 \ (8+1)^2$	
			$d_{\rm NC}[i] = \sqrt[3]{\frac{6}{\pi}} N_{\rm NC}[i] \times V_{\rm atom}$
$\sim \rightarrow$	$=4(i+1)^2$	$\rightsquigarrow = 4(i+1)^2$	$u_{\rm NC}[\iota] = \sqrt{\pi^{1} v_{\rm NC}[\iota]} \wedge v_{\rm atom}$

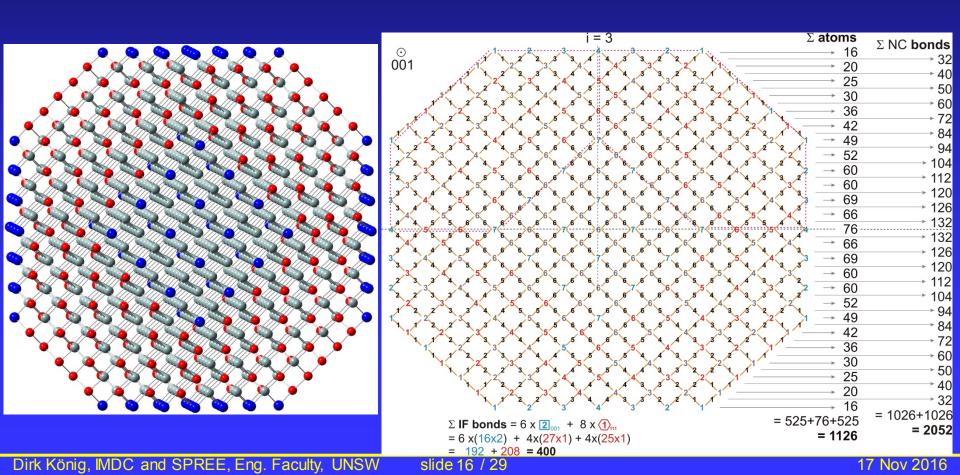
4. Simple Example: Deriving the Number of

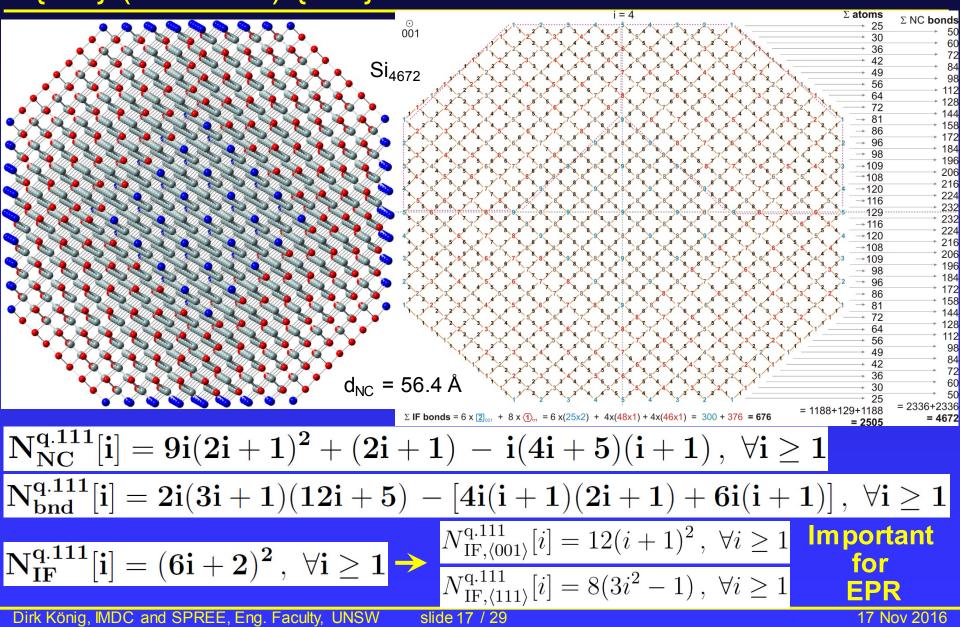
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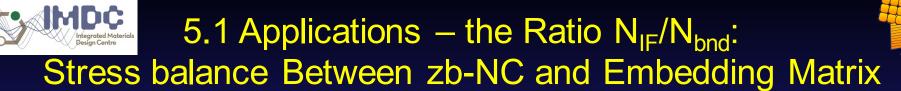


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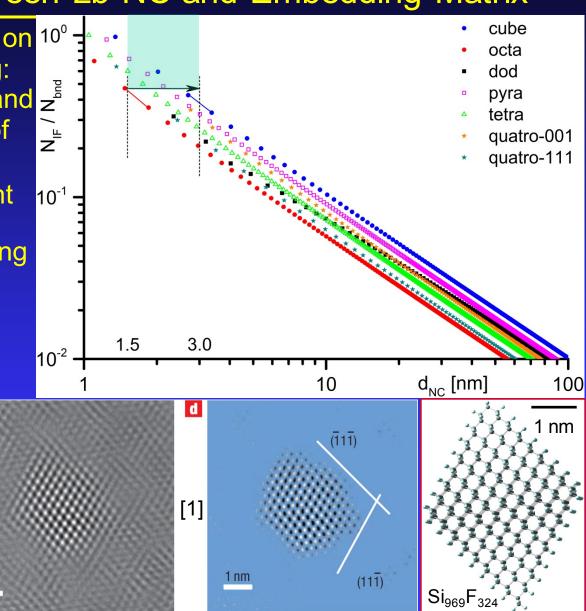






- significant influence of stress on balance zb-NC vs. embedding: matrix dominates NC growth and hence NC shape as function of NC size
   10°
- octahedral Si NCs is dominant 1 shape for  $d_{NC} \le ca. 3 \text{ nm}^{[1]}$
- porous Si etching & self-limiting oxidation provided smallest Si NCs = 1.5 ± 0.2 nm<sup>[2]</sup>
- → no Si NC formation for  $N_{IF}/N_{bnd} \ge 0.41 \pm 0.06;$
- value specific to NC- and matrix-material, e.g. smallest Si NC size should be bigger in Si<sub>3</sub>N<sub>4</sub> due to its higher Young's modulus
- [1] Nature Nanotech 3, 174 (2008)[2] PRL 72, 2648 (1994)

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1 nm

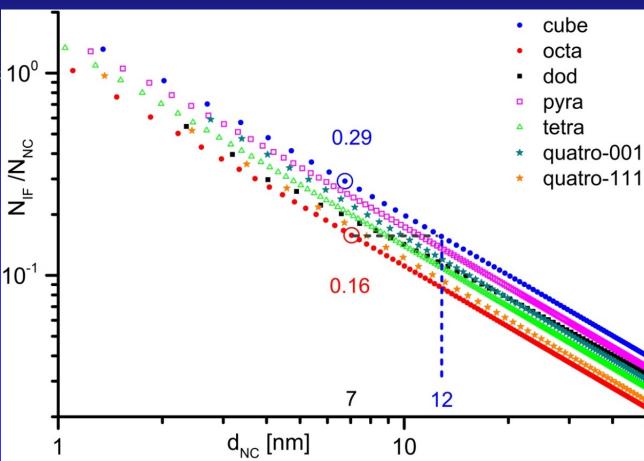
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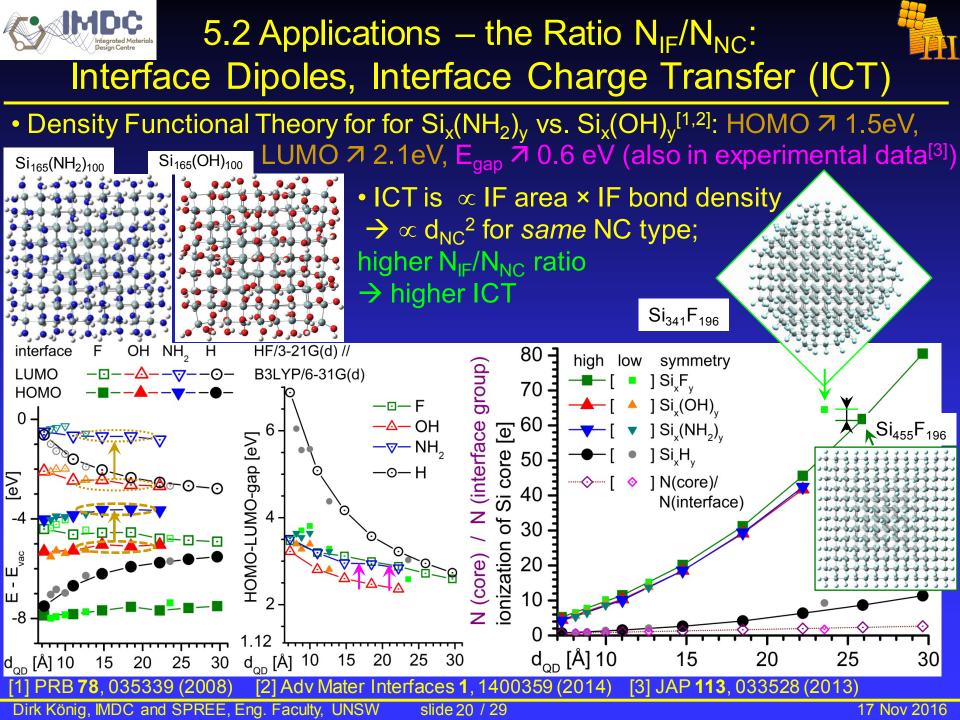


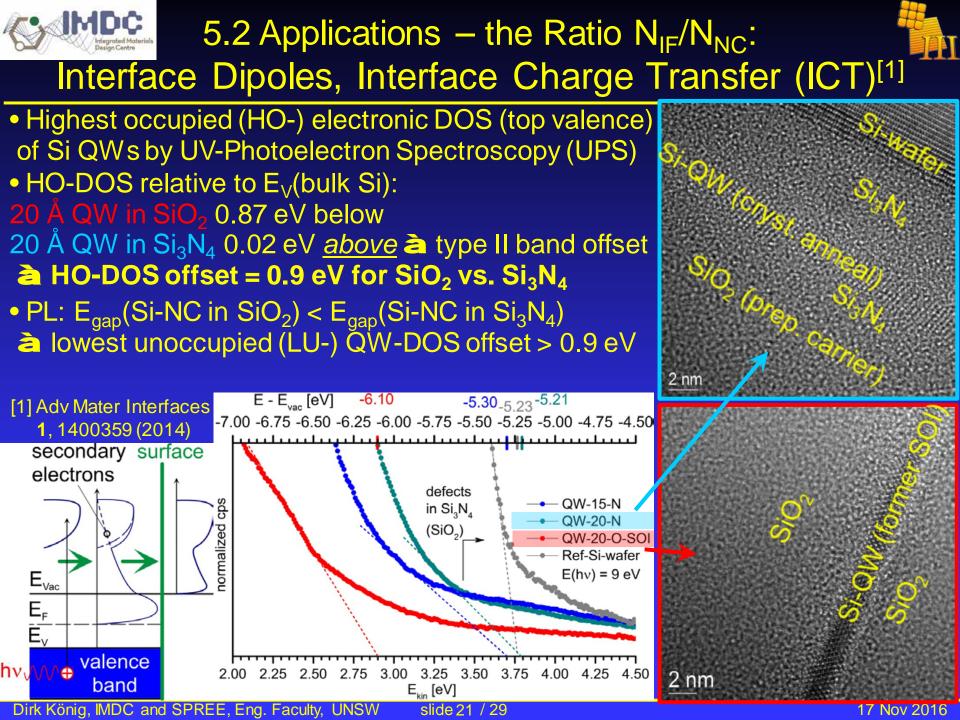
- N<sub>IF</sub>/N<sub>NC</sub> depends on # surface density of IF bonds, hence interface faceting – {001}, {110}, {111} # ratio of surface atoms to all atoms which form zb-NC
- $N_{IF}/N_{NC}$  much higher for elongated structures; fins (fin-FETs) vs. {001} cube

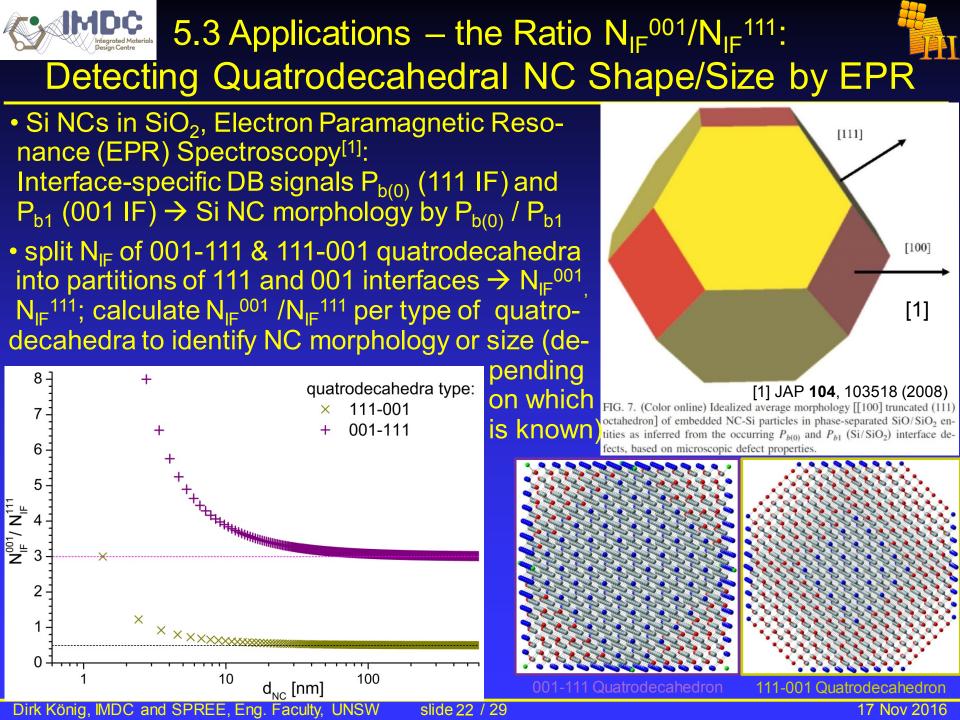
• N<sub>IF</sub>/N<sub>NC</sub> decisive for NC electronic structure modification by dielectric 10° • graph shows  $N_{IF}/N_{NC}$ , ICT dominates electronic structure of Si NCs for  $d_{NC} \ge 7 \text{ nm}^{[1]}$ ; going to a {001}-cube extends the NC size to 12 nm, further increase in structure length can be obtained by cubicle (fin-FET, 14 nm technology)

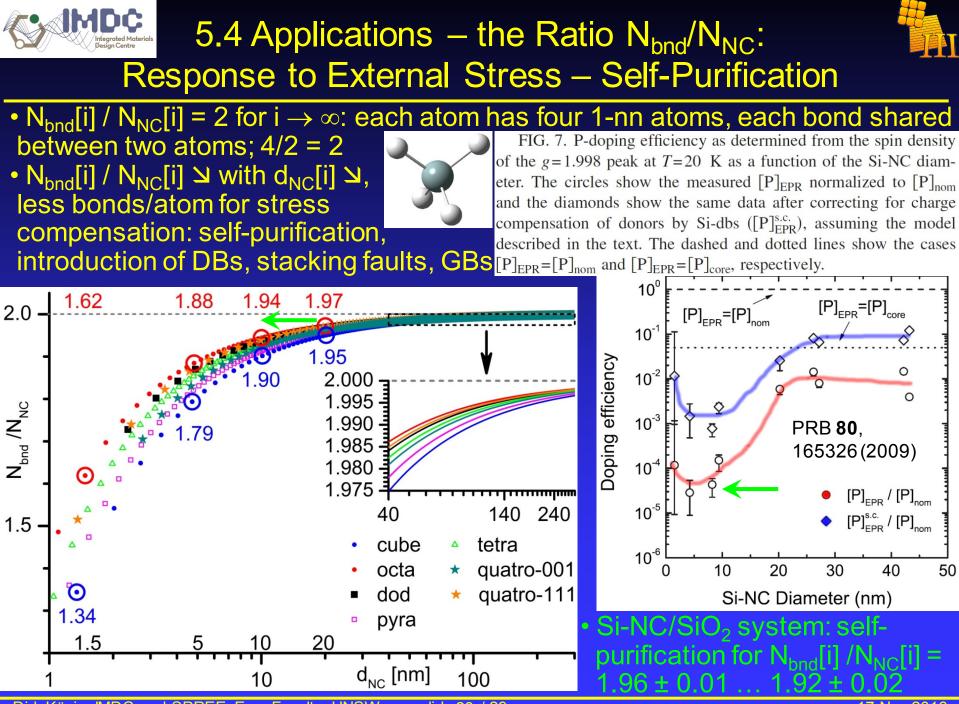
[1] Appl. Mater. Interfaces 1, 1400359 (2014); DOI: 10.1002/admi.201400359





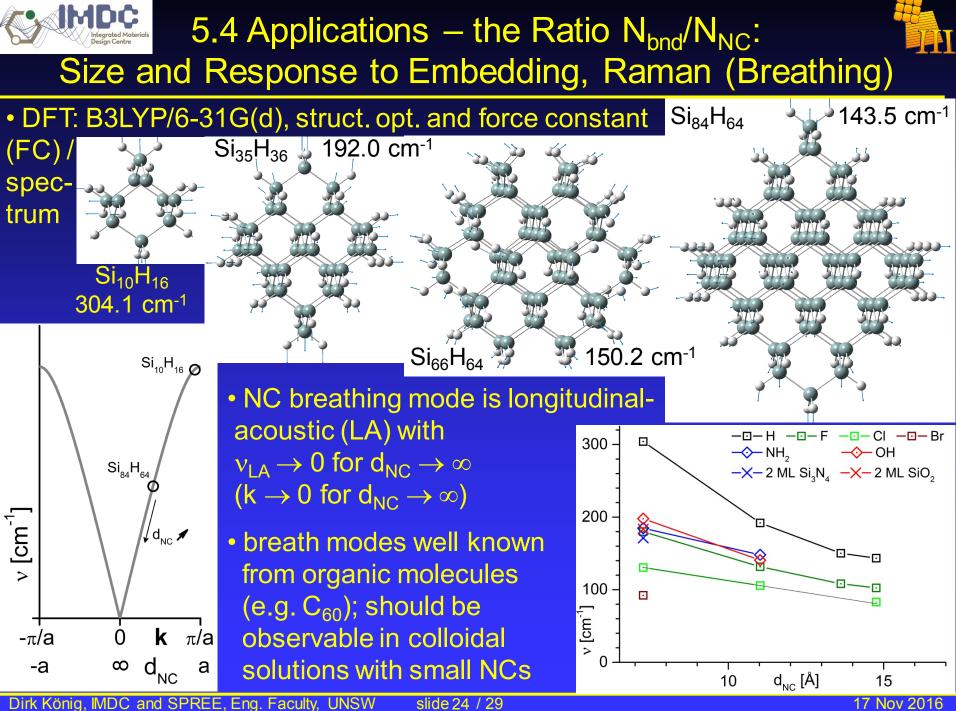


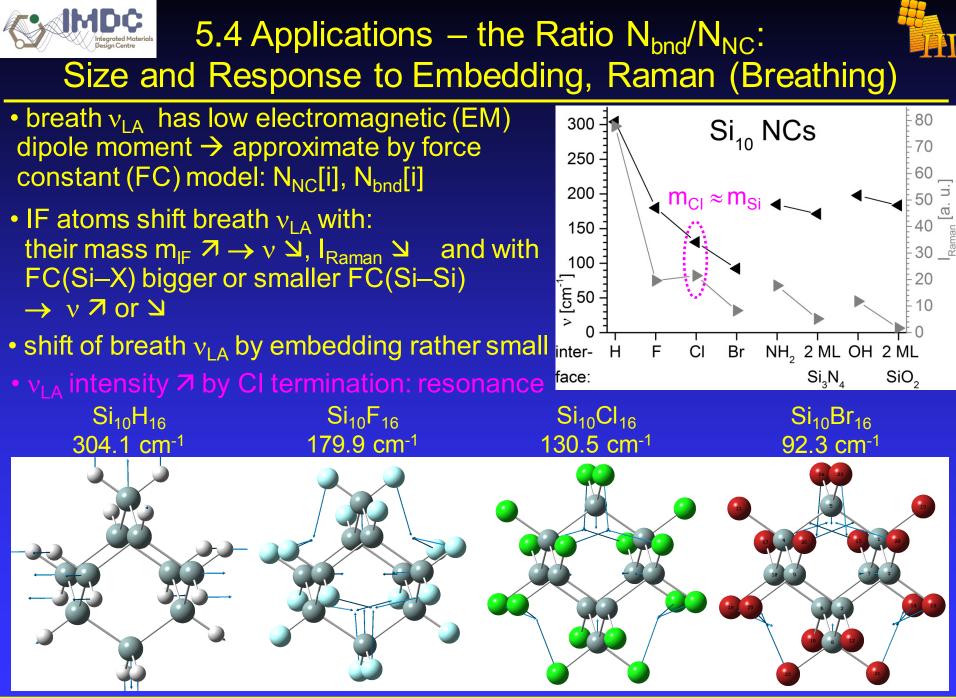




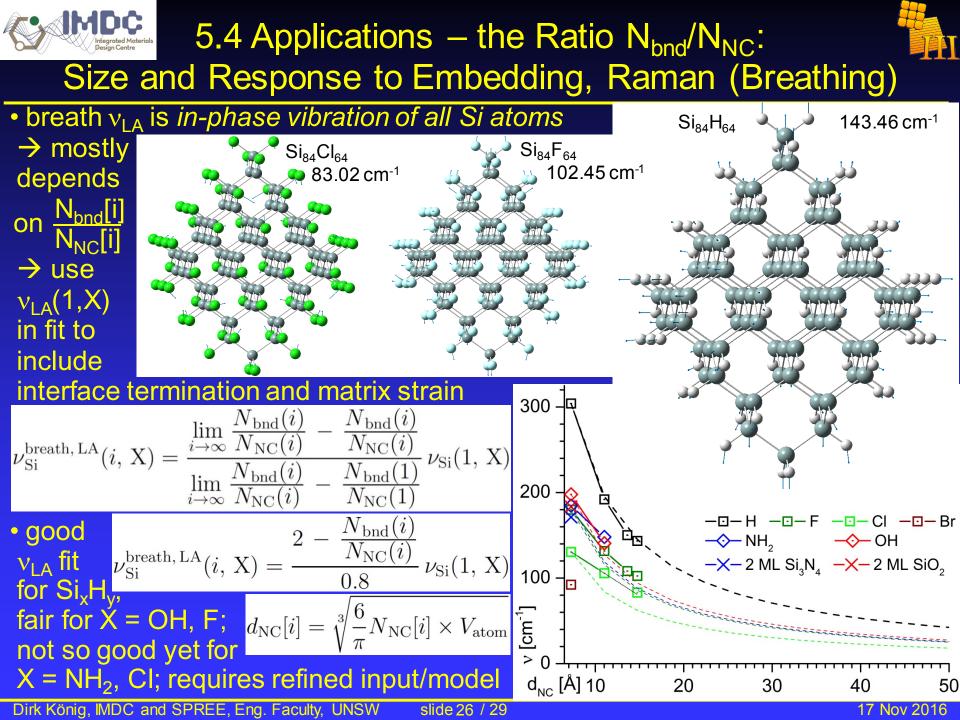
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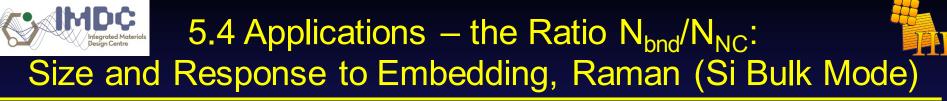
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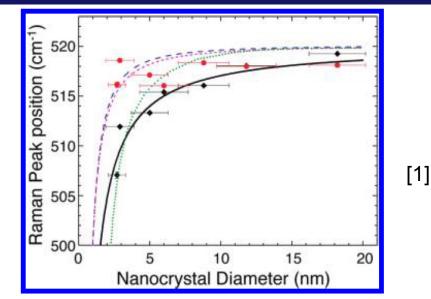


- no analytical description of fundamental LO/TO Si  $v^{LO/TO}$  mode (521 cm<sup>-1</sup>) as f(d<sub>NC</sub>) which fit experimental data, just least square fits<sup>[1]</sup>
- $\rightarrow$  no causality, hence no interpretation and assignment of results
- analytical fits with NC-specific parameters ( $N_{NC}$ ,  $N_{bnd}$ ,  $N_X$ ) allow for real data interpretation

More theoretical work is needed to provide a quantitatively accurate model of the size dependence of the Raman spectra of Si nanocrystals. However, a convenient scaling relationship between the Raman peak position and nanocrystal size might be used<sup>28,29,31</sup>

$$\omega(D) = \omega_0 - A \left(\frac{a}{D}\right)^{\gamma} \tag{2}$$

 $\omega(D)$  is the diameter-dependent peak Raman frequency,  $\omega_0$  is the bulk Raman peak position of bulk crystalline Si (521 cm<sup>-1</sup> for crystalline Si), *a* is the lattice constant (0.543 nm for Si), and *D* is the particle diameter. *A* and  $\gamma$  are fitting parameters

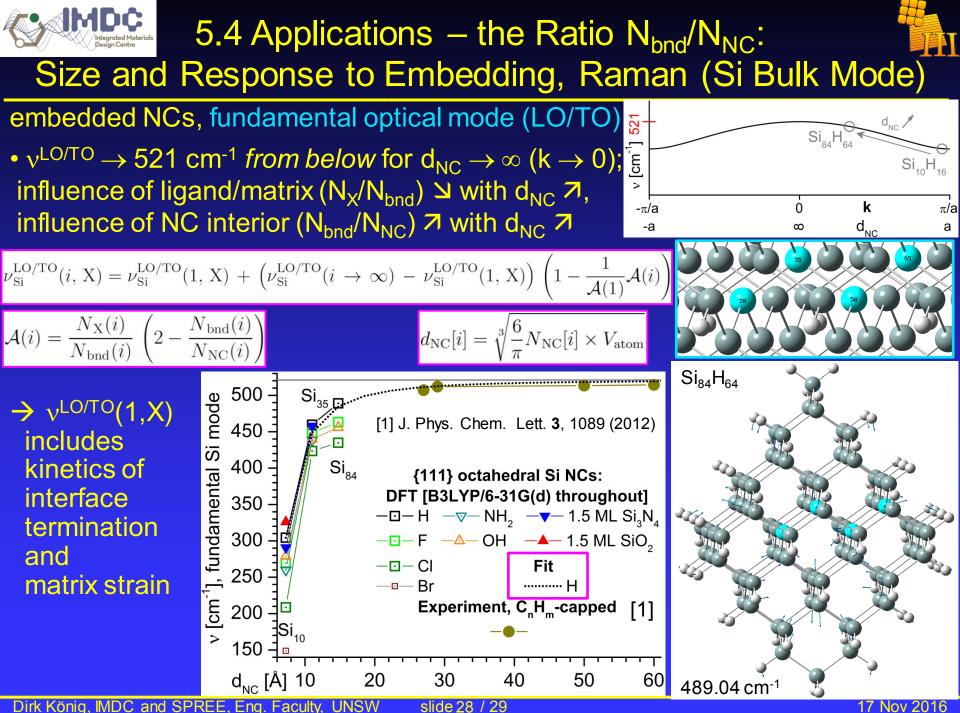


**Figure 3.** Raman peak position as a function of the nanocrystal diameter for freestanding ligand-passivated (black diamonds) and oxide-embedded (red circles) nanocrystals. The black line is a least-squares analysis fit of eq 2 to the Raman peak positions of the ligand-passivated freestanding nanocrystals; the blue and magenta curves are the predicted Raman peak positions versus size from the RWL and BP models, respectively. The green dotted curve shows recent model predictions by Faraci et al.<sup>17</sup> The error bars represent the particle size polydispersity determined from SAXS.

[1] J. Phys. Chem. Lett. **3**, 1089 (2012) 17 Nov 2016

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### 6. Conclusions



- seven high symmetry zb-NCs with minimum surface energy (max. occurrence): 001 cubes, 111 octahedra, 001 dodecahedra, 111 tetrahedra, 111 pyramids with 001 base, 111-001 quatrodecahedra, 001-111 quatrodecahedra
- Analytical description of NC atoms (N<sub>NC</sub>[i]), number of bonds between these (N<sub>bnd</sub>[i]) and NC interface bonds (N<sub>IF</sub>[i]) by explicit number series as f(d<sub>NC</sub>[i]), surface faceting and shape; useable for *any* solid state spectroscopy technique
- $N_{IF}[i] / N_{bnd}[i]$  as f(d<sub>NC</sub>[i]): stress balance zb-NC  $\leftrightarrow$  embedding matrix; crystallization limit and polymorphism (segregation anneal, gas phase formation, ...)
- ratio N<sub>IF</sub>[i] / N<sub>NC</sub>[i] as f(d<sub>NC</sub>[i]): interface dipoles, interface charge transfer which dominates (influences) electronic properties of Si (III-V) NCs
- ratio N<sub>IF</sub><sup>001</sup>[i] / N<sub>IF</sub><sup>111</sup>[i] as f(d<sub>NC</sub>[i]): key tool to detect shape of quatrodecahedral NCs (and other zb-NC shapes, see above) using Electron Paramagnetic Resonance (EPR) Spectroscopy
- N<sub>bnd</sub>[i] / N<sub>NC</sub>[i] as f(d<sub>NC</sub>[i]): gauge for stress response of zb-NC by DBs, stacking faults, self-purification; working on analytic Raman-fits (maintaining causality)

 For details and all number series → DOI: 10.1063/1.4960994 {open access}

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